





mitosis (Jalink et al., 1993b). Additional factors, such as insulin-like growth factors, were required for the progression of the cell cycle. Once the cells have undergone morphological differentiation, the addition of LPA reverses this morphological change. Thus, LPA-induced neurite retractions result from the contraction of the actin-cytoskeleton, rather than from loss of adhesion to the substratum (Jalink et al., 1993b; Jalink et al., 1994b).

LPA, similar to other physiological chemoattractants (e.g., interleukin-8), induces cell migration by a haptotactic mechanism in human monocytes (Zhou et al., 1995). In addition to inducing cell migration, LPA promotes the invasion of hepatoma and carcinoma cells into the monolayer of mesothelial cells (Imamura et al., 1993). The mechanism that underlies this invasion is still unclear, but it may be due to enhanced cell motility and increased cell adhesion. Finally, LPA is also known to block neuronal cardiomyocyte apoptosis (Umnansky et al., 1997).

A unique natural phospholipid, namely cyclic-PA, was shown to be responsible for cellular actions that were similar to or opposite to other GPMs, depending on the cell type. When tested on the *Xenopus* oocyte, it elicited chloride current just like other GPMs, but its response was not desensitized by LPA (Fischer et al., 1990). Murakami-Murofushi et al. (1993) showed that cyclic-PA exhibited antiproliferative actions, unlike LPA, which induces proliferation.

PLGF receptors (PLGFRs) belong to a seven-transmembrane (7 TM) guanine nucleotide-binding regulatory protein (G protein)-coupled receptors (GPCR) superfamily. Seven-TM GPCRs are a family of cell-surface receptors that mediate their cellular responses via interacting with the heterotrimeric G-protein. A number of LPA receptors have been identified including, among others, EDG-2, EDG-4, EDG-7, and PSP-24. A phylogenetic tree illustrating the relatedness of these LPA receptors and others is shown in Figure 1.

In 1996, Hecht et al. used differential hybridization to clone a cDNA encoding a putative serpentine receptor from mouse neocortical cell lines (Hecht et al., 1996). The gene was termed as ventricular zone gene-1 (*Vzg-1*). The gene was expressed in cortical neurogenic regions and encoded a protein with a molecular weight of 41 kDa (364 amino acids). *Vzg-1* was very similar to an unpublished sheep sequence termed endothelial differentiation gene-2 (EDG-2). The same cDNA was also isolated as an orphan receptor from mouse and bovine libraries, and was known as *ret-3* (Macrae et al., 1996). It was widely distributed in the mouse tissue, with the highest expression in the brain and heart.

In 1996, Guo et al., using a PCR base protocol, isolated another putative LPA receptor PSP-24 (372 amino acids) from *Xenopus* oocyte (Guo et al.,

1996). This receptor showed little similarity with *Vzg-1*/EDG-2/*ret-3* (Guo et al., 1996). A sequence based search for sphingolipid receptors, using the cDNA sequence of the EDG-2 human LPA receptor, led to two closely related GPCRs, namely, rat H218 (EDG-3, 354 amino acids) and EDG-3 (378 amino acids) (An et al., 1997a). Northern analysis showed a high expression of mRNA that encoded EDG-3 and EDG-5 in heart tissue.

The recent identification of EDG-2 as a functional receptor for LPA prompted An et al. to perform a sequence-based search for a novel subtype of LPA receptor (An et al., 1998a). A human cDNA, encoding a GPCR, was discovered and designated EDG-4 (An et al., 1998a). Northern blot analysis showed that, although EDG-2 and EDG-4 both serve as GPM receptors, their tissue distributions were very different. Unlike EDG-2, EDG-4 was primarily expressed in peripheral blood leukocytes and testes (An et al., 1998a).

PCR amplification cDNA from human Jurkat T cells identified a previously unknown GPCR that belongs to the EDG family. The identified GPCR was designated EDG-7. It has a molecular mass of 40 kDa (353 amino acids). Northern blot analysis of EDG-7 expression in human tissues showed that it is expressed in heart, pancreas, prostate, and testes (Randall et al., 1999). Thus, there are two distinct families of PLGFs receptors PSP-24 and EDG; with a total of ten individual PLGFRs (Figure 1). The list continues to grow. These various receptors can be classified based on their ligand specificities for GPMs or SPVs, as shown in Table 1 below.

Table 1: Phospholipid Growth Factor Receptor, Length and Principle Ligand

PLGFR	Number of amino acids	Principle Ligand
EDG-1	381	SPP
EDG-2	364	LPA
EDG-3	378	SPP
EDG-4	382	LPA
EDG-5	354	SPP
EDG-6	385	SPP
EDG-7	353	LPA
EDG-8	400	SPP
<i>Xenopus</i> PSP24	372	LPA
Murine PSP24	373	LPA

*Xenopus* PSP24 and murine expressed PSP24 specifically transduce GPM (LPA, Fischer et al., 1998) evoked oscillatory chloride-currents. These are not structurally

homologous to the EDG family (Tigyi and Miledi, 1992; Fernhout et al., 1992). The EDG family can be divided into two distinct subgroups. The first group includes EDG-2, EDG-4, and EDG-7, which serve as receptors for only QPM (Hecht et al., 1996; An et al., 1998; Bandoh et al., 1999; An et al., 1998b) and transmit numerous signals in response to ligand binding. The second group involves EDG-1, EDG-3, EDG-5, EDG-6, and EDG-8, and is specific for SPMs (An et al., 1997a; Im et al., 2000; van Brocklyn et al., 1998; van Brocklyn et al., 2000; Spiegel and Milstein, 2000). Principle tissue expression of the various PLGFR's is shown in Table 2 below.

Table 2: Human Tissue Expression of Phospholipid Growth Factor Receptors

PLGFR	Human Tissue with Highest Expression
EDG-1	Ubiquitous
EDG-2	Cardiovascular, CNS, Gonadal tissue, GI
EDG-3	Cardiovascular, Leukocyte
EDG-4	Leukocyte, Testes
EDG-5	Cardiovascular, CNS, Gonadal tissue, Placenta
EDG-6	Lymphoid, Hematopoietic tissue
EDG-7	Heart, Pancreas, Prostate, Testes
EDG-8	Brain
PSF24	CNS

PLGFs activate multiple G-protein-mediated signal transduction events. These processes are mediated through the heterotrimeric G-protein families  $G_{\alpha}i$ ,  $G_{\alpha}o$  and  $G_{\alpha}13$  (Moolenaar, 1997; Spiegel and Milstein, 1995; Gohla, et al., 1998).

The  $G_{\alpha}i$  pathway is responsible for phospholipase C (PLC) activation, which in turn induces inositol triphosphate (IP<sub>3</sub>) production with subsequent mobilization of  $Ca^{2+}$  in a wide variety of cells (Tokumura, 1995). In some cells, this response is PTX-sensitive, implying that there is involvement of multiple PTX-sensitive and insensitive pathways (Tigyi et al., 1996). This pathway is also responsible for the diacyl glycerol (DAG)-mediated activation of protein kinase C (PKC). PKC activates cellular phospholipase D (PLD), which is responsible for the hydrolysis of phosphatidyl choline into free choline and PA (van der Bend et al., 1992a). Also, PLD is capable of activating MAP kinase directly, or via DAG activation of PKC in some cell types (Ghosh et al., 1997).

The mitogenic-signaling pathway is mediated through the G-protein heterotrimeric G<sub>o</sub> subunit. Transfection studies indicate that the G<sub>o</sub> dimer rather than the G<sub>i</sub> subunit is responsible for Ras-MAP kinase activation. The activation of

Ras is preceded by the transactivation of the receptor tyrosine kinases (RTKs) such as EGF (Cummins et al., 1998) or PDGF receptors (Herrlich et al., 1998). The transactivated RTKs activate Ras, which leads to the activation of MAP kinases (ERK 1,2) via Ref. The G<sub>13</sub> subunit, which is PTX-sensitive, inhibits adenylyl cyclase (AC), resulting in p70 dimer docking to a G-protein-coupled receptor kinase (GRKs) that phosphorylates and desensitizes the receptor. The phosphorylated receptor is recruited by  $\beta$ -arrestin, thus recruiting src kinase, which phosphorylates the EGF-receptor, generating its active conformation (Lin et al., 1997; Ahn et al., 1999; Luttrell et al., 1999). The transactivated RTKs, in turn, activate Ras, which leads to the activation of MAP kinases (ERK 1,2) via Raf. The G<sub>q</sub> subunit, which is PTX-sensitive, inhibits AC, resulting in decreased levels of cyclic-AMP (cAMP). The opposite cellular effects by LPA, that is, mitogenesis and antimitogenesis, are accompanied by opposing effects on the cAMP second messenger system. Mitogenesis is mediated through the G<sub>i</sub> pathway, which results in decreased levels of cAMP (van Corven et al., 1989; van Corven et al., 1992), whereas antimitogenesis is accompanied by a non-PTX sensitive  $Ca^{2+}$ -dependent elevation of cAMP (Tigyi et al., 1994; Fischer et al., 1998).

In contrast, very little is known about the PTX-insensitive G<sub>13</sub> signaling pathway, which leads to the rearrangement of the actin-cytoskeleton. This pathway may also involve the transactivation of RTKs (Lin et al., 1997; Ahn et al., 1999; Luttrell et al., 1999; Gohla et al., 1998) and converge on a small GTPase, Rho (Moolenaar, 1997). Much more is known about the down-stream signaling of Rho because various protein partners have been isolated and identified. Rho activates Ser/Thr kinases, which phosphorylate, and as a result inhibit, myosin light chain phosphatase (MLC-phosphatase) (Kimura et al., 1996). This path results in the accumulation of the phosphorylated form of MLC, leading to cytoskeletal responses that lead to cellular effects like retraction of neurites (Tigyi and Miledi, 1992; Tigyi et al., 1996; Dyer et al., 1992; Postma et al., 1996; Sato et al., 1997), induction of stress fibers (Ridley and Hall, 1992; Gonda et al., 1999), stimulation of chemotaxis (Jalink et al., 1993a), cell migration (Zhou et al., 1995; Kimura et al., 1992), and tumor cell invasiveness (Imamura et al., 1993; Imamura et al., 1996). The PLG-induced Rho-mediated, tumor cell invasiveness is blocked by *C. Botulinum* C3-toxin, which specifically ribosylates Rho in an ADP-dependent mechanism (Imamura et al., 1996).

Rho also has the ability to stimulate DNA synthesis in quiescent fibroblasts (Machesky and Hall, 1996; Ridley, 1996). The expression of Rho family GTPase activates serum-response factor (SRF), which mediates early gene transcription (Hill et al., 1995). Furthermore, PLGF (GPA) induces tumor cell

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EDG-1  
EDG-2  
EDG-3  
EDG-4  
EDG-5  
EDG-6  
EDG-7  
EDG-8  
PSF24

Cardiovascular, CNS, Gonadal tissue, GI  
Cardiovascular, Leukocyte  
Leukocyte, Testes  
Cardiovascular, CNS, Gonadal tissue, Placenta  
Lymphoid, Hematopoietic tissue  
Heart, Pancreas, Prostate, Testes  
Brain  
CNS

Human Tissue with Highest Expression

invasion (Inamura et al., 1996); however, it is still unclear whether it involves cytoskeletal changes or gene transcription, or both.

By virtue of LPA/LPA receptor involvement in a number of cellular pathways and cell activities such as proliferation and/or migration, as well as their implication in wound healing and cancer, it would be desirable to identify novel compounds which are capable of acting, preferably selectively, as either antagonists or agonists at the LPA receptors identified above.

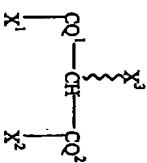
There are currently very few synthetic or endogenous LPA receptor inhibitors which are known. Of the antagonists reported to date, the most work was done on SPP, N-palmitoyl-L-serine (Bittman et al., 1996), and N-palmitoyl-L-tyrosine (Bittman et al., 1996). It is known that the above-mentioned compounds inhibit LPA-induced chloride currents in the *Xenopus* oocyte (Bittman et al., 1996; Zsivs et al., 1998). However, these compounds have not been studied in all cell systems. It is also known that SPP inhibits tumor cell invasiveness, but it is uncertain whether SPP does so by being an inhibitor of LPA or via the actions of its own receptors. N-palmitoyl-L-serine and N-palmitoyl-L-tyrosine also inhibited LPA-induced platelet aggregation (Sugara et al., 1994), but it remains to be seen whether these compounds act at the LPA receptor. Lysophosphatidyl glycerol (LPG) was the first lipid to show some degree of inhibition of LPA actions (van der Bend et al., 1996). None of these inhibitors was shown to selectively act at specific LPA receptors.

A polysulfonated compound, Suramin, was shown to inhibit LPA-induced DNA synthesis in a reversible and dose-dependent manner. However, it was shown that Suramin does not have any specificity towards the LPA receptor and blocked the actions of LPA only at very high millimolar (mM) concentrations (van Corven et al., 1992).

The present invention is directed to overcoming the deficiencies associated with current LPA agonists and LPA antagonists.

## SUMMARY OF THE INVENTION

The present invention relates to compounds according to formula (I) as follows:



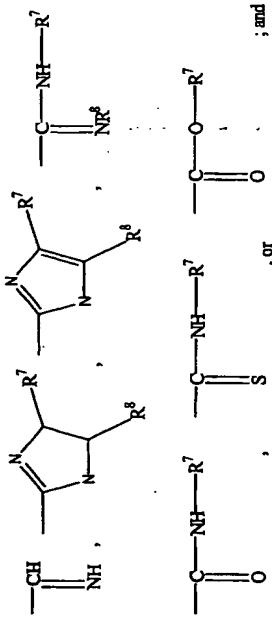
wherein,

at least one of X<sup>1</sup>, X<sup>2</sup>, and X<sup>3</sup> is (HO)<sub>2</sub>PO-Z<sup>1</sup>- or (HO)<sub>2</sub>PO-Z<sup>2</sup>-P(OR)<sub>2</sub>-Z<sup>1</sup>-, X<sup>1</sup> and X<sup>2</sup> are linked together as —O—PO(OH)—O—, or X<sup>1</sup> and X<sup>2</sup> are linked together as —O—PO(OH)—NH—; at least one of X<sup>1</sup>, X<sup>2</sup>, and X<sup>3</sup> is R<sup>1</sup>—Y<sup>1</sup>—A— with each being the same or different when two of X<sup>1</sup>, X<sup>2</sup>, and X<sup>3</sup> are R<sup>1</sup>—Y<sup>1</sup>—A—, or X<sup>2</sup> and X<sup>3</sup> are linked together as —N(H)—C(O)—N(R<sup>1</sup>);

optionally, one of X<sup>1</sup>, X<sup>2</sup>, and X<sup>3</sup> is H; A is either a direct link, (CH<sub>2</sub>)<sub>k</sub> with k being an integer from 0 to 30, or O; Y<sup>1</sup> is —(CH<sub>2</sub>)<sub>m</sub>— with m being an integer from 1 to 30, —O—, O—, —S—, or —NR<sup>2</sup>—;

Z<sup>1</sup> is —(CH<sub>2</sub>)<sub>m</sub>— or —O(CH<sub>2</sub>)<sub>m</sub>— with m being an integer from 1 to 50, —C(R<sup>3</sup>)H—, —NH—, —O—, or —S—; Z<sup>2</sup> is —(CH<sub>2</sub>)<sub>n</sub>— or —O(CH<sub>2</sub>)<sub>n</sub>— with n being an integer from 1 to 50 or —O—; Q<sup>1</sup> and Q<sup>2</sup> are independently H, =NR<sup>4</sup>, =O, or a combination of H and —NR<sup>5</sup>R<sup>6</sup>.

R<sup>1</sup> for each of X<sup>1</sup>, X<sup>2</sup>, or X<sup>3</sup> is independently hydrogen, a straight or branched-chain C1 to C30 alkyl, a straight or branched-chain C2 to C30 alkyl, an aromatic or heteroaromatic ring with or without mono-, di-, or trisubstitution of the ring, an acyl including a C1 to C30 alkyl or an aromatic or heteroaromatic ring, an arylalkyl including straight or branched-chain C1 to C30 alkyl, an arylalkoxyalkyl including straight or branched-chain C1 to C30 alkyl,



5  $R^2, R^3, R^4, R^5, R^6, R^7$ , and  $R^8$  are independently hydrogen, a straight or branched-chain C1 to C30 alkyl, a straight or branched-chain C2 to C30 alkenyl, an aromatic or heteroaromatic ring with or without mono-, di-, or tri-  
10 substitutions of the ring, an acyl including a C1 to C30 alkyl or aromatic or heteroaromatic ring, an arylalkyl including straight or branched-chain C1 to C30 alkyl, or an arylalkoxy including straight or branched-chain C1 to C30 alkyl;

wherein the compound of formula I is not lysophosphatidic acid, phosphatidic acid, cyclic phosphatidic acid, alkanyl glycerolphosphate, dioctyl glycerol pyrophosphate, or N-palmitoyl-L-serine.

Also disclosed are pharmaceutical compositions which include a pharmaceutically-acceptable carrier and a compound of the present invention.

15 A further aspect of the present invention relates to a method of inhibiting LPA activity on an LPA receptor which includes providing a compound of the present invention which has activity as an LPA receptor antagonist and contacting an LPA receptor with the compound under conditions effective to inhibit LPA-induced activity of the LPA receptor.

Another aspect of the present invention relates to a method of modulating LPA receptor activity which includes providing a compound of the present invention which has activity as either an LPA receptor agonist or an LPA receptor antagonist and contacting an LPA receptor with the compound under conditions effective to modulate the activity of the LPA receptor.

20 Still another aspect of the present invention relates to a method of treating cancer which includes providing a compound of the present invention and administering an effective amount of the compound to a patient in a manner effective to treat cancer.

Yet another aspect of the present invention relates to a method of enhancing cell proliferation which includes providing a compound of the present invention which has activity as an agonist of an LPA receptor and contacting the LPA receptor on a cell with the compound in a manner effective to enhance LPA receptor-induced proliferation of the cell.

5 A further aspect of the present invention relates to a method of treating a wound which includes providing a compound of the present invention which has activity as an agonist of an LPA receptor and delivering an effective amount of the compound to a wound site, where the compound binds to LPA receptors on cells that promote healing of the wound, thereby stimulating LPA receptor agonist-induced cell proliferation to promote wound healing.

10 A still further aspect of the present invention relates to a method of making the compounds of the present invention. One approach for making the compounds of the present invention includes:

15 reacting  $(Y^2O)_xPO—Z^{11}—Z^{13}$  or  $(Y^2O)_xPO—Z^{12}—P(OH)O—Z^{11}—Z^{13}$ , where

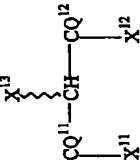
$Z^{11}$  is  $—(CH_2)_m—$  or  $—O(CH_2)_m—$  with  $m$  being an integer from 1 to 50,  $—CR^3H—$ , or  $—O—$ ;

$Z^{12}$  is  $—(CH_2)_n—$  or  $—O(CH_2)_n—$  with  $n$  being an integer from 1 to 50 or  $—O—$ ;

$Z^{13}$  is H or a first leaving group or  $—Z^{11}—Z^{13}$  together form the first leaving group; and

20  $Y^2$  is H or a protecting group, with an intermediate compound according to formula (VI)

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where,

at least one of  $X^{11}$ ,  $X^{12}$ , and  $X^{13}$  is  $R^{11}—Y^{11}—A—$  with each being the same or different when two of  $X^{11}$ ,  $X^{12}$ , and  $X^{13}$  are  $R^{11}—Y^{11}—A—$ , or  $X^{12}$  and  $X^{13}$  are linked together as  $—N(H)—C(O)N(R^{11})—$ ;

at least one of  $X^{11}$ ,  $X^{12}$ , and  $X^{13}$  is  $\text{OH}$ ,  $\text{NH}_2$ ,  $\text{SH}$ , or a second leaving group;

optionally, one of  $X^{11}$ ,  $X^{12}$ , and  $X^{13}$  is  $\text{H}$ ;

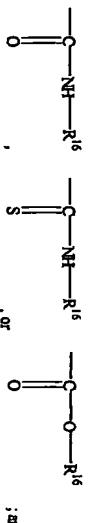
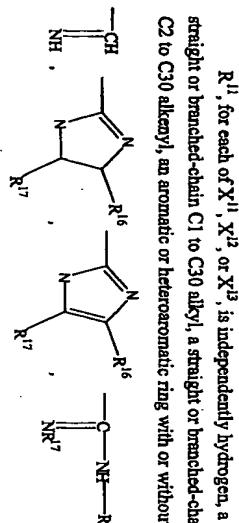
$A$  is either a direct link,  $(\text{CH}_2)_k$  with  $k$  being an integer from 0 to 30, or  $\text{O}$ ;

$Y^{11}$  is  $-(\text{CH}_2)_l$  with  $l$  being an integer from 1 to 30,  $-\text{O}-$ ,

$\text{O}$   
 $\parallel$   
 $-\text{C}-$ ,  $-\text{S}-$ , or  $-\text{NR}^{12}-$ ,

$Q^1$  and  $Q^2$  are independently  $\text{H}_2$ ,  $=\text{NR}^{13}$ ,  $=\text{O}$ , a combination of  $\text{H}$  and  $-\text{NR}^{14}\text{R}^{15}$ ;

$R^{11}$ , for each of  $X^{11}$ ,  $X^{12}$ , or  $X^{13}$ , is independently hydrogen, a straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl, a straight or branched-chain  $\text{C}2$  to  $\text{C}30$  alkenyl, an aromatic or heteroaromatic ring with or without



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mono-, di-, or tri-substitutions of the ring, an acyl including a  $\text{C}1$  to  $\text{C}30$  alkyl or an aromatic or heteroaromatic ring, an arylalkyl including straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl, an aryloxymethyl including straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl,

$R^{12}$ ,  $R^{13}$ ,  $R^{14}$ ,  $R^{15}$ , and  $R^{17}$  are independently hydrogen, a

straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl, a straight or branched-chain  $\text{C}2$  to  $\text{C}30$  alkenyl, an aromatic or heteroaromatic ring with or without

mono-, di-, or tri-substitutions of the ring, an acyl including a  $\text{C}1$  to  $\text{C}30$  alkyl or aromatic or heteroaromatic ring, an arylalkyl including straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl, or an aryloxymethyl including straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl;

including straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl, a straight or branched-chain  $\text{C}2$  to  $\text{C}30$  alkenyl, an aromatic or heteroaromatic ring with or without mono-, di-, or tri-substitutions of the ring, an acyl including a  $\text{C}1$  to  $\text{C}30$  alkyl or aromatic or heteroaromatic ring, an arylalkyl including straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl, or an aryloxymethyl including straight or branched-chain  $\text{C}1$  to  $\text{C}30$  alkyl;

followed by a de-protection step, if necessary, with both said reacting and the deprotection step being performed under conditions effective to afford a compound according to formula (1) where one or two of  $X^1$ ,  $X^2$ , and  $X^3$  is  $(\text{HO})_2\text{PO}-\text{Z}^1-$  or  $(\text{HO})_2\text{PO}-\text{Z}^2-\text{P}(\text{OH})\text{O}-\text{Z}^1-$ .

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The compounds of the present invention which have been identified herein as being either agonists or antagonists of one or more LPA receptors find uses to inhibit or enhance, respectively, biochemical pathways mediated by LPA receptor signaling. By modulating LPA receptor signaling, the antagonists and agonists find specific and substantial uses in treating cancer and enhancing wound healing.

#### BRIEF DESCRIPTION OF THE DRAWINGS

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Figure 1 is a phylogenetic tree illustrating the classification and relatedness of ten phospholipid growth factor receptors, including LPA receptors EDC-2, EDC-4, EDC-7, and PSP-24 ( $\alpha, \beta$ ).

Figure 2 illustrates the synthesis scheme employed for preparation of serine amide compounds 35-43.

Figure 3 illustrates the synthesis scheme employed for preparation of serine amide phosphate compounds 55-59.

Figure 4 illustrates the synthesis scheme employed for preparation of bisphosphate compounds 66-68.

Figures 5A-B illustrate synthesis of bisphosphate compounds. Figure 5A illustrates the synthesis scheme employed for preparation of 1,2-bisphosphate compounds 85-92. Figure 5B illustrates a synthesis scheme for preparing 1,3-bisphosphate compounds.

Figures 6A-C illustrate synthesis schemes for preparation of pyrophosphate compounds.

Figures 7A-C illustrate synthesis schemes for preparation of substituted mono-phosphates and mono-phosphonates from a tosylate-protected di-ether intermediate.

Figure 8 illustrates the synthesis scheme employed for preparation of straight-chain fatty acid phosphate compounds 106-110.

Figure 9 illustrates synthesis of straight-chain thiophosphate acid monoalkyl esters.

Figure 10 illustrates synthesis of straight-chain alkylamido-phosphonic acid.

Figure 11 illustrates a synthesis scheme for preparation of conformationally restrained, cyclic phosphate compounds.

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Figures 12A-C illustrate synthesis schemes for preparation of substituted mono-phosphates and mono-phosphonates from a tosylate-protected di-ether intermediate.

Figure 13 illustrates the synthesis scheme employed for preparation of straight-chain fatty acid phosphate compounds 116-110.

Figure 14 illustrates synthesis of straight-chain thiophosphate acid monoalkyl esters.

Figure 15 illustrates synthesis of straight-chain alkylamido-phosphonic acid.

Figure 16 illustrates a synthesis scheme for preparation of conformationally restrained, cyclic phosphate compounds.

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Figures 17A-C illustrate synthesis schemes for preparation of substituted mono-phosphates and mono-phosphonates from a tosylate-protected di-ether intermediate.

Figure 18 illustrates the synthesis scheme employed for preparation of straight-chain fatty acid phosphate compounds 116-110.

Figure 19 illustrates synthesis of straight-chain thiophosphate acid monoalkyl esters.

Figure 20 illustrates synthesis of straight-chain alkylamido-phosphonic acid.

Figure 21 illustrates a synthesis scheme for preparation of conformationally restrained, cyclic phosphate compounds.

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Figures 22A-C illustrate synthesis schemes for preparation of substituted mono-phosphates and mono-phosphonates from a tosylate-protected di-ether intermediate.

Figure 23 illustrates the synthesis scheme employed for preparation of straight-chain fatty acid phosphate compounds 116-110.

Figure 24 illustrates synthesis of straight-chain thiophosphate acid monoalkyl esters.

Figure 25 illustrates synthesis of straight-chain alkylamido-phosphonic acid.

Figure 26 illustrates a synthesis scheme for preparation of conformationally restrained, cyclic phosphate compounds.

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Figure 12 illustrates a synthesis scheme for preparation of conformationally restrained, cyclic phosphate compounds. Figure 13 illustrates a synthesis scheme for preparation of conformationally restrained, cyclic phosphate compounds. Figure 14 illustrates a synthesis scheme for preparation of conformationally restrained compounds with a free phosphate moiety. Figure 15 illustrates an alternative synthesis scheme for preparing 2-monophosphates.

Figure 16 illustrates an alternative synthesis scheme for preparing 1,3-bisphosphate compounds.

Figure 17 illustrates a synthesis scheme for preparing compounds having an  $-\text{N}(\text{R})-\text{acyl}$  group as  $\text{X}^3$ .

Figure 18 illustrates a synthesis scheme for preparing compounds having an  $-\text{N}(\text{R})-\text{imidazole}$  group as  $\text{X}^3$ .

Figure 19 illustrates a synthesis scheme for preparing compounds having an  $-\text{N}(\text{R})-\text{C}(\text{O})-\text{O}-\text{R}'$  as  $\text{X}^3$ .

Figure 20 illustrates a synthesis scheme for preparing compounds having an  $-\text{N}(\text{R})-\text{C}(\text{S})-\text{O}-\text{R}'$  as  $\text{X}^3$ .

Figure 21 is a graph illustrating the dose-dependent inhibition of LPA-induced chloride currents in *Xenopus* oocytes by extracellular application of 56 (SAP, 14:0).

Figure 22 is a graph illustrating the dose-dependent inhibition of LPA-induced chloride currents in *Xenopus* oocytes by extracellular application of 57 (SAP, 18:0).

Figures 23A-B are graphs illustrating the dose-dependent inhibition of LPA-induced chloride currents in *Xenopus* oocytes by extracellular application of 66 (MAGDP, 18:0). The arrow indicates the time of the intracellular injection of 5  $\mu\text{M}$  66, followed by the extracellular application of LPA.

Figure 24 is a graph illustrating dose-inhibitory effect of 66 (MAGDP, 18:0). A constant amount of LPA (5 nM) was applied to oocytes together with increasing amounts of 66. Data points represent the peak amplitude of the measured chloride currents.

Figure 25 is a graph illustrating the dose-dependent inhibition of LPA-induced chloride currents in *Xenopus* oocytes by extracellular application of 92 (MAGDP, 22:0).

Figure 26 is a graph illustrating the dose-dependent effect of 56a (SDAP, 14:0/2:0) on *Xenopus* oocytes.

Figure 27 is a bar graph depicting the effects of compounds 56 (SAP, 14:0), 56a (SDAP, 14:0/2:0), and 66 (MAGDP, 18:0) on LPA-induced HEV cell migration. Test compound concentration was 1  $\mu\text{M}$ ; LPA concentration was 0.1  $\mu\text{M}$ .

Figures 28A-C are graphs illustrating the dose response relationship for  $\text{Ca}^{2+}$  responses in RH7777 cells heterologously expressing Edg-2 (28A), Edg-4 (28B), or Edg-7 (28C). Each data point represents the average of at least three measurements  $\pm$  S.D.

Figures 29A-D are graphs illustrating DGPP 8:0 inhibition of  $\text{Ca}^{2+}$  responses elicited by LPA in Edg-2 and -7, but not Edg-4 expressing RH7777 cells.

Figures 29A-C are graphs illustrating the dose response relationship for  $\text{Ca}^{2+}$  responses in RH7777 cells expressing Edg-2, 4, or -7, were exposed to a mixture of 100 nM LPA 18:1 and 1  $\mu\text{M}$  DGPP 8:0. Control cells were exposed to 100 nM LPA 18:1. Representative  $\text{Ca}^{2+}$  responses are shown for stable Edg-2 (29A), Edg-4 (29B), and Edg-7 (29C) expressing cells, or cells transiently expressing Edg-4 (29D).

Figures 30A-C are graphs which illustrate the pharmacological characterization of the inhibition of the LPA response by DGPP 8:0 in RH7777 cells expressing Edg-7 (Edg-7 cells). Cells were exposed to a 250 nM concentration of LPA 18:1 mixed with increasing concentrations of DGPP 8:0 and the peak area of the resulting  $\text{Ca}^{2+}$  responses were measured (30A). Cells were also exposed to increasing concentrations of LPA 18:1 mixed with a 500 nM concentration of DGPP 8:0 (30B).

Figures 31A-C are graphs which illustrate the pharmacological characterization of the inhibition of the LPA response by DGPP 8:0 in RH7777 cells expressing Edg-2 (Edg-2 cells). Stable Edg-2 cells exposed to a 250 nM concentration of LPA 18:1 mixed with a 500 nM concentration of LPA 18:1 mixed with a 500 nM concentration of DGPP 8:0 (31B).

Figures 32A-B are graphs which illustrate the structure-activity relationship for DGPP in Edg-4-expressing RH7777 cells. Stable Edg-4 cells were exposed to a 500 nM concentration of LPA 18:1 mixed with a 5  $\mu\text{M}$  concentration of the indicated lipids (32A). Cells transiently expressing Edg-4 cells were exposed to a 100 nM concentration of LPA 18:1 mixed with a 1  $\mu\text{M}$  concentration of the indicated

Figures 33A-B are graphs which illustrate the structure-activity relationship for DGPP in Edg-4-expressing RH7777 cells. Stable Edg-4 cells were exposed to a 500 nM concentration of LPA 18:1 mixed with a 5  $\mu\text{M}$  concentration of the indicated lipids (33A). Cells transiently expressing Edg-4 cells were exposed to a 100 nM concentration of LPA 18:1 mixed with a 1  $\mu\text{M}$  concentration of the indicated

Figures 34A-B are graphs which illustrate the structure-activity relationship for DGPP in Edg-4-expressing RH7777 cells. Stable Edg-4 cells were exposed to a 500 nM concentration of LPA 18:1 mixed with a 5  $\mu\text{M}$  concentration of the indicated lipids (34A). Cells transiently expressing Edg-4 cells were exposed to a 100 nM concentration of LPA 18:1 mixed with a 1  $\mu\text{M}$  concentration of the indicated

Figures 35A-B are graphs which illustrate the structure-activity relationship for DGPP in Edg-4-expressing RH7777 cells. Stable Edg-4 cells were exposed to a 500 nM concentration of LPA 18:1 mixed with a 5  $\mu\text{M}$  concentration of the indicated lipids (35A). Cells transiently expressing Edg-4 cells were exposed to a 100 nM concentration of LPA 18:1 mixed with a 1  $\mu\text{M}$  concentration of the indicated

## DETAILED DESCRIPTION OF THE INVENTION

lipids (32B). The peak areas of the  $\text{Ca}^{2+}$  responses were measured and are represented as the average values of a minimum of three measurements  $\pm$  S.D.

Figures 33A-C are graphs which illustrate the pharmacological characterization of DGPP 8:0 on the LPA-elicited  $\text{Cl}^-$  currents in *Xenopus* oocytes. Oocytes were exposed to a 5 nM concentration of LPA 18:1 mixed with increasing concentrations of DGPP 8:0 and the peak amplitude of the resulting oscillatory  $\text{Cl}^-$  currents were measured (33A). Oocytes were exposed to increasing concentrations of LPA 18:1 mixed with a 200 nM concentration of DGPP 8:0 (33B). Data points represent the average values of a minimum of three measurements  $\pm$  S.D. Oocytes were treated with 5 nM LPA 18:1, or a mixture of 5 nM LPA 18:1 and 1  $\mu\text{M}$  DGPP 8:0 as indicated (33C). The intracellular injection of 1  $\mu\text{M}$  DGPP 8:0 is indicated by the arrow.

Figures 34A-D are graphs which illustrate DGPP 8:0 inhibiting the LPA-elicited  $\text{Ca}^{2+}$  responses in NIH3T3 fibroblasts and HEY ovarian cancer cells.

RT-PCR analysis of NIH3T3 cells for Edg and PSP24 receptor transcripts (34A). NIH3T3 cells were exposed to a 100 nM concentration of LPA 18:1, or SfP, mixed with a 10  $\mu\text{M}$  concentration of DGPP 8:0 (34B). RT-PCR analysis of HEY cells for the presence of the Edg and PSP24 transcripts (34C). HEY cells were exposed to a 100 nM concentration of LPA 18:1, or SfP, mixed with a 1  $\mu\text{M}$  concentration of DGPP 8:0 (34D). The peak areas of the resulting  $\text{Ca}^{2+}$  responses were measured and are represented as the average of a minimum of three measurements  $\pm$  S.D.

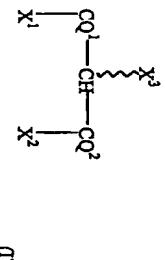
Figure 35 is a graph illustrating DGPP 8:0 inhibition of LPA-elicited proliferation of NIH3T3 cells. NIH3T3 cells were serum-starved for 6 hr and exposed to a 5  $\mu\text{M}$  concentration of LPA 18:1 mixed with a 10  $\mu\text{M}$  concentration of the indicated lipids. Control cells received solvent (BSA) in place of LPA 18:1. The cells were incubated for 24 hr with the lipids and counted. Data are representative of three experiments.

Figure 36 is a graph which illustrates the pharmacological characterization of the inhibition of the LPA response by straight-chain fatty acid phosphate compounds 106-110 in *Xenopus* oocytes.

Figure 37 is a graph which illustrates the pharmacological characterization of the inhibition of the LPA response by straight-chain fatty acid phosphate compound 108 in *Xenopus* oocytes.

Figure 38 is a graph illustrating the pharmacological characterization of the antagonist or agonist induced response of RH7777 cells individually expressing Edg-2, Edg-4, or Edg-7 receptors, following exposure of the cells to straight-chain fatty acid phosphate compound 108. Peak areas of the  $\text{Ca}^{2+}$  responses were measured.

One aspect of the present invention relates to a compound according to formula (I)



wherein,

at least one of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  is  $(\text{HO})_2\text{PO---Z}^2\text{---P}(\text{O}(\text{OH})\text{---O---})\text{---}$  or  $\text{O---}$ , or  $\text{X}^1$  and  $\text{X}^2$  are linked together as  $---\text{O---PO}(\text{OH})\text{---}$  or  $\text{X}^1$  and  $\text{X}^3$  are linked together as  $---\text{O---PO}(\text{OH})\text{---NH---}$ ;

at least one of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  is  $\text{R}^1\text{---Y}^1\text{---A---}$  with each

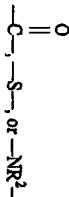
$\text{X}^3$  being the same or different when two of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  are  $\text{R}^1\text{---Y}^1\text{---A---}$ , or  $\text{X}^1$  and  $\text{X}^3$  are linked together as  $---\text{N}(\text{R})\text{---C}(\text{O})\text{---NR}^1\text{---}$ ;

optionally, one of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  is H;

$\text{A}$  is either a direct link,  $(\text{CH}_2)_k$  with  $k$  being an integer from 0

to 30, or O;

$\text{Y}^1$  is  $---(\text{CH}_2)_l---$  with  $l$  being an integer from 1 to 30,  $---\text{O---}$ ,



$\text{Z}^1$  is  $---(\text{CH}_2)_m---$  or  $---\text{O}(\text{CH}_2)_m---$  with  $m$  being an integer from 1 to 50,  $---\text{C}(\text{R}^3)\text{H}_n---$ ,  $---\text{NH---}$ ,  $---\text{O---}$ , or  $---\text{S---}$ ;

$\text{Z}^2$  is  $---(\text{CH}_2)_k---$  or  $---\text{O}(\text{CH}_2)_k---$  with  $n$  being an integer from 1 to 50 or  $---\text{O---}$ ;

$\text{Q}^1$  and  $\text{Q}^2$  are independently  $\text{H}$ ,  $=\text{NR}^4$ ,  $=\text{O}$ , a combination of H

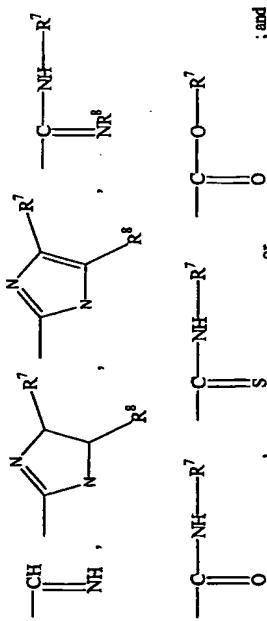
and  $-\text{NR}^5\text{R}^6$ ;

$\text{R}^1$ , for each of  $\text{X}^1$ ,  $\text{X}^2$ , or  $\text{X}^3$ , is independently hydrogen, a

straight or branched-chain C1 to C30 alkyl, a straight or branched-chain C2 to C30 alkyl, an aromatic or heteroaromatic ring with or without mono-, di-, or tri-

substitutions of the ring, an acyl including a C1 to C30 alkyl or an aromatic or heteroaromatic ring, an arylalkyl including straight or branched-chain C1 to C30 alkyl,

an arylalkoxy including straight or branched-chain C1 to C30 alkyl,



5  $R^2, R^3, R^4, R^5, R^6, R^7$ , and  $R^8$  are independently hydrogen, a straight or branched-chain C1 to C30 alkyl, a straight or branched-chain C2 to C30 alkenyl, an aromatic or heteroaromatic ring with or without mono-, di-, or tri-substitutions of the ring, an acyl including a C1 to C10 alkyl or aromatic or an arylalkoxyl including straight or branched-chain C1 to C30 alkyl, or an arylalkyl including straight or branched-chain C1 to C30 alkyl, or an arylalkoxyl including straight or branched-chain C1 to C30 alkyl.

For each of the above-identified R groups (e.g.,  $R^1 - R^5$ ), it is intended that straight chain alkyls have the formula  $-(CH_2)_xCH_3$  where  $x$  is from 0 to 29; branched chain alkyls have the formula as defined above for straight chain alkyl, except that one or more  $CH_2$  groups are replaced by CW groups where W is an alkyl side chain; straight chain alkenyls have the formula  $-(CH_2)_{x_1}CH=CH(CH_2)_{x_2}CH_3$  where  $x_1$  and  $x_2$  each are from 0 to 27 and  $(x_1 + x_2)$  is not more than 27; and branched chain alkenyls have the formula as defined above for straight chain alkenyl, except that one or more  $CH_2$  groups are replaced by CW groups or a CH group is replaced by a CW group, where W is an alkyl side chain.

Aromatic or heteroaromatic rings include, without limitation, phenyls, indenes, pyrroles, imidazoles, oxazoles, pyrazoles, pyridines, pyrimidines, pyrrolidines, piperidines, furans, naphthalins, bi-phenyls, and indoles. The aromatic or heteroaromatic rings can include mono-, di-, or tri-substitutions of the ring located at the *ortho*, *meta*, or *para* positions on the rings relative to where the ring binds to the  $Y^1$  group of the  $R^1 - Y^1 - A -$  chain. Substitutions on the rings can include, without limitation, alkyl, alkoxy, amine (including secondary or tertiary amines), alkylamine, amide, alkylamide, acids, alcohols.

Acyl groups can include either alkyl, alkenyl, or aromatic or heteroaromatic rings as described above.

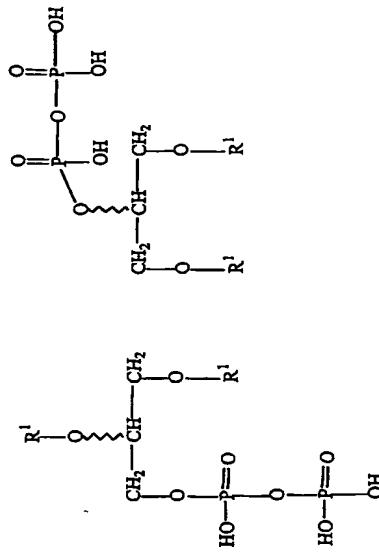
Arylalkyl and arylalkoxyl groups can include, without limitation, straight or branched-chain C1 to C30 alkyl groups as described above, with the alkyl group binding to the  $Y^1$  group of the  $R^1 - Y^1 - A -$  chain.

Specifically excluded from the above-identified definition of the compound according to formula (I) are the following previously known endogenous or synthetic compounds: lysophosphatidic acid, phosphatidic acid, cyclic phosphatidic acid, alkenyl glycerolphosphate, diocyl glycerol pyrophosphate, and N-palmitoyl-L-serine.

Exemplary compounds according to formula (I) are the subclass compounds according to formulas (II)-(V) below.

In the structures of formulas (II)A and (II)B,  $Q^1$  and  $Q^2$  are both  $H_2$ ; one of  $X^1, X^2$ , and  $X^3$  is  $(HO)_2PO - Z^2 - P(OH)O - Z^1 -$ , with  $Z^1$  and  $Z^2$  being O, and two of  $X^1, X^2$ , and  $X^3$  are  $R^1 - Y^1 - A -$ , with A being a direct link and  $Y^1$  being O for each. Each  $R^1$  is defined independently as above for formula (I).

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(II)A

(II)B

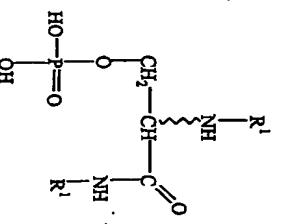
In the structures of formulas (III),  $Q^1$  is  $H_2$ ,  $Q^2$  is  $=O$ ;  $X^1$  is  $(HO)_2PO - Z^1 -$ , with  $Z^1$  being O, and  $X^2$  and  $X^3$  are  $R^1 - Y^1 - A -$ , with A being a direct link and  $Y^1$  being  $-NH-$  for each. Each  $R^1$  is defined independently as above for formula (I). Preferred species of within the scope of formula (III) are where  $X^3$  is  $-NH_2$  and  $X^2$  is  $-NHR^1$  with  $R^1$  being a C14 to C18 alky, more preferably either a

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$(HO)_2PO - Z^1 -$ , with  $Z^1$  being O, and  $X^2$  and  $X^3$  are  $R^1 - Y^1 - A -$ , with A being a direct link and  $Y^1$  being  $-NH-$  for each. Each  $R^1$  is defined independently as above for formula (I). Preferred species of within the scope of formula (III) are where  $X^3$  is  $-NH_2$  and  $X^2$  is  $-NHR^1$  with  $R^1$  being a C14 to C18 alky, more preferably either a

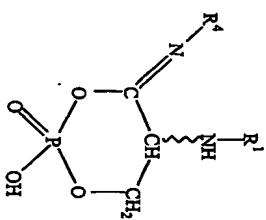
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C14 alkyl or a C18 alkyl; or where X<sup>2</sup> is —NHR<sup>1</sup> with R<sup>1</sup> being an acetyl group and X<sup>2</sup> is —NHR<sup>1</sup> with R<sup>1</sup> being a C14 alkyl.

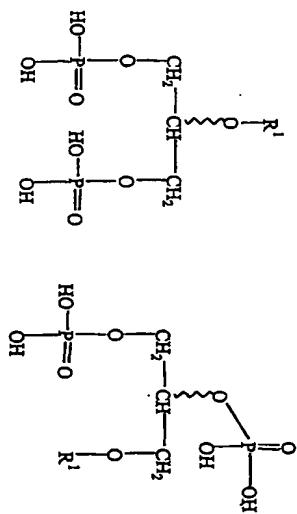


(III)

In the structures of formula (IV), Q<sup>1</sup> is —NR<sup>4</sup>; Q<sup>2</sup> is H<sub>2</sub>; X<sup>1</sup> and X<sup>2</sup> are linked together as —O—PO(OH)—O—; and X<sup>3</sup> is R<sup>1</sup>—Y<sup>1</sup>—A—, with A being a direct link and Y<sup>1</sup> being —NH—. R<sup>1</sup> and R<sup>4</sup> are as defined above for formula (I).



(IV)



(V)A

(V)B

The compounds according to formula (I), as well as the subgenus compounds according to formulae (II)A, (II)B, (III), (IV), (V)A, and (V)B, can be prepared using the synthesis schemes described below.

To synthesize the serine amides (SA) and serine amide phosphate (SAP) series (formula (II)), the precursor *t*-Boc protected  $\beta$ -fucose (25) was first synthesized. Starting with commercially available *t*-Boc-L-serine (Figure 2, 24), triphenyl phosphite (PPh<sub>3</sub>) and diethylazidodicarboxylate (DEAD) were introduced under Mitsunobu conditions, affording compound 25 in ca. 50% yield (Sun et al., 1996). Attempts using procedure developed by Sun et al. to open the highly labile  $\beta$ -lactone 25 with various primary amines to obtain hydroxy amides 26-34 failed, in spite of using various reagents (triethyl amine, etc.). Instead, by refluxing the primary amines with the  $\beta$ -lactone in THF, the *t*-Boc protected hydroxy amides 26-34 were obtained. Compounds 26-34 were purified using flash column chromatography.

Trifluoroacetic acid (TFA)-mediated removal of the *t*-Boc protecting group afforded compounds 35-43 as TFA salts.

To synthesize compounds 55-59, the *t*-Boc protected hydroxy amides 26-30 were phosphorylated. A careful study of the final compound suggested that the final compound would possess a highly hydrophobic region and a highly hydrophilic region. Both regions may cause problems during the extraction process and/or attach to the column during the purification stage. To circumvent these potential problems, phosphoramidate chemistry was employed. By using phosphoramidate chemistry, it was hypothesized that the phosphate hydroxyl groups could be protected to render the molecule completely hydrophobic, thereby facilitating its smooth purification.

10 R<sup>1</sup> is a C18 alkyl.

15 In the structures of formulae (V)A and (V)B, Q<sup>1</sup> and Q<sup>2</sup> are both H<sub>2</sub>; two of X<sup>1</sup>, X<sup>2</sup>, and X<sup>3</sup> are (PO(OH))<sub>2</sub>Z<sup>1</sup>—, with Z<sup>1</sup> being O for each; and one of X<sup>1</sup>, X<sup>2</sup>, and X<sup>3</sup> is R<sup>1</sup>—Y<sup>1</sup>—A—, with A being a direct link and Y<sup>1</sup> being —O—. R<sup>1</sup> is as defined above for formula (I). Preferred species within the scope of formulae (V)A and (V)B include the compounds where R<sup>1</sup> is an acyl including a C21 alkyl or where R<sup>1</sup> is a C18 alkyl.

Essentially, a combination of procedures was used to obtain the desired products (55-59) (Lynch et al., 1997; Bitman et al., 1996; Liu et al., 1999). Starting hydroxymides (26-30) were repeatedly washed with anhydrous pyridine, and dried in high vacuum for over 48 hrs. The pyridine-washed hydroxymides were maintained under an atmosphere of argon. 1H-terazole and a freshly distilled 1:1 mixture of THF/CH<sub>2</sub>Cl<sub>2</sub> were then added. The phosphorylating agent, dibenzylidisisopropyl phosphoramide, was added. After monitoring the reaction by TLC, the phosphonate was oxidized to the phosphate *in situ* with peracetic acid. The reaction mixture was purified via column chromatography to afford compounds 50-54 as benzyl-protected phosphates. The removal of the protecting benzyl groups was carried out in ethanol by subjecting compounds 50-54 to catalytic reduction using 10% palladium on activated carbon (Pd/C) under H<sub>2</sub> atmosphere at 60 psi to yield compounds 55-59 (Figure 3). Reacting 56 with acetic anhydride afforded compound 56a (Figure 3).

Once the phosphorylation technique was elucidated for the synthesis of the SAP series (compounds 55-59), a similar procedure was used for the synthesis of bisphosphates (formulae (V)A and (V)B) (Figures 4 and 5A-B). The commercially available diols 60-62 were washed with anhydrous pyridine, and were dried for 48 hrs under high vacuum. These dried diols (60-62) were dissolved in freshly distilled 1:1 THF/CH<sub>2</sub>Cl<sub>2</sub>, followed by the addition of 1H-terazole. To this stirred mixture was added dibenzylidisisopropyl phosphoramide. The reaction mixture was monitored via TLC, and at the appropriate time the phosphate was oxidized to the phosphate *in situ* with peracetic acid. The reaction mixture was purified with column chromatography to afford compounds 63-65 as benzyl-protected bisphosphates. The removal of the protecting benzyl groups was carried out in ethanol by subjecting compounds 63-65 to catalytic reduction using 10% palladium on activated carbon (Pd/C) under H<sub>2</sub> atmosphere at 60 psi to yield compounds 66-68 as bisphosphates. A similar procedure as described above for the synthesis of 66-68 was followed to obtain compounds 65-92.

While compounds 85-92 are 1,2-biphosphates, Figure 5B illustrates the synthesis of 1,3-biphosphates. Commercially available 2-phenoxy-1,3-propane-diol was used as the starting material. The starting compound was first protected with t-BuOK in the presence of methyl iodide, followed by catalytic hydrogenation to give an intermediate which was then reacted with a halide (RX, where R is as defined above for R'). The recovered intermediate was subsequently treated with AlCl<sub>3</sub> in the presence of ethyl-SH to yield a 1,3 diol possessing the RO group bound to C2 of the backbone. The recovered 1,3 diol was dissolved in freshly distilled 1:1 THF/CH<sub>2</sub>Cl<sub>2</sub>, followed by the addition of 1H-terazole. To this stirred mixture was added

dibenzylidisisopropyl phosphoramide. The reaction mixture was monitored via TLC, and at the appropriate time the phosphate was oxidized to the phosphate *in situ* with peracetic acid. The reaction mixture was purified with column chromatography to afford benzyl-protected bisphosphate compounds. Removal of the protecting benzyl groups was carried out in ethanol by subjecting the compounds to catalytic reduction using 10% palladium on activated carbon (Pd/C) under H<sub>2</sub> atmosphere at 60 psi to yield 1,3-bisphosphate compounds.

To synthesize the pyrophosphates of formulae (II)A and (II)B, glycidyl tosylate ((2R)(+)- or (2R)(-)) was used as the starting material (Figures 6A-B). Opening of the ring was catalyzed by a Lewis acid, such as BF<sub>3</sub>, in the presence of an alcohol, affording an intermediate which was tosylate-protected at the C1 position. In the next step, the alcohol at the C2 position was replaced with an R group (e.g., R<sup>1</sup> as described above), using as excess of R-triflate and 2,6-di-*tert*-butyl-4-methylpyridine, affording the di-ether intermediate. Treatment of the di-ether intermediate with tris(tert-a-butylammonium) hydrogen pyrophosphate caused nucleophilic attack of the tosylate, replacing the tosylate with a pyrophosphate of formula (II)B. The tosylate protected intermediate was treated with benzyl alcohol in the presence of triflic anhydride and 2,6-di-*tert*-butyl-4-methylpyridine, which benzylates the intermediate at the C2 position. The tosylate protecting group on the benzylate intermediate was removed first by the action of potassium superoxide in the presence of 18-crown-6, affording a hydroxyl group at the C1 position which was subject to replacement with an R group (e.g., R<sup>1</sup> as described above) using an excess of R-triflate and 2,6-di-*tert*-butyl-4-methylpyridine. The resulting di-ether intermediate still possessed the benzyl protecting group at the C2 position. The benzyl protecting group was removed by hydrogenation and the subsequent hydroxyl group was tosylated by the action of pyridine and *p*-toluenesulfonyl chloride, producing a di-ether bearing a tosyl group at the C2 position. The tosylate group was removed by nucleophilic attack upon treatment with tris(tert-a-butylammonium) hydrogen pyrophosphate, replacing the tosylate with a pyrophosphate substituent at the C2 position.

Alternative schemes for preparing phosphates and biphenophates (as well as pyrophosphates, phosphonates, etc.) are illustrated in Figures 15 and 16. In Figure 15, glycidal bromide was used as the starting material along with an alcohol (ROH). The reaction conditions included treatment with K<sub>2</sub>CO<sub>3</sub>, followed by treatment with the ammonium salt C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>N<sup>+</sup>(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>Cl<sup>-</sup>, resulting in displacement of the bromide with the R group. The ring of the glycidal intermediate was then opened following treatment with 1M HCl in ether and an alcohol (R'-OH),

5 the SAP series (compounds 55-59), a similar procedure was used for the synthesis of bisphosphates (formulae (V)A and (V)B) (Figures 4 and 5A-B). The commercially available diols 60-62 were washed with anhydrous pyridine, and were dried for 48 hrs under high vacuum. These dried diols (60-62) were dissolved in freshly distilled 1:1 THF/CH<sub>2</sub>Cl<sub>2</sub>, followed by the addition of 1H-terazole. To this stirred mixture was added dibenzylidisisopropyl phosphoramide. The reaction mixture was monitored via TLC, and at the appropriate time the phosphate was oxidized to the phosphate *in situ* with peracetic acid. The reaction mixture was purified with column chromatography to afford compounds 63-65 as benzyl-protected bisphosphates. The removal of the protecting benzyl groups was carried out in ethanol by subjecting compounds 63-65 to catalytic reduction using 10% palladium on activated carbon (Pd/C) under H<sub>2</sub> atmosphere at 60 psi to yield compounds 66-68 as bisphosphates. A similar procedure as described above for the synthesis of 66-68 was followed to obtain compounds 65-92.

10 While compounds 85-92 are 1,2-biphosphates, Figure 5B illustrates the synthesis of 1,3-biphosphates. Commercially available 2-phenoxy-1,3-propane-diol was used as the starting material. The starting compound was first protected with t-BuOK in the presence of methyl iodide, followed by catalytic hydrogenation to give an intermediate which was then reacted with a halide (RX, where R is as defined above for R'). The recovered intermediate was subsequently treated with AlCl<sub>3</sub> in the presence of ethyl-SH to yield a 1,3 diol possessing the RO group bound to C2 of the backbone. The recovered 1,3 diol was dissolved in freshly distilled 1:1 THF/CH<sub>2</sub>Cl<sub>2</sub>, followed by the addition of 1H-terazole. To this stirred mixture was added

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which afforded a di-ether intermediate having a hydroxy group at the C2 position. The di-ether was mixed with 1H-tetrazole and to this stirred mixture was added 5 di-tert-bisopropyl phosphoramide. The reaction mixture was monitored via TLC, and at the appropriate time the phosphonate was oxidized to the phosphate *in situ* with peracetic acid. The reaction mixture was purified with column chromatography to afford benzyl-protected phosphates. The removal of the protecting benzyl groups was carried out in ethanol by subjecting the benzyl-protected phosphates to catalytic reduction using 10% palladium on activated carbon (Pd/C) under H<sub>2</sub> atmosphere at 60 psi to yield monophosphate compounds.

In Figure 16, a similar reaction scheme was employed, except instead of reacting the glycidal bromide with an alcohol (ROH), BuOH was used to protect the C3 site. The reaction conditions included treatment with K<sub>2</sub>CO<sub>3</sub> followed by treatment with the ammonium salt C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>N<sup>+</sup>(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>Cl<sup>-</sup>, resulting in displacement of the bromide with the Bu group. The ring of the glycidal intermediate was then opened following treatment with 1M HCl in ether and anhydrous BuOH, which protected the C1 site. The resulting di-ether intermediate has a hydroxy group at the C2 position. The di-ether was mixed with a halide salt (RCl) in aqueous K<sub>2</sub>CO<sub>3</sub>, yielding a protected intermediate having an R group attached via ether bond at the C2 position. This intermediate was de-protected via catalytic reduction using 10% palladium on activated carbon (Pd/C) under H<sub>2</sub> atmosphere at 60 psi to yield a 1,3 diol. The diol was combined with 1H-tetrazole and to this stirred mixture was added 20 di-tert-bisopropyl phosphoramide. The reaction mixture was monitored via TLC, and at the appropriate time the phosphonate was oxidized to the phosphate *in situ* with peracetic acid. The reaction mixture was purified with column chromatography to afford benzyl-protected phosphates. The removal of the protecting benzyl groups was carried out in ethanol by subjecting the benzyl-protected phosphates to catalytic reduction using 10% palladium on activated carbon (Pd/C) under H<sub>2</sub> atmosphere at 60 psi to yield 1,3 bisphosphates.

Using the di-ether intermediate prepared as shown in Figure 6A (e.g., bearing R and R<sup>1</sup> substituents), a number of modified phosphates and phosphonates can be attached at the C1 site upon removal of the tosyl group. As shown in Figure 7A, the intermediate is reacted under basic conditions with X<sup>1</sup>-Z<sup>1</sup>-PO(O-protecting group)<sub>2</sub> where Z<sup>1</sup> is -(R<sup>3</sup>)CH- and X<sup>1</sup> is H. The basic conditions remove the tosylate protecting group and allow the modified phosphate -Z<sup>1</sup>-PO(O-protecting group)<sub>2</sub> to form a single bond to the C1 site. The protecting groups are removed following treatment with TMSBr, affording a -(R<sup>3</sup>)CH-PO(OH)<sub>2</sub> group at the C1 site. As shown in Figure 7B, the intermediate is reacted under basic

conditions using tri(tert-butylammonium) with X<sup>1</sup>-Z<sup>1</sup>-PO(OH)-Z<sup>2</sup>-PO(OH)<sub>2</sub> where Z<sup>1</sup> is -O-, Z<sup>2</sup> is -CH<sub>2</sub>- and X<sup>1</sup> is H. The basic conditions remove the tosylate protecting group and allow the modified phosphonate -Z<sup>1</sup>-PO(OH)-Z<sup>2</sup>-PO(OH)<sub>2</sub> to form a single bond to the C1 site. Upon treatment with acidic conditions and CH<sub>2</sub>CN, the -O-PO(OH)-CH<sub>2</sub>-PO(OH)<sub>2</sub> group is installed at the C1 site. As shown in Figure 7C, the intermediate is reacted under basic conditions with X<sup>1</sup>-Z<sup>1</sup>-PO(O-protecting group) where Z<sup>1</sup> is -OCH<sub>2</sub>CH<sub>2</sub>- and X<sup>1</sup> is H. The basic conditions remove the tosylate protecting group and allow the modified phosphate -Z<sup>1</sup>-PO(O-protecting group)<sub>2</sub> to form a single bond to the C1 site. The protecting groups are removed following treatment with TMSBr in collidine and water wash, affording a -OCH<sub>2</sub>CH<sub>2</sub>-PO(OH)<sub>2</sub> group at the C1 site.

To prepare the conformationally restricted cyclo-phosphane compound of formula (III), compounds 26-30 were used as starting materials in the synthesis scheme illustrated in Figure 11. Compounds 26-30 were reacted with 1H-tetrazole and the resulting product was treated with di-tert-butyl diisopropylphosphoramide, causing an intramolecular cyclization. *In situ* oxidation of the phosphonate with peracetic acid yielded a cyclic phosphate intermediate. Reduction with TFA yielded the compounds of formula (III).

Other conformationally restricted compounds can also be prepared. As shown in Figure 12, an alternative scheme is shown for preparing cyclic phosphates where X<sup>1</sup> and X<sup>2</sup> together are -O-PO(OH)-O-. A benzyl-protected 1,3 diol intermediate is reacted with POC<sub>6</sub>, which results in an intramolecular cyclization. Treatment with 10% palladium on activated carbon (Pd/C) under H<sub>2</sub> atmosphere (as described above) affords a cyclic phosphate bearing a hydroxyl group bound to the C2 carbon. The cyclic intermediate is then treated with an excess of R-triflate and 2,6-di-tert-butyl-4-methylpyridine to afford the final compound.

As shown in Figure 13, a scheme is shown for preparing a cyclic phosphate where X<sup>1</sup> and X<sup>2</sup> together are -O-PO(OH)-NH-. Using the intermediates 35-43 prepared above as starting material, they are treated with tri(2,4-triazole)phosphate followed by 2% HCl wash, resulting in intramolecular cyclization.

As shown in Figure 14, a scheme is shown for preparing a cyclic compound where the phosphate group is not a part of the ring, specifically, X<sup>2</sup> and X<sup>3</sup> together are -N(H)-C(O)-N(R<sup>1</sup>)-. Using the intermediates 50-54 prepared above as starting materials, they are treated with anhydrous COCl<sub>2</sub>, which inserts a carbonyl between the amines bound to the C2 and C3 carbons during cyclization.





intralesionally, or by application to mucous membranes, such as, that of the nose, throat, and bronchial tubes.

Compositions within the scope of this invention include all compositions wherein the compound of the present invention is contained in an amount effective to achieve its intended purpose. While individual needs vary, determination of optimal ranges of effective amounts of each component is within the skill of the art. Typical dosages comprise about 0.01 to about 100 mg/kg body wt. The preferred dosages comprise about 0.1 to about 100 mg/kg body wt. The most preferred dosages comprise about 1 to about 100 mg/kg body wt. Treatment regimen for the administration of the compounds of the present invention can also be determined readily by those with ordinary skill in art.

Certain compounds of the present invention have been found to be useful as agonists of LPA receptors while other compounds of the present invention have been found useful as antagonists of LPA receptors. Due to their differences in activity, the various compounds find different uses. The preferred animal subject of the present invention is a mammal, i.e., an individual belonging to the class Mammalia. The invention is particularly useful in the treatment of human subjects.

One aspect of the present invention relates to a method of modulating LPA receptor activity which includes providing a compound of the present invention which has activity as either an LPA receptor agonist or an LPA receptor antagonist and contacting an LPA receptor with the compound under conditions effective to modulate the activity of the LPA receptor.

The LPA receptor is present on a cell which either normally expresses the LPA receptor or has otherwise been transformed to express a particular LPA receptor. Suitable LPA receptors include, without limitation, EDG-2, EDG-4, EDG-7, and PSP-24 receptors. The tissues which contain cells that normally express these receptors are indicated in Table 1 above. When contacting a cell with the LPA receptor agonist or LPA receptor antagonist of the present invention, the contacting

can be carried out while the cell resides *in vitro* or *in vivo*. To heterologously express these receptors in host cells which do not normally express them, a nucleic acid molecule encoding one or more of such receptors can be inserted in sense orientation into an expression vector which includes appropriate transcription and translation regulatory regions (i.e., promoter and transcription termination signals) and then host cells can be transformed with the expression vector. The expression vector may integrate in the cellular genome or simply be present as extrachromosomal nuclear material. Expression can be either

constitutive or inductive; although constitutive expression is suitable for most microcosms.

The nucleotide and amino acid sequences for EDG-2 is known and reported in An et al. (1997) and Genbank Accession No. U80811, which is hereby incorporated by reference. An EDG-2 encoding nucleic acid molecule has a nucleotide sequence according to SEQ ID No. 1 as follows:

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The encoded EDG-2 receptor has an amino acid sequence according to SEQ. ID. No. 2 as follows:

The nucleotide and amino acid sequences for EDG-4 is known and reported in An et al. (1998) and Genbank Accession No. NM\_004720, which is hereby incorporated by reference. An EDG-4 encoding nucleic acid molecule has a nucleotide sequence according to SEQ ID No. 3 as follows:

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The encoded EUG-4 receptor has an amino acid sequence according to SEQ. ID. No. 4 as follows:

The nucleotide and amino acid sequences for EDG-7 is known and reported in Bandoh et al. (1999) and Genbank Accession No. NM\_02152, which is hereby incorporated by reference. An EDG-7 encoding nucleic acid molecule has a nucleotide sequence according to SEQ. ID. No. 5 as follows:

The encoded EDG-7 receptor has an amino acid sequence according to SEQ. ID. No. 6 as follows:

MURCHISON DIFFERENTIERT TUDOMOTEL VIVUGVTFY CLEPFSNL VIVAVENRK  
FIFTYFLA NLAADTAG IAYFMENT GIVSKTLTN RWTIARQGLD SSITASLNL 6 12

The nucleotide and amino acid sequences for PSP-24 is known and reported in Kawasawa et al. (2000) and Genbank Accession No. AB030566, which is hereby incorporated by reference. A PSP-24 encoding nucleic acid molecule has a nucleotide sequence according to SEQ. ID. No. 7 as follows:

The encoded PEP-24 receptor has an amino acid sequence according to SEQ. ID. No. 8 as follows:

LPA receptor agonists will characteristically induce LPA-like activity from an LPA receptor, which can be measured either chemically, e.g.,  $\text{Ca}^{2+}$  or  $\text{Cl}^-$  current in oocytes, or by examining changes in cell morphology, mobility, proliferation, etc. In contrast, LPA receptor antagonists will characteristically block LPA-like activity from an LPA receptor. This too can be measured either chemically, e.g.,  $\text{Ca}^{2+}$  or  $\text{Cl}^-$  current in oocytes, or by examining changes in cell morphology, mobility, proliferation, etc.

By virtue of the compounds of the present invention acting as LPA receptor antagonists, the present invention also relates to a method of inhibiting LPA-induced activity on an LPA receptor. This method includes providing a compound of the present invention which has activity as an LPA receptor antagonist and contacting an LPA receptor with the compound under conditions effective to inhibit LPA-induced activity of the LPA receptor. The LPA receptor can be as defined above. The LPA receptor is present on a cell which normally expresses the receptor or which heterologously expresses the receptor. The contacting of the LPA receptor with the compound of the present invention can be performed either *in vitro* or *in vivo*.

As noted above, LPA is a signaling molecule involved in a number of different cellular pathways which involve signaling through LPA receptors, including those LPA receptors described above. Therefore, it is expected that the compounds of the present invention will modulate the effects of LPA on cellular behavior, either by acting as LPA receptor antagonists or LPA receptor agonists.

One aspect of the present invention relates to a method of treating cancer which includes providing a compound of the present invention and administering an effective amount of the compound to a patient in a manner effective to treat cancer. The types of cancer which can be treated with the compounds of the present invention includes those cancers characterized by cancer cells whose behavior is attributable at least in part to LPA-mediated activity. Typically, these types of cancer are characterized by cancer cells which express one or more types of LPA receptors. Exemplary forms of cancer include, without limitation, prostate cancer and ovarian cancer.

The compounds of the present invention which are particularly useful for cancer treatment are the LPA receptor antagonists. When administering the compounds of the present invention, they can be administered systemically or, alternatively, they can be administered directly to a specific site where cancer cells are present. Thus, administering can be accomplished in any manner effective for delivering the compound to cancer cells. Without being bound by theory, it is believed that the LPA receptor antagonists, upon binding to LPA receptors, will inhibit proliferation or metastasis of the cancer cells or otherwise destroy those cancer cells. As shown in Example 12, *Infa*, several LPA antagonist compounds of the present invention were cytotoxic to prostate cancer cell lines which express one or more LPA receptors of the type described above.

When the LPA antagonist compounds or pharmaceutical compositions

therapeutic agents or treatment regimen presently known or hereafter developed for the treatment of various types of cancer.

Cancer invasion is a complex multistep process in which individual cells or cell clusters detach from the primary tumor and reach the systemic circulation or the lymphatics to spread to different organs (Liotta et al., 1987). During this process, tumor cells must arrest in capillaries, extravasate, and migrate into the stroma of the tissue to make secondary foci. First, tumor cells must recognize signals on the endothelial cell that arrest them from the circulation. Second, tumor cells must attach to the basement membrane glycoprotein laminin via the cell surface laminin receptors. Following attachment to the basement membrane, tumor cells secrete proteases to degrade the basement membrane. Following attachment and local proteolysis, the third step of invasion is tumor cell migration. Cell motility plays a central role in tumor cell invasion and metastasis. The relationship between motility of tumor cells *in vitro* and the metastatic behavior in animal experiments indicates a strong direct correlation (Hoffman-Wellenhof et al., 1995). It is a well-documented fact that PLGF<sub>9</sub> promote proliferation and increase invasiveness of cancer cell *in vitro*. Imanura and colleagues established that cancer cells require serum factors for their invasion (Imanura et al., 1991), and later identified LPA as the most important serum (Imanura et al., 1995), and later identified LPA as the most important serum component that is fully capable of restoring tumor cell invasion in serum-free systems (Ku et al., 1995; Imanura et al., 1993; Mukai et al., 1993).

It has been shown that PLGFR are expressed in ovarian cancer cell lines, namely, OVC1 and HEY cells. Specifically, RT-PCR analyses show the presence of EDG-2 and EDG-7 receptors in these cell lines. Recently, Im et al. (2000) demonstrated that EDG-7 is expressed in prostate cancer cell lines; namely, PC-3 and LNCaP cells. RT-PCR analysis on the prostate cancer cell lines DU-145, PC-3, and LNCaP lines showed that EDG-2, 4, 5, and EDG-7 are present in all three prostate cancer cell lines, whereas EDG-3 is present in LNCaP and DU-145 prostate cancer cell lines.

As shown in the Examples, several LPA receptor antagonists of the present invention are capable of targeting specific prostate cancer cell lines and specific ovarian cancer cell lines. Thus, the LPA antagonists of the present invention provide an alternative approach for treatment of LPA-mediated cancers, including prostate cancer and ovarian cancer.

Another aspect of the present invention relates to a method of

enhancing cell proliferation. This method of enhancing cell proliferation includes the

steps of providing a compound of the present invention which has activity as an

agonist of an LPA receptor and contacting the LPA receptor on a cell with the

compound in a manner effective to enhance LPA receptor-induced proliferation of the cell.

In addition to the roles that LPA plays in modulating cancer cell activity, there is strong evidence to suggest that LPA also has a physiological role in natural wound healing. At wound sites, LPA derived from activated platelets is believed to be responsible, at least in part, for stimulating cell proliferation at the site of injury and inflammation possibly in synchronization with other platelet-derived factors (Balazs et al., 2000). Moreover, LPA by itself stimulates platelet aggregation, which may in turn be the factor that initiates an element of positive feedback to the initial aggregatory response (Schumacher et al., 1979; Tokumura et al., 1981; Gerard et al., 1979; Simon et al., 1982).

Due to the role of LPA in cell proliferation, compounds having LPA receptor agonist activity can be used in a manner effective to promote wound healing. Accordingly, another aspect of the present invention relates to a method of treating a wound. This method is carried out by providing a compound of the present invention which has activity as an agonist of an LPA receptor and delivering an effective amount of the compound to a wound site, where the compound binds to LPA receptors on cells that promote healing of the wound, thereby stimulating LPA receptor agonist-induced cell proliferation to promote wound healing.

The primary goal in the treatment of wounds is to achieve wound closure. Open cutaneous wounds represent one major category of wounds and include burn wounds, neuropathic ulcers, pressure sores, venous stasis ulcers, and diabetic ulcers. Open cutaneous wounds routinely heal by a process which comprises six major components: i) inflammation, ii) fibroblast proliferation, iii) blood vessel proliferation, iv) connective tissue synthesis, v) epithelialization, and vi) wound contraction. Wound healing is impaired when these components, either individually or as a whole, do not function properly. Numerous factors can affect wound healing, including malnutrition, infection, pharmacological agents (e.g., actinomycin and steroids), diabetes, and advanced age (see Flum and Goodson, 1988).

Phospholipids have been demonstrated to be important regulators of cell activity, including mitogenesis (Xu et al., 1995b), apoptosis, cell adhesion, and regulation of gene expression. Specifically, for example, LPA elicits growth factor-like effects on cell proliferation (Moolenaar, 1995) and cell migration (Imamura et al., 1993). It has also been suggested that LPA plays a role in wound healing and regeneration (Tigyi and Miledi, 1992).

In general, agents which promote a more rapid influx of fibroblasts, endothelial and epithelial cells into wounds should increase the rate at which wounds

heal. Compounds of the present invention that are useful in treating wound healing can be identified and tested in a number of *in vitro* and *in vivo* models.

*In vitro* systems model different components of the wound healing process, for example the return of cells to a "wounded" confluent monolayer of tissue culture cells, such as fibroblasts (Venter et al., 1986), endothelial cells (Miyata et al., 1990) or epithelial cells (Kurtha et al., 1992). Other systems permit the measurement of endothelial cell migration and/or proliferation (Muller et al., 1987; Sato et al., 1988).

*In vivo* models for wound healing are also well-known in the art, including wounded pig epidermis (Ohkawa et al., 1977) or drug-induced oral mucosal lesions in the hamster cheek pouch (Chemick et al., 1974).

The compounds of the present invention which are effective in wound healing can also be administered in combination, i.e., in the pharmaceutical composition of the present invention or simultaneously administered *in different* routes, with a medicament selected from the group consisting of an antibacterial agent, an antiviral agent, an antifungal agent, an antiparasitic agent, an antiinflammatory agent, an analgesic agent, an antipuritic agent, or a combination thereof.

For wound healing, a preferred mode of administration is by the topical route. However, alternatively, or concurrently, the agent may be administered by parenteral, subcutaneous, intravenous, intramuscular, intraperitoneal or transdermal routes. Alternatively, or concurrently, administration may be by the oral route. The dosage administered will be dependent upon the age, health, and weight of the recipient, kind of concurrent treatment, if any, frequency of treatment, and the nature of the effect desired.

For the preferred topical applications, especially for treatment of humans and animals having a wound, it is preferred to administer an effective amount of a compound according to the present invention to the wounded area, e.g., skin surfaces. This amount will generally range from about 0.001 mg to about 1 g per application, depending upon the area to be treated, the severity of the symptoms, and the nature of the topical vehicle employed. A preferred topical preparation is an ointment wherein about 0.01 to about 50 mg of active ingredient is used per ml of ointment wherein, such as PFG-1000.

## EXAMPLES

The following examples are intended to illustrate, but by no means are intended to limit, the scope of the present invention as set forth in the appended claims.

## Materials and Methods

A Thomas-Hoover capillary melting point (mp) apparatus was used to measure all melting points (mps), which were uncorrected.

<sup>1</sup>H and <sup>13</sup>C nuclear magnetic resonance (NMR) spectra were recorded on a Bruker AX 300 spectrometer (300, 75.5 MHz). Chemical shift values ( $\delta$ ) are expressed as parts per million (ppm) relative to tetramethylsilane (TMS). Peaks are abbreviated as follows: s – singlet; d – doublet; t – triplet; q – quartet; bs – broad singlet; m – multiplet.

Proton, carbon-13, and phosphorous-31 magnetic resonance spectra were obtained on a Bruker AX 300 spectrometer. Chemical shifts for proton and carbon-13 are reported as parts per million ( ) relative to tetramethylsilane (TMS). Spectra for phosphorous-31 are reported as parts per million ( ) relative to 0.0435 M triphenylphosphate in acetone-d<sub>6</sub> at = 0 ppm.

Infrared (IR) spectra were recorded on Perkin Elmer System 200-FITR. Mass spectra (MS) were recorded on either a Bruker Esquire AG or a Bruker Esquire LC/MS spectrometer by direct infusion utilizing the Electrospray Interface (ESI) either in the positive or negative mode. Spectral data were consistent with assigned structures.

Elemental analysis was performed by Atlantic Microlabs, Inc. (Norcross, GA), and values found are within  $\pm 0.4\%$  of the theoretical values. Silica gel (Merck, 230-400 mesh or 200-425 mesh, 60 Å) was used for flash column chromatography.

Analytical TLC was performed on Sigma-Aldrich silica gel 60 F 254 TLC sheets with aluminum backings (thickness 200 or 250 microns).

All reagents, solvents, and chromatography media, unless otherwise noted, were purchased from either Aldrich Chemical Company (Milwaukee, WI), Fisher Scientific (Pittsburgh, PA), or Sigma Chemical Co. (St. Louis, MO) without further purification. Tetrahydrofuran (THF) was dried by distillation from sodium metal with benzophenone as an indicator. Anhydrous methylene chloride (CH<sub>2</sub>Cl<sub>2</sub>) was distilled from calcium hydride (CaH<sub>2</sub>). All the mono glycerides were from Nu-Check-<sup>+</sup>-Prep (Minneapolis, MN).  $\alpha$ -Boc-L-serine was purchased from Fluka.

## EXAMPLES

All lipids were purchased from Avanti Polar Lipids (Alabaster, AL). Fatty acid-free bovine serum albumin (BSA). Prior to use, LPA was complexed, at a 1:1 ratio molar ratio, with 1 mM BSA dissolved in Ca<sup>2+</sup>-free Hanks' balanced salt solution containing 1 mM EGTA. Aliquots of all the other lipids were dissolved in MeOH and mixed with LPA prior to application, or as otherwise indicated.

Cytofectene transfection reagent was from Bio-Rad (Hercules, CA). Fura-2 AM was from Molecular Probes (Eugene, OR).

Culture media, fetal bovine serum (FBS), and G418 were obtained from Cellgro (Herndon, VA).

RH7777 cells, stably expressing human Edg-4, were kindly provided by Dr. Kevin Lynch (University of Virginia, Charlottesville, VA). Flag-tagged cDNA's encoding human Edg-4 and -7 inserted into the pCDNA3 expression plasmid (Invitrogen, Carlsbad, CA), were a generous gift from Dr. Junken Aida (University of Tokyo, Tokyo, Japan). RH7777 and NIH3T3 cells were obtained from the American Type Culture Collection (Manassas, VA). HEK cells were provided by Dr. Lisa Jennings (University of Tennessee, Memphis). All cell lines were maintained in Dulbecco's Modified Eagle's Medium (DMEM) containing 10% FBS and 2 mM glutamine. Oocytes were obtained from adult *Xenopus laevis* frogs as previously described (Tigyi et al., 1999).

**Stable transfection**  
RH7777 cells were transfected with the cDNA constructs encoding human Edg-2, Edg-4, or Edg-7 and then were subcloned into the pCDNA3 expression vector using the Cytofectene transfection reagent according to the manufacturers' protocol. Transfected cells were selected in DMEM containing 10% FBS and 1 mg/ml geneticin. Resistant cells were collected and subcloned by limiting dilution. The resulting clones were then screened using functional assays and RT-PCR analysis. Data are representative of three individual clones.

**Transient transfection**  
RH7777 cells were plated on polylysine-coated glass coverslips (Becton, Vineland, NJ) one day prior to transfection. The following day, cells were transfected overnight (16-18 hr) with 1  $\mu$ g of plasmid DNA mixed with 6  $\mu$ l of Cytofectene. The cells were then rinsed twice with DMEM and cultured in DMEM containing 10% FBS. The next day, the cells were rinsed with DMEM and serum was withdrawn for a minimum of 2 hr prior to monitoring intracellular Ca<sup>2+</sup>.

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**Measurement of intracellular  $\text{Ca}^{2+}$  and data analysis**

Changes in intracellular  $\text{Ca}^{2+}$  were monitored using the fluorescent  $\text{Ca}^{2+}$  indicator Fluo-2 AM as previously described (Tigyi et al., 1999). Data points from the intracellular  $\text{Ca}^{2+}$  measurements represent the total peak area of the  $\text{Ca}^{2+}$  transients elicited, as determined by the FLWinLab software (Perkin-Elmer, Wellesley, MA).

Data points represent the average of at least 3 measurements  $\pm$  standard deviation.

The significance of the data points was determined using the Student's t-test and values were considered significant at  $p < 0.05$ .

**Electrophysiological recording in *Xenopus* oocytes**

Oscillatory  $\text{Cl}^-$  currents, elicited by LPA, were recorded using a two-electrode voltage clamp system as previously described (Tigyi et al., 1999).

**RT-PCR analysis of Edg and PSP24 mRNA**

15 The identification of Edg and PSP24 receptor mRNA by RT-PCR was performed as previously described (Tigyi et al., 1999), using the following oligonucleotide sequences:

**EDG-1**

forward primer 5'-uTCATCGTCGGCATACAACTA-3' (SEQ. ID No. 9);

reverse primer 5'-GAGTGAQCTGTGTAQGGGGTg-3' (SEQ. ID No. 10);

**EDG-2**  
forward primer 5'-uAGATCTGTGACCAGCCGACTCAC-3' (SEQ. ID No. 11);

reverse primer 5'-GTGGCCATCAAGTAATAATAg-3' (SEQ. ID No. 12);

**EDG-3**  
forward primer 5'-u<sub>13</sub>CTGTGTCATCTGCAQCTCATC-3' (SEQ. ID No. 13);

reverse primer 5'-TGCTGTGTCAGAGGCAATGTA<sub>397</sub>-3' (SEQ. ID No. 14);

**EDG-4**  
forward primer 5'-u<sub>3</sub>CTGCTCAGCCGCTCTATTG-3' (SEQ. ID No. 15);

reverse primer 5'-AGAGCCACCCACAGTCATCAGGAg-3' (SEQ. ID No. 16);

**EDG-5**  
forward primer 5'-uATGGGCAQCTGTACTCGGAG-3' (SEQ. ID No. 17);

reverse primer 5'-uAGCCAGCAGACGATAAAGAC<sub>70</sub>-3' (SEQ. ID No. 18);

**EDG-6**  
forward primer 5'-u<sub>30</sub>TOAACATCACCTGAGTGACCT-3' (SEQ. ID No. 19);

reverse primer 5'-GATCATCAGCACCGTCTTCAGC<sub>39</sub>-3' (SEQ. ID No. 20);

**EDG-7**  
forward primer 5'-uAGCAACACTGATACTGCGATG-3' (SEQ. ID No. 21);

reverse primer 5'-GCATCTCATGATGACATGTG<sub>46</sub>-3' (SEQ. ID No. 22);

**EDG-8**

forward primer 5'-uATCTGTGGCTCTATGCAAGQA-3' (SEQ. ID No. 23);

reverse primer 5'-GGTGTAGATGATAGGATTAGCA<sub>161</sub>-3' (SEQ. ID No. 24);

**PSP24**

forward primer 5'-u<sub>10</sub>CTGCATCATGTTGACAGAG-3' (SEQ. ID No. 25); and

reverse primer 5'-ACGAACCTATOCAGGCCTCC<sub>114</sub>-3' (SEQ. ID No. 26).

**Cell proliferation assay**

10 Proliferation of NIH3T3 cells was assessed by direct cell counting as previously described (Tigyi et al., 1999). NIH3T3 cells were plated in 24-well plates at a density of 10,000 cells/well, in DMEM containing 10% FBS. The following day, the cells were rinsed and serum starved in DMEM for 6 hr. Lipids were then added for 24 hr. Cell numbers were determined by counting in a Coulter counter (Coulter

**Electronics, Hialeah, FL).****Incorporation of  $^3\text{H}$ -thymidine**

15 The incorporation of  $^3\text{H}$ -thymidine into RHT777 cells was determined as previously described (Tigyi et al., 1994).

**Example 1. Synthesis of N-(*tert*-butyloxycarbonyl)-L-serine  $\beta$ -lactone, Intermediate Compound 25**

20 A 500 ml three-neck flask was equipped with a low temperature thermometer and a 100 ml dropping funnel. All glassware were flame-dried and cooled to room temperature under Argon ( $\text{Ar}$ ) before use. To the flask were added triphenylphosphine ( $\text{PPh}_3$ ) (10 g, 38 mmol, dried over  $\text{P}_2\text{O}_5$  under vacuum for 72 hr) and freshly distilled THF (190 ml). The solution was cooled and stirred at  $-78^\circ\text{C}$  (dry ice-acetone bath) under Argon. With vigorous stirring, freshly distilled diethyl azodicarboxylate (DEAD) (6.2 ml, 39.9 mmol) was added with a syringe over a period of 30 min. After the addition was complete, the mixture was stirred until a milky white paste was obtained (ca. 30-40 min). A solution of N-(*tert*-butyloxycarbonyl)-L-serine (24) (7.79 g, 38 mmol, dried over  $\text{P}_2\text{O}_5$  under vacuum for 72 hr) in freshly distilled THF (75 ml) was added dropwise over a period of 45 min to the reaction mixture. The mixture was stirred overnight at  $-78^\circ\text{C}$  under argon and allowed to warm to  $0^\circ\text{C}$ . The flask was placed in an ice bath when the temperature reached  $-10^\circ\text{C}$ . After 30 min (ca. the ice bath was replaced with a water bath, and the reaction mixture was stirred for 2 hrs and concentrated on the rotary evaporator to pale yellow oil at  $30^\circ\text{C}$ . The oil was then treated with 25%  $\text{EtOAc}/\text{hexanes}$  (100 ml), the resulting white solid was removed by filtration and washed with 25%  $\text{EtOAc}/\text{hexanes}$  ( $2 \times 70$

ml), the combined filtrate was concentrated, and the residual oil subjected to flash chromatography on silica gel with 25% (500 ml) and 30% (1500 ml)  $\text{EtOAc}/\text{hexanes}$ , successively.

Appropriate fractions were combined to afford 1.4 g (47%) of 25 as a white solid: mp 119-121  $^{\circ}\text{C}$  dec (Lit. 119.5-120.5  $^{\circ}\text{C}$  dec);  $^1\text{H}$  NMR ( $\text{CD}_2\text{Cl}_2$ )  $\delta$  1.44 (s, 9H), 4.38-4.42 (m, 2H), 4.96-5.03 (q,  $J$ =6.1 Hz, 1H), 5.39 (s, br, 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_2\text{Cl}_2$ )  $\delta$  28.31, 60.01, 66.63, 81.50, 155.01, 169.94; IR (KBr) 3361, 2978, 1843, 1680, 1533, 1370, 1292  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{12}\text{H}_{15}\text{NO}_4$ : C, 51.33; H, 6.94; N, 7.50. Found: C, 51.41; H, 7.01; N, 7.51.

**Example 2 - Synthesis of Compounds 26-34**

The glassware used were flame-dried and cooled to room temperature under argon atmosphere. The reaction was carried out in argon atmosphere. THF was freshly distilled prior to use.

15 **Compound 26: *tert*-Butyl N-[1-(hydroxymethyl)-2-(nonylamino)-2-oxoethyl]carbamate**

To a solution of decyl amine (490 mg, 3.20 mmol) in THF (60 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (300 mg, 1.60 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 290 mg (52%) of 26 as a white waxy powder: mp 50-52  $^{\circ}\text{C}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.88 (t,  $J$ =6.4 Hz, 3H), 1.26 (s, 14H), 1.46 (s, 9H), 3.04 (bs, 1H), 3.16-3.34 (m, 2H), 3.63 (m, 1H), 4.06-4.15 (m, 2H), 5.53 (bs, 1H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  1409, 22.65, 26.80, 28.27, 29.24, 29.27, 29.37, 29.50, 29.51, 31.86, 39.43, 54.34, 62.87, 77.20, 86.34, 171.52; IR (KBr) 3282, 3098, 2929, 2836, 1666, 1547, 1467, 1369, 1300, 1248, 1179  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{22}\text{H}_{32}\text{N}_2\text{O}_4$ : C, 62.76; H, 10.53; N, 8.13. Found: C, 63.00; H, 10.46; N, 7.98.

**Compound 27: *tert*-Butyl N-[1-(hydroxymethyl)ethyl]carbamate**

To a solution of tetradecyl amine (273 mg, 1.28 mmol) in THF (40 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (200 mg, 1.06 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

16 **Compound 28: *tert*-Butyl N-[1-(hydroxymethyl)-2-(octadecylamino)-2-oxoethyl]carbamate**

To a solution of octadecyl amine (516 mg, 2.08 mmol) in THF (60 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (300 mg, 1.60 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 245 mg (57%) of 27 as a white powder: mp 59-62  $^{\circ}\text{C}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.88 (t,  $J$ =6.3 Hz, 3H), 1.25 (s, 9H), 1.45 (s, 9H), 3.15-3.36 (m, 3H), 3.63-3.65 (m, 1H), 4.07-4.13 (m, 2H), 5.60-5.63 (m, 1H), 6.72 (bs, 1H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  14.10, 22.66, 26.81, 27.98, 28.27, 29.25, 29.33, 29.37, 29.50, 29.57, 29.62, 29.66, 31.90, 39.47, 54.58, 62.87, 77.20, 80.52, 156.34, 171.37; IR (KBr) 3345, 2920, 2852, 1708, 1688, 1655, 1637, 1572, 1529, 1472, 1248, 1173  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{22}\text{H}_{44}\text{N}_2\text{O}_4$ : C, 65.96; H, 11.07; N, 6.99. Found: C, 66.04; H, 11.17; N, 6.96.

17 **Compound 29: *tert*-Butyl N-[1-(hydroxymethyl)-2-(tetradecylamino)-2-oxoethyl]carbamate**

18 To a solution of octadecyl amine (516 mg, 2.08 mmol) in THF (60 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (300 mg, 1.60 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

19 **Compound 30: *tert*-Butyl N-[1-(hydroxymethyl)-2-(hexadecylamino)-2-oxoethyl]carbamate**

20 To a solution of octadecyl amine (516 mg, 2.08 mmol) in THF (60 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (300 mg, 1.60 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

21 **Compound 31: *tert*-Butyl N-[1-(hydroxymethyl)-2-(octadecylamino)-2-oxoethyl]carbamate**

22 To a solution of octadecyl amine (516 mg, 2.08 mmol) in THF (60 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (300 mg, 1.60 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

23 **Compound 32: *tert*-Butyl N-[1-(hydroxymethyl)-2-(hexadecylamino)-2-oxoethyl]carbamate**

24 To a solution of octadecyl amine (516 mg, 2.08 mmol) in THF (60 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (300 mg, 1.60 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

25 **Compound 33: *tert*-Butyl N-[1-(hydroxymethyl)-2-(hexadecylamino)-2-oxoethyl]carbamate**

26 To a solution of octadecyl amine (516 mg, 2.08 mmol) in THF (60 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (300 mg, 1.60 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

27 **Compound 34: *tert*-Butyl N-[1-(hydroxymethyl)-2-(hexadecylamino)-2-oxoethyl]carbamate**

28 To a solution of octadecyl amine (516 mg, 2.08 mmol) in THF (60 ml),  $\text{N}-(\text{tert}-\text{butoxycarbonyl})-\text{L}\text{-serine } \beta\text{-lactone}$  (300 mg, 1.60 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

62, 53, 68-30, 77-20, 111-17, 114-81, 121-70, 130-25, 156-22, 169-78; IR (KBr) 3304, 2920, 2852, 1658, 1514, 1472, 1238, 1174 cm<sup>-1</sup>; Anal. Calcd. for C<sub>24</sub>H<sub>40</sub>N<sub>2</sub>O<sub>3</sub>: 0.05CHCl<sub>3</sub>; C, 67.56; H, 9.71; N, 5.62. Found: C, 67.80; H, 9.67; N, 5.60.

**5** **Compound 30:** *tert*-Butyl N-[1-(hydroxymethyl)-2-(4-methoxyanilino)-2-oxethyl]carbamate

To a solution of *o*-anisidine (100 mg, 0.8 mmol) in THF (20 ml) N-(*tert*-butoxycarbonyl)-L-serine  $\beta$ -lactone (51 mg, 0.8 mmol), was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and were crystallized from

CHCl<sub>3</sub>/hexanes to afford 135 mg (34%) of 30 as a white powder: mp 109-111°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.48 (s, 9H), 3.68-3.73 (m, 1H), 3.80 (s, 3H), 4.24-4.27 (m, 2H), 5.68 (bs, 1H), 6.83-6.88 (m, J<sub>0</sub>=9 Hz, 2H), 7.37-7.42 (m, J<sub>0</sub>=9 Hz, 2H), 8.61 (bs, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  28.29, 54.96, 55.47, 62.54, 81.00, 114.18, 121.78, 130.45, 156.64, 156.98, 169.59; IR (KBr) 3340, 2978, 1673, 1603, 1516, 1298, 1238, cm<sup>-1</sup>; Anal. Calcd. for C<sub>13</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub>: C, 58.05; H, 7.15; N, 9.03. Found: C, 58.04; H, 7.17; N, 9.06.

**20** **Compound 31:** *tert*-Butyl N-[1-(hydroxymethyl)-2-oxo-2-[3-(tetradecyloxy)anilino]ethyl] carbamate

To a solution of 3-(tetradecyloxy)aniline (179 mg, 0.588 mmol) in THF (25 ml), N-(*tert*-butoxycarbonyl)-L-serine  $\beta$ -lactone (91 mg, 0.490 mmol) was added, and the mixture was refluxed for 48 hrs under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in

vacuo to afford 105 mg (43%) of 31 as a white powder: mp 70-72 °C; <sup>1</sup>H NMR

(CDCl<sub>3</sub>)  $\delta$  0.88 (t, J=6.5 Hz, 3H), 1.26 (s, 2H), 1.48 (s, 9H), 1.76 (m, 2H), 3.67-3.73 (dd, J<sub>1</sub>=5.1 Hz, J<sub>2</sub>=6.9 Hz, 1H), 3.93 (t, J=5.5 Hz, 2H), 4.23-4.26 (m, 2H), 5.66 (bs, 1H), 6.64-6.68 (m, 1H), 6.9-7.0 (m, 1H), 7.19 (t, J<sub>0</sub>=8.1 Hz, 1H), 7.23 (t, J<sub>1</sub>=2 Hz, 1H), 6.93-6.96 (m, 1H), 7.19 (t, J<sub>0</sub>=8.1 Hz, 1H), 7.23 (t, J<sub>1</sub>=2 Hz, 1H), 8.75 (bs, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  14.10, 22.68, 25.88, 28.30, 29.17, 29.35, 29.58, 29.64, 29.68, 31.91, 55.73, 65.03, 68.71, 77.20, 111.06, 119.86, 119.86, 120.78, 124.21, 127.27, 147.75, 157.22, 169.25.

**30** **Compound 34:** *tert*-Butyl N-[1-(hydroxymethyl)-2-(2-methoxyanilino)-2-oxethyl]carbamate

To a solution of *o*-anisidine (238 mg, 1.93 mmol) in THF (30 ml), N-(*tert*-butoxycarbonyl)-L-serine  $\beta$ -lactone (200 mg, 1.06 mmol) was added, and the mixture was refluxed for 48 hrs under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and crystallized from

CHCl<sub>3</sub>/hexanes to afford 150 mg (45%) of 34 as a yellow powder: mp 92-94°C; <sup>1</sup>H

**Compound 32:** *tert*-Butyl N-[1-(hydroxymethyl)-2-oxo-2-[3-methoxyanilino]-2-oxethyl]carbamate

To a solution of *m*-anisidine (171 mg, 1.38 mmol) in THF (30 ml), N-(*tert*-butoxycarbonyl)-L-serine  $\beta$ -lactone (200 mg, 1.06 mmol) was added, and the mixture was refluxed overnight under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, to afford 154 mg (46%) of 32 as a yellow oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.48 (s, 9H), 3.68-3.73 (dd, J<sub>1</sub>=8 Hz, J<sub>2</sub>=6.9 Hz, 1H), 3.75 (s, 3H), 4.22-4.25 (d, J<sub>0</sub>=10.23 Hz, 2H), 5.66 (bs, 1H), 6.66-6.69 (m, 1H), 6.96-6.99 (m, 1H), 7.21 (m, J<sub>0</sub>=8.1 Hz, 1H), 7.24 (m, 1H), 8.79 (bs, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  28.28, 29.68, 55.30, 62.39, 77.20, 81.11, 105.67, 110.55, 112.15, 129.73, 138.63, 160.19, 169.89.

**15** **Compound 33:** *tert*-Butyl N-[1-(hydroxymethyl)-2-oxo-2-[2-(tetradecyloxy)anilino]ethyl] carbamate

To a solution of 2-(tetradecyloxy)aniline (200 mg, 0.554 mmol) in THF (25 ml), N-(*tert*-butoxycarbonyl)-L-serine  $\beta$ -lactone (102 mg, 0.545 mmol) was added, and the mixture was refluxed for 48 hrs under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in

vacuo to afford 33 mg (<10%) of 33 as a yellow oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.88 (t, J=6.6 Hz, 3H), 1.26 (s, 2H), 1.48 (s, 9H), 1.76 (m, 2H), 3.67-3.73 (dd, J<sub>1</sub>=5.1 Hz, J<sub>2</sub>=6.9 Hz, 1H), 3.93 (t, J=5.5 Hz, 2H), 4.23-4.26 (m, 2H), 5.66 (bs, 1H), 6.64-6.68 (m, 1H), 6.9-7.0 (m, 1H), 7.19 (t, J<sub>0</sub>=8.1 Hz, 1H), 7.23 (t, J<sub>1</sub>=2 Hz, 1H), 8.75 (bs, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  14.10, 22.68, 25.88, 28.30, 29.17, 29.35, 29.58, 29.64, 29.68, 31.91, 55.73, 65.03, 68.71, 77.20, 111.06, 119.86, 119.86, 120.78, 124.21, 127.27, 147.75, 157.22, 169.25.

**30** **Compound 34:** *tert*-Butyl N-[1-(hydroxymethyl)-2-(2-methoxyanilino)-2-oxethyl]carbamate

To a solution of *o*-anisidine (238 mg, 1.93 mmol) in THF (30 ml), N-(*tert*-butoxycarbonyl)-L-serine  $\beta$ -lactone (200 mg, 1.06 mmol) was added, and the mixture was refluxed for 48 hrs under argon. The reaction mixture was concentrated on a rotary evaporator. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and crystallized from

-46-

-47-

**Example 3-3 - Synthesis of Compounds 35-43**

**Compound 35: N-1-nonyl-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 26 (20 mg, 0.0580 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (1 ml), TFA (1 ml) was added dropwise under argon atmosphere. After the addition was complete, the reaction was allowed to stir at r.t., for 3 hrs, concentrated under reduced pressure at room temperature, and dried on a vacuum pump to give 35 as a white solid (9 mg (95%); mp 168-170 °C; <sup>1</sup>H NMR (CD<sub>3</sub>OD), δ 0.88 (t, J=6.3 Hz, 3H), 1.27 (s, 14H), 1.50 (m, 2H), 1.20 (t, J=6.0 Hz, 2H), 3.70-3.73 (m, 1H), 3.81-3.88 (m, 2H); <sup>13</sup>C NMR (CD<sub>3</sub>OD) δ 14.44, 23.74, 27.96, 30.30, 30.42, 30.47, 30.70, 30.73, 30.78, 30.80, 33.10, 40.71, 56.30, 61.77, 167.97; IR (KBr) 3280, 2919, 2830, 1654, 1573, 1464, 1231, 1141, 1089, 1059, 1059 cm<sup>-1</sup>. Anal. Calcd. for C<sub>17</sub>H<sub>24</sub>N<sub>2</sub>O<sub>2</sub>·CF<sub>3</sub>COOH: C, 50.27; H, 8.16; N, 7.82. Found: C, 50.15; H, 8.30; N, 7.95.

**Compound 36: N-1-tetradecyl-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 27 (50 mg, 0.124 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (1.5 ml), TFA (1.5 ml) was added dropwise under argon atmosphere. After the addition was complete, the reaction was allowed to stir at r.t. for 3 hrs, concentrated under reduced pressure at room temperature, and dried on a vacuum pump to give 36 as a white solid (48 mg (94%); mp 158-171 °C; <sup>1</sup>H NMR (CD<sub>3</sub>OD), δ 0.89 (t, J=6.3 Hz, 3H), 1.28 (s, 22H), 3.22 (t, J=6.0 Hz, 2H), 3.73-3.80 (m, 1H), 3.84-3.91 (m, 2H); <sup>13</sup>C NMR (CD<sub>3</sub>OD) δ 14.43, 23.73, 27.95, 30.29, 30.41, 30.47, 30.69, 30.73, 30.78, 30.80, 33.08, 40.71, 56.29, 61.77, 167.99; IR (KBr) 3277, 2919, 2850, 1636, 1573, 1464, 1231, 1141, 1089, 1059, 1059 cm<sup>-1</sup>. Anal. Calcd. for C<sub>17</sub>H<sub>36</sub>N<sub>2</sub>O<sub>2</sub>·CF<sub>3</sub>COOH: C, 55.06; H, 9.00; N, 6.76. Found: C, 54.94; H, 8.99; N, 6.58.

**Compound 37: N-1-octadecyl-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 28 (25 mg, 0.0547 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (1 ml), TFA (1 ml) was added dropwise under argon atmosphere. After the addition was complete, the reaction was allowed to stir at r.t., for 3 hrs, concentrated under reduced pressure at room temperature, and dried on a vacuum pump to give 37 as a white solid (23 mg (92%); mp 170-172 °C; <sup>1</sup>H NMR (CD<sub>3</sub>OD) δ 0.89 (t, J=6.4 Hz, 3H), 1.27 (s, 30H), 1.49-1.54 (m, 2H), 3.22 (t, J=7.0 Hz, 2H), 3.74-3.81 (m, 1H), 3.83-3.91 (m, 2H); <sup>13</sup>C NMR (CD<sub>3</sub>OD) δ 14.43, 23.74, 27.95, 30.30, 30.41, 30.47, 30.69, 30.78, 33.07, 40.71, 56.30, 61.77, 167.97; IR (KBr) 3276, 2919, 2850, 1657, 1468, 1207, 1181, 1138, 1059 cm<sup>-1</sup>. Anal. Calcd. for C<sub>21</sub>H<sub>44</sub>N<sub>2</sub>O<sub>2</sub>·CF<sub>3</sub>COOH 0.15CH<sub>2</sub>Cl<sub>2</sub>; C, 57.53; H, 9.45; N, 5.80. Found: C, 57.45; H, 9.35; N, 5.81.

**Compound 38: N-1-[4-(tetradecyl)phenyl]-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 29 (34 mg, 0.110 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (0.050 ml), TFA (0.050 ml) was added dropwise under argon atmosphere. After the addition was complete, the reaction was allowed to stir at r.t., for 3 hrs, concentrated under reduced pressure at room temperature, and then, dried on a vacuum pump to give 38 as a white solid (55 mg (99%); mp 135-139 °C; <sup>1</sup>H NMR (CD<sub>3</sub>OD), δ 0.89 (t, J=6.3 Hz, 3H), 1.28 (s, 21H), 1.43 (m, 2H), 1.74 (m, 1H), 1.74 (m, 2H), 2F), 3.86-4.03 (m, 5H), 6.84-6.88 (m, J=9.0 Hz, 2H), 7.41-7.47 (m, J=9.0 Hz, 2H); <sup>13</sup>C NMR (CD<sub>3</sub>OD) δ 14.42, 23.72, 30.41, 30.46, 30.50, 30.67, 30.74, 33.06, 36.81, 61.72, 69.26, 115.71, 122.96, 131.84, 157.80, 166.05; IR (KBr) 3281, 2920, 2852, 1672, 1604, 1559, 1515, 1240, 1210, 1132 cm<sup>-1</sup>. Anal. Calcd. for C<sub>29</sub>H<sub>44</sub>N<sub>2</sub>O<sub>2</sub>·CF<sub>3</sub>COOH: C, 59.27; H, 8.16; N, 5.53. Found: C, 59.48; H, 8.09; N, 5.49.

**Compound 39: N-1-(4-methoxyphenyl)-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 30 (50 mg, 0.161 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (0.049ml), TFA (0.049 ml) was added dropwise under argon atmosphere. After the addition was complete, the reaction was allowed to stir at r.t., for 3 hrs, concentrated under reduced pressure at r.t., and concentrated to dryness *in vacuo* to give 39 as a white solid (50 mg (96%); mp 182-183 °C dec; <sup>1</sup>H NMR (CD<sub>3</sub>OD), δ 3.74-3.89 (s, 3H), 3.87-3.94 (m, 1H), 3.97-4.04 (m, 2H), 6.85-6.91 (m, J=9.1 Hz, 2H), 7.44-7.49 (m, J=9.0 Hz, 2H); <sup>13</sup>C NMR (CD<sub>3</sub>OD) δ 55.86, 56.80, 61.73, 115.07, 122.95, 131.99, 158.31, 166.10; IR (KBr) 3278, 3099, 2964, 1673, 1562, 1517, 1196, 1131, cm<sup>-1</sup>. Anal. Calcd. for C<sub>17</sub>H<sub>24</sub>N<sub>2</sub>O<sub>3</sub>·CF<sub>3</sub>COOH: C, 44.45; H, 4.66; N, 8.64. Found: C, 44.31; H, 4.67; N, 8.58.

**Compound 40: N-1-[3-(tetradecyloxy)phenyl]-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 31 (45 mg, 0.091 mmol) in  $\text{CH}_2\text{Cl}_2$  (0.062 mL), TFA (0.062 mL) was added dropwise under argon atmosphere.

After the addition was complete, the reaction was allowed to stir at r.t. for 3 hrs, concentrated under reduced pressure at room temperature, and dried on a vacuum pump to give 40 as a yellowish green solid 45 mg (99%); mp 115-119 °C;  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 (*s*,  $J=6.5\text{ Hz}$ , 3H), 1.28 (*s*,  $J=2.5\text{ Hz}$ , 1H), 1.43 (*m*, 2H), 1.75 (*m*,  $J=6.5\text{ Hz}$ , 2H), 3.8-3.93 (*m*, 4H), 4.01-4.05 (*m*, 1H), 6.67-6.71 (*m*, 1H), 7.04-7.07 (*m*, 1H), 7.20 (*t*,  $J=8.1\text{ Hz}$ , 1H), 7.28 (*t*,  $J=2.1\text{ Hz}$ , 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.75, 27.18, 30.38, 30.49, 30.52, 30.73, 30.78, 33.09, 56.56, 61.66, 69.05, 107.71, 111.75, 113.16, 130.72, 140.16, 161.07, 166.36; IR (KBr) 3266, 2920, 2852, 1676, 1608, 1566, 1496, 1458, 1211, 1130, 1045  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{23}\text{H}_{40}\text{N}_2\text{O}_3 \cdot \text{CF}_3\text{COOH}$ : C, 59.27; H, 8.16; N, 5.53. Found: C, 59.49; H, 8.13; N, 5.41.

**Compound 41: N-1-[3-methoxyphenyl]-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 32 (120 mg, 0.386 mmol) in  $\text{CH}_2\text{Cl}_2$  (1 mL), TFA (1 mL) was added dropwise under argon atmosphere. After the addition was complete, the reaction was allowed to stir at r.t. for 3 hrs, concentrated under reduced pressure at r.t., and dried on a vacuum pump to give 41 as a offwhite solid 123 mg (98%); mp 131-140 °C;  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  3.77 (*s*, 3H), 3.88-3.99 (*m*, 2H), 4.01-4.06 (*m*, 1H), 6.68-6.71 (*m*, 1H), 7.02-7.10 (*m*, 1H), 7.22 (*t*,  $J=8.1\text{ Hz}$ , 1H), 7.29 (*t*,  $J=2.1\text{ Hz}$ , 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  55.70, 56.94; 61.67, 107.14, 111.11, 113.26, 130.73, 140.22, 161.61, 166.42; IR (KBr) 3265, 1675, 1609, 1566, 1496, 1433, 1268, 1196, 1044,  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{10}\text{H}_{14}\text{N}_2\text{O}_3 \cdot \text{CF}_3\text{COOH}$ : C, 44.45; H, 4.66; N, 8.64. Found: C, 44.18; H, 4.57; N, 8.59.

61.67, 69.84, 112.93, 121.40, 123.38, 126.80, 127.53, 150.93, 166.74; IR (KBr) 3282, 2925, 2851, 1679, 1556, 1496, 1458, 1213, 750,  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{23}\text{H}_{40}\text{N}_2\text{O}_3 \cdot \text{CF}_3\text{COOH}$  0.5H<sub>2</sub>O: C, 58.24; H, 8.21; N, 5.43. Found: C, 58.59; H, 8.09; N, 5.24.

**Compound 43: N-1-[2-methoxyphenyl]-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 34 (80 mg, 0.257 mmol) in  $\text{CH}_2\text{Cl}_2$  (1 mL), TFA (1 mL) was added dropwise under argon atmosphere. After the addition was complete, the reaction was allowed to stir at r.t. for 3 hrs, concentrated under reduced pressure at room temperature, and dried on a vacuum pump to give 43 as a off white solid 81 mg (97%); mp 131-133 °C;  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  3.88 (*s*, 3H), 3.91-4.02 (*m*, 2H), 4.18-4.22 (*m*, 1H), 6.39-6.94 (*m*, 1H), 7.01-7.04 (*m*, 1H), 7.10-7.16 (*t*,  $J=8.1\text{ Hz}$ , 1H), 8.00-8.03 (*t*,  $J=2.1\text{ Hz}$ , 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  56.27, 56.34, 56.47, 61.81, 111.94, 121.52, 123.21, 126.71, 127.54, 151.43, 166.80; IR (KBr) 3271, 1675, 1546, 1499, 1465, 1439, 1268, 1207, 1130,  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{10}\text{H}_{14}\text{N}_2\text{O}_3 \cdot \text{CF}_3\text{COOH}$ : C, 44.45; H, 4.66; N, 8.64. Found: C, 44.18; H, 4.57; N, 8.59.

**Example 4 - Synthesis of Intermediate Compounds 50-54**

The glassware used is flame-dried and cooled to room temperature under an argon atmosphere. The starting alcohol was washed with anhydrous pyridine (3 times), and dried (high vacuum for 48 hrs). The reaction was carried out in an argon atmosphere. THF and  $\text{CH}_2\text{Cl}_2$  were freshly distilled prior to their use.

**Compound 50: *tert*-Butyl N-[1-((di(hexyloxy)phosphoryl)oxy)acetyl]-2-(bromomethyl)-2-oxoethyl] carbonate**

To the pyridine-washed starting 28 (252 mg, 0.551 mmol) was added 1H-tetrazole (231 mg, 3.31 mmol). To this mixture was added a 1:1 mixture of freshly distilled  $\text{THF}(\text{CH}_2\text{Cl}_2)$  (50 mL). After 10 mins, diphenylbis(propyl) phosphoramidate (1.14 g, 3.31 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of pemicetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess pemicetic acid. The THF and  $\text{CH}_2\text{Cl}_2$  were removed under reduced pressure. The concentrate was treated with  $\text{BaOAc}$  (70 mL), and was washed with Na-metabisulfite (2x25 mL),  $\text{NaHCO}_3$  (2x30 mL), water (2x30 mL) ( $\text{CD}_3\text{OD}$ )  $\delta$  14.43, 23.73, 27.07, 30.27, 30.48, 30.57, 30.79, 33.07, 56.198,

**Compound 42: N-1-[2-(tetradecyloxy)phenyl]-2-amino-3-hydroxypropanamide trifluoroacetate**

To a cooled (0 °C, ice bath) solution of 33 (21 mg, 0.044 mmol) in  $\text{CH}_2\text{Cl}_2$  (1 mL), TFA (1 mL) was added dropwise under argon atmosphere. After the addition was complete, the reaction was allowed to stir at r.t., for 3 hrs, concentrated under reduced pressure at room temperature, and dried on a vacuum pump to give 42 as a offwhite solid 21 mg (95%); mp 63-66 °C;  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.88 (*t*,  $J=6.5\text{ Hz}$ , 3H), 1.27 (*s*, 21H), 1.46 (*m*, 2H), 1.83 (*m*,  $J=7.8\text{ Hz}$ , 2H), 3.90-4.07 (*m*, 4H), 4.18 (*t*,  $J=5.8\text{ Hz}$ , 1H), 6.87-6.93 (*m*, 1H), 6.99-7.02 (*m*, 1H), 7.08-7.14 (*m*, 1H), 7.96-7.99 (*m*, 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.43, 23.73, 27.07, 30.27, 30.48, 30.57, 30.79, 33.07, 56.198,

ml), and brine (2x30 ml). The organic portion was dried over  $\text{NaSO}_4$ , and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 195 mg (49 %) of 50 as a colorless oil:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.87 ( $\text{t}$ ,  $J=6.4$  Hz, 3H), 1.25 (br, 2H), 1.34 (m, 2H), 1.44 (s, 9H), 3.17-3.23 (m, 2H), 4.01-4.09 (m, 1H), 4.31-4.43 (m, 2H), 4.96-5.09 (m, 4H), 5.55 (bs, 1H), 6.33 (bs, 1H) 7.31-7.39 (m, 10H);  $^{13}\text{C}$  ( $\text{CDCl}_3$ )  $\delta$  14.09, 22.66, 26.79, 28.25, 29.24, 29.42, 29.50, 29.53, 31.86, 39.68, 66.98, 69.66, 69.73, 77.20, 128.06, 128.10, 128.64, 128.70, 128.72, 135.02, 168.50; MS  $m/z$  603 (M+H); IR (KBr) 3349, 2919, 2852, 1717, 1685, 1654, 1516, 1470, 1457, 1242, 1163, 1037, 1025, 999  $\text{cm}^{-1}$ .

**Compound 51: *tert*-Butyl N-[1-[(di(benzoyloxy)phosphoryl)oxy]methyl]-2-oxo-2-(tetradecylamino)ethyl carbamate**

To the pyridine-washed starting 27 (305 mg, 0.761 mmol) was added 1H-tetrazole (319 mg, 4.56 mmol). To this mixture was added a 1:1 mixture of freshly distilled  $\text{THF}/\text{CH}_2\text{Cl}_2$  (40 ml). After 10 mins, dibenzylidisopropyl phosphoramide (1.57 gm, 4.56 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF and  $\text{CH}_2\text{Cl}_2$  were removed under reduced pressure. The concentrate was treated with  $\text{EtOAc}$  (50 ml), and was washed with Na-metabisulfite (2x25 ml),  $\text{NaHCO}_3$  (2x25 ml), water (2x25 ml), and brine (2x25 ml). The organic portion was dried over  $\text{NaSO}_4$ , and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 135 mg (28 %) of 51 as a white solid: mp 52-54°C;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.87 ( $\text{t}$ ,  $J=6.4$  Hz, 3H), 1.23 (br, 14H), 1.44 (s, 9H), 1.63 (m, 2H), 3.17-3.24 (m, 2H), 4.01-4.09 (m, 1H), 4.30-4.44 (m, 2H), 5.00-5.05 (m, 4H), 5.56 (bs, 1H), 6.32 (bs, 1H) 7.29-7.39 (m, 10H);  $^{13}\text{C}$  ( $\text{CDCl}_3$ )  $\delta$  14.11, 22.68, 26.80, 28.25, 29.26, 29.35, 29.42, 29.52, 29.60, 29.64, 29.69, 31.91, 39.68, 67.00, 67.07, 69.69, 69.74, 77.20, 127.33, 128.06, 128.10, 128.65, 128.70, 128.73, 135.43, 168.51, 170.07; IR (KBr) 3349, 2919, 2852, 1717, 1685, 1654, 1516, 1242, 1163, 1037, 1025, 999  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{24}\text{H}_{36}\text{N}_2\text{O}_2\text{P}$ : 0.75H<sub>2</sub>O; C, 64.36; H, 9.17; N, 3.42. Found: C, 64.23; H, 9.05; N, 3.78.

**Compound 52: *tert*-Butyl N-[1-[(di(benzoyloxy)phosphoryl)oxy]methyl]-2-oxo-2-(octadecylamino)-2-oxoethyl carbamate**

To the pyridine-washed starting 26 (270 mg, 0.783 mmol) was added 1H-tetrazole (329 mg, 4.70 mmol). To this mixture was added a 1:1 mixture of freshly distilled  $\text{THF}/\text{CH}_2\text{Cl}_2$  (50 ml). After 10 mins, dibenzylidisopropyl phosphoramide (1.62 gm, 4.70 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF and  $\text{CH}_2\text{Cl}_2$  were removed under reduced pressure. The concentrate was treated with  $\text{EtOAc}$  (50 ml), and was washed with Na-metabisulfite (2x25 ml),  $\text{NaHCO}_3$  (2x25 ml), water (2x25 ml), and brine (2x25 ml). The organic portion was dried over  $\text{NaSO}_4$ , and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 135 mg (28 %) of 52 as a white solid: mp 52-54°C;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.87 ( $\text{t}$ ,  $J=6.4$  Hz, 3H), 1.23 (br, 14H), 1.44 (s, 9H), 1.63 (m, 2H), 3.17-3.24 (m, 2H), 4.01-4.09 (m, 1H), 4.30-4.44 (m, 2H), 5.00-5.05 (m, 4H), 5.56 (bs, 1H), 6.32 (bs, 1H) 7.29-7.39 (m, 10H);  $^{13}\text{C}$  ( $\text{CDCl}_3$ )  $\delta$  14.11, 22.68, 26.80, 28.25, 29.26, 29.35, 29.42, 29.52, 29.60, 29.64, 29.69, 31.91, 39.68, 67.00, 67.07, 69.69, 69.74, 77.20, 127.33, 128.06, 128.10, 128.65, 128.70, 128.73, 135.43, 168.51, 170.07; IR (KBr) 3349, 2919, 2852, 1717, 1685, 1654, 1516, 1242, 1163, 1037, 1025, 999  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{30}\text{H}_{54}\text{N}_2\text{O}_2\text{P}$ : 0.75H<sub>2</sub>O; C, 64.36; H, 9.17; N, 3.42. Found: C, 62.80; H, 8.38; N, 4.21.

**Compound 53: *tert*-Butyl N-[1-[(di(benzoyloxy)phosphoryl)oxy]methyl]-2-oxo-2-[4-(tetradecyloxy)anilino]ethyl carbamate**

To the pyridine-washed starting 29 (310 mg, 0.647 mmol) was added 1H-tetrazole (450 mg, 6.42 mmol). To this mixture was added a 1:1 mixture of freshly distilled  $\text{THF}/\text{CH}_2\text{Cl}_2$  (40 ml). After 10 mins, dibenzylidisopropyl phosphoramide (2.21 gm, 6.42 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF and  $\text{CH}_2\text{Cl}_2$  were

removed under reduced pressure. The concentrate was treated with  $\text{BzOAc}$  (70 ml), and was washed with  $\text{Na}-\text{metabisulfite}$  ( $2 \times 25$  ml),  $\text{NaHCO}_3$  ( $2 \times 35$  ml), water ( $2 \times 35$  ml), and brine ( $2 \times 35$  ml). The organic portion was dried over  $\text{NaSO}_4$ , and concentrated under reduced pressure. The residue was subjected to flash column

**5** chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in *vacuo* to afford 81 mg (17%) of 53 as a white solid: mp 74-76°C;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.87 ( $t$ ,  $J=6.5$  Hz, 3H), 1.30 ( $s$ , 22H), 1.46 ( $s$ , 9H), 1.71-1.80 ( $m$ , 2H), 3.91 (*t*,  $J=6.5$  Hz, 3H), 4.01-4.16 ( $m$ , 1H), 4.42-4.49 ( $m$ , 2H), 4.96-5.09 ( $m$ , 4H), 5.65 ( $bs$ , 1H), 6.80-6.86 ( $m$ ,  $J=9.0$  Hz, 2H), 7.31-7.39 ( $m$ , 12H), 8.82 ( $bs$ , 1H);  $^{13}\text{C}$  ( $\text{CDCl}_3$ )  $\delta$  14.10, 22.67, 26.02, 28.26, 29.26, 29.34, 29.40, 29.57, 29.64, 31.91, 68.31, 69.84, 77.20, 114.79, 121.72, 128.07, 128.13, 128.65, 128.74, 130.03, 166.71; IR ( $\text{KBr}$ ) 3340, 2920, 2852, 1717, 1677, 1513, 1457, 1237, 1059, 998  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{14}\text{H}_{26}\text{N}_2\text{O}_2\text{P}$ : C, 66.31; H, 8.63; N, 3.46. Found: C, 65.92; H, 9.02; N, 3.84.

**15** **Compound 54: *tert*-Butyl  $N$ -(1-[(di(2-oxoethyl)phosphoryloxy) methyl]-2-(4-methoxyanilino)-2-oxethyl]carbamate**

To the pyridine-washed starting 30 (225 mg, 0.725 mmol) was added 1H-tetrazole (254 mg, 3.625 mmol). To this mixture was added a 1:1 mixture of freshly distilled  $\text{THF}/\text{CH}_2\text{Cl}_2$  (20 ml). After 10 mins, dibenzylidioisopropyl phosphorimidate (1.25 gm, 3.625 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of *peracetic acid* was added. The mixture was stirred for another 35 mins, followed by the addition of  $\text{Na}-\text{metabisulfite}$  to quench the excess *peracetic acid*. The  $\text{THF}$  and  $\text{CH}_2\text{Cl}_2$  were removed on a rotary evaporator. The concentrate was treated with  $\text{BzOAc}$  (50 ml), and was washed with  $\text{Na}-\text{metabisulfite}$  ( $2 \times 15$  ml),  $\text{NaHCO}_3$  ( $2 \times 25$  ml), water ( $2 \times 25$  ml), and brine ( $2 \times 25$  ml). The organic portion was dried over  $\text{NaSO}_4$ , and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in *vacuo* to afford 195 mg (47%) of 54 as a white solid: mp 82-84°C;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$  1.44 ( $s$ , 9H), 4.11 ( $s$ , 3H), 4.09-4.18 ( $m$ , 1H), 4.43-4.51 ( $m$ , 2H), 4.98-5.05 ( $m$ , 4H), 5.72 ( $bs$ , 1H), 6.78-6.82 ( $m$ ,  $J=9.0$  Hz, 2H), 7.26-7.33 ( $m$ , 10H), 7.36-7.41 ( $m$ ,  $J=9.0$  Hz, 2H), 8.41 ( $bs$ , 1H);  $^{13}\text{C}$  ( $\text{CDCl}_3$ )  $\delta$  28.26, 55.45, 66.93, 67.00, 69.76, 69.83, 69.90, 77.20, 80.91, 114.11, 121.75, 128.06, 128.12, 128.64, 128.72, 128.73, 130.38, 135.28, 135.42, 156.62, 166.75;  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  16.72 (1P); IR ( $\text{KBr}$ ) 3337, 2959, 1716,

**5** **Example 5 - Synthesis of Compounds 55-59**

**Compound 55: 2-Amino-3-(nonylamino)-3-oxopropyl dihydrogen phosphate**

To a solution of 50 (100 mg, 0.165 mmol) in  $\text{EtOH}$  (15 ml) was added 10 % Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 50 psi. After 4 hours TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 48 mg (90%) of 55 as a white powder: mp 196-198 °C;  $^1\text{H NMR}$  ( $\text{CF}_3\text{COOD}$ )  $\delta$  0.81-0.82 ( $m$ , 3H), 1.26-1.30 ( $m$ , 14H), 1.59 ( $m$ , 2H), 3.37-3.38 ( $m$ , 2H), 4.54-4.59 ( $m$ , 1H), 4.72-4.81 ( $m$ , 2H);  $^{13}\text{C}$  NMR ( $\text{CF}_3\text{COOD}$ )  $\delta$  14.66, 24.39, 28.60, 28.60, 30.46, 30.94, 31.16, 31.30, 31.39, 33.81, 43.53, 57.21, 66.42, 167.86; MS  $m/z$  323 (M $^+$ ); IR ( $\text{KBr}$ ) 3314, 2920, 2853, 1670, 1575, 1477, 1246, 1063, 1043  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{13}\text{H}_{28}\text{N}_2\text{O}_2\text{P}$ : C, 47.64; H, 9.18; N, 8.23. Found: C, 47.24; H, 8.84; N, 8.02.

**20** **Compound 56: 2-Amino-3-oxo-3-(tetradecylamino)propyl dihydrogen phosphate**

To a solution of 51 (145 mg, 0.219 mmol) in  $\text{EtOH}$  (15 ml) was added 10 % Pd/C (catalytic amount). Hydrogenation was carried out for 3 hrs at 45 psi. After 3 hours TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 75 mg (90%) of 56 as a white powder: mp 189-190 °C;  $^1\text{H NMR}$  ( $\text{CF}_3\text{COOD}$ )  $\delta$  0.81 ( $bs$ , 3H), 1.24 ( $s$ , 23H), 1.57 ( $m$ , 2H), 3.37 ( $m$ , 2H), 4.54-4.58 ( $m$ , 1H), 4.73-4.78 ( $m$ , 2H);  $^{13}\text{C}$  NMR ( $\text{CF}_3\text{COOD}$ )  $\delta$  14.43, 24.16, 28.34, 30.21, 30.69, 31.01, 31.17, 31.22, 31.27, 33.62, 43.27, 56.96, 66.16, 67.60;  $^{31}\text{P}$  NMR ( $\text{CF}_3\text{COOD}$ )  $\delta$  17.93 (1P); MS  $m/z$  379 (M $^+$ ); IR ( $\text{KBr}$ ) 3318, 2923, 2852, 1671, 1657, 1563, 1475, 1242, 1055  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{20}\text{H}_{45}\text{N}_2\text{O}_2\text{P}$ : C, 53.67; H, 9.80; N, 7.36. Found: C, 53.40; H, 9.73; N, 7.31.

**35** **Compound 56a: 2-(Acethylamino)-3-oxo-3-(tetradecylamino)propyl dihydrogen phosphate**

To a sample of 56 (20 mg, 0.052 mmol) in 0.5 ml pyridine was added a large excess of acetic anhydride. The mixture was allowed to stir at r.t. overnight. Excess pyridine and acetic anhydride were on a rotary evaporator. The resultant mixture was stirred with 20 ml of aqueous HCl. The acidic mixture was extracted

with  $\text{EtOAc}$  (2×25 ml). The  $\text{EtOAc}$  layer was washed with water (2×25 ml) and brine (2×25 ml). The organic portion was dried over  $\text{NaSO}_4$  and filtered. The eluate was concentrated under reduced pressure to afford 15 mg (71%) of 56a as a gummy solid. <sup>1</sup>H NMR ( $\text{CD}_3\text{OD}$ ) δ 0.89 (t,  $J$ =6.3 Hz, 3H), 1.27 (s, 22H), 1.99-2.02 (m, 3H), 3.15-3.20 (m, 2H), 4.10-4.28 (m, 2H), 4.54-4.62 (m, 1H); <sup>13</sup>C NMR ( $\text{CDCl}_3/\text{CD}_3\text{OD}$ ) 13.48, 16.19, 22.23, 26.50, 28.91, 29.21, 31.48, 30.21, 31.01, 31.17, 31.22, 31.27, 33.62, 43.27, 56.96, 66.16, 163.02, 174.96; IR (KBr) 3316, 2953, 1671, 1657, 1560, 1467, 1247, 1059  $\text{cm}^{-1}$ .

**Compound 57: 2-Amino-3-(octadecylamino)-3-oxopropyl dihydrogen phosphate**

To a solution of 52 (117 mg, 0.164 mmol) in  $\text{EtOH}$  (15 ml) was added 10 % Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 50 psi.

After 4 hours TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 70 mg (98 %) of 57 as a white powder: mp 190-192 °C; <sup>1</sup>H NMR ( $\text{CF}_3\text{COOD}$ ) δ 0.81 (t,  $J$ =6.9 Hz, 3H), 1.25 (s, 31H), 1.58 (m, 2H), 3.34-3.44 (m, 2H), 4.49-4.59 (m, 1H), 4.71-4.81 (m, 2H); <sup>13</sup>C NMR ( $\text{CF}_3\text{COOD}$ ) δ 14.70, 24.43, 28.60, 30.46, 30.95, 31.28, 31.31, 31.44, 31.48, 31.55, 33.89, 43.53, 57.12, 57.21, 66.35, 167.85; MS  $m/z$  435 (M+H); IR (KBr) 3325, 2922, 2852, 1674, 1655, 1560, 1472, 1045  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{21}\text{H}_{43}\text{N}_2\text{O}_6\text{P}$ : C, 57.77; H, 10.39; N, 6.42. Found: C, 57.61; H, 10.22; N, 6.25.

**Compound 58: 2-Amino-3-oxo-3-(tetradecyloxy)aniline] propyl dihydrogen phosphate**

To a solution of 53 (40 mg, 0.054 mmol) in  $\text{EtOH}$  (15 ml) was added 10 % Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 50 psi. After 4 hours TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 22 mg (88 %) of 58 as a white powder: mp 187-190 °C; <sup>1</sup>H NMR ( $\text{CF}_3\text{COOD}$ ) δ 0.80-0.82 (m, 3H), 1.25 (m, 20H), 1.77-1.84 (m, 2H), 4.20 (t,  $J$ =6.0 Hz, 2H), 4.64-4.74 (m, 1H), 4.90-4.91 (m, 2H), 7.04-7.07 (d,  $J$ =9.0 Hz, 2H), 7.32-7.35 (d,  $J$ =9.0 Hz, 2H); <sup>13</sup>C NMR ( $\text{CF}_3\text{COOD}$ ) δ 14.81, 24.54, 27.57, 30.62, 31.19, 31.38, 31.46, 31.52, 31.60, 31.65, 33.95, 57.70, 66.53, 73.66, 119.32, 126.55, 131.25, 158.87, 167.06; MS  $m/z$  471 (M+H); IR (KBr) 3325, 2923, 2832, 1665, 1553, 1515, 1469, 1240, 1046  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{24}\text{H}_{41}\text{N}_2\text{O}_6\text{P}$ : C, 52.58; H, 8.00; N, 5.11. Found: C, 52.89; H, 7.83; N, 5.29.

**Compound 59: 2-Amino-3-(4-methoxyanilino)-3-oxopropyl dihydrogen phosphate**

To a solution of 54 (125 mg, 0.219 mmol) in  $\text{EtOH}$  (15 ml) was added 10 % Pd/C (catalytic amount). Hydrogenation was carried out for 2 hrs at 45 psi. After 2 hours TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 82 mg (96 %) of 59 as a white powder: mp 199-202 °C; <sup>1</sup>H NMR ( $\text{CF}_3\text{COOD}$ ) δ 3.93 (s, 3H), 4.65-4.75 (m, 1H), 4.88-4.94 (m, 2H), 7.01-7.04 (d,  $J$ =9.0 Hz, 2H), 7.31-7.34 (d,  $J$ =9.0 Hz, 2H); <sup>13</sup>C NMR ( $\text{CDCl}_3$ ) δ 57.60, 58.00, 66.54, 117.69, 126.64, 131.07, 159.62, 167.07; MS  $m/z$  289 (M+H); IR (KBr) 3317, 2961, 1680, 1565, 1515, 1478, 1236, 1045  $\text{cm}^{-1}$ ; Anal. Calcd. for  $\text{C}_{10}\text{H}_{13}\text{N}_2\text{O}_6\text{P}$ : C, 41.39; H, 5.21; N, 9.65. Found: C, 41.25; H, 5.35; N, 9.73.

**Example 6 - Synthesis of Intermediate Compounds 63-65**

15 The glassware used was flame-dried and cooled to room temperature under an argon atmosphere. The starting alcohol was washed with anhydrous pyridine (3 times) and dried on high vacuum for 48 hrs. The reaction was carried out in an argon atmosphere. THF and  $\text{CH}_2\text{Cl}_2$  were freshly distilled prior to their use.

20 Compound 63: 1,2-3-Ortadecyloxypropane]-bis(dihenylphosphate)

To the pyridine-washed starting *d*-butyl alcohol (60, 225 mg, 0.652 mmol) was added 1H-terazole (229 mg, 3.26 mmol). To this mixture was added a 1:1 mixture of freshly distilled THF/ $\text{CH}_2\text{Cl}_2$  (50 ml). After 10 mins, diisobutyl phosphoramide (1.12 gm, 3.26 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF and  $\text{CH}_2\text{Cl}_2$  were removed under reduced pressure. The concentrate was treated with  $\text{EtOAc}$  (70 ml), and was washed with Na-metabisulfite (2×25 ml),  $\text{NaHCO}_3$  (2×30 ml), water (2×30 ml), and brine (2×30 ml). The organic portion was dried over  $\text{NaSO}_4$ , and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with  $\text{EtOAc}/\text{hexanes}$  of various compositions. Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 303 mg (53 %) of 63 as a clear oil: <sup>1</sup>H NMR ( $\text{CDCl}_3$ ) δ 0.66 ( $t$ ,  $J$ =6.4 Hz, 3H), 1.24 (bm, 28H), 1.33-1.35 (m, 2H), 1.45 (m, 2H), 3.29-3.36 (m, 2H), 3.48-

3.50 (d,  $J=5.2$  Hz, 2H), 4.04-4.22 (m, 2H), 4.60 (m, 1H), 5.00 (m, 8H), 7.27-7.33 (m, 20H);  $^{13}\text{C}$  ( $\text{CDCl}_3$ )  $\delta$  14.05, 18.96, 22.62, 25.95, 29.29, 29.41, 29.49, 29.53, 29.59, 29.63, 31.85, 46.48, 66.58, 69.20, 69.23, 69.28, 69.36, 71.75, 75.37, 127.76, 127.82, 127.86, 127.88, 127.94, 128.36, 128.45, 128.49, 128.61, 128.62, 135.46, 135.54, 135.59, 135.65, 135.68, 135.75, 135.79; MS  $m/z$  866 ( $\text{M}+\text{H}^+$ ).

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**Compound 66: 1,2-(3-Dodecyloxypropane)-  
bis(dibenzylphosphate)**

To the pyridine-washed starting *d*-3-O-n-hexadecyl-1,2-propanediol (61, 400 mg, 1.5 mmol) was added 1H-tetrazole (645 mg, 9.2 mmol). To this mixture was added a 1:1 mixture of freshly distilled THF/CH<sub>2</sub>Cl<sub>2</sub> (40 mL). After 10 mins, dibenzylidisopropyl phosphororamidate (3.18 g, 9.2 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of pemicetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess pemicetic acid. The THF and CH<sub>2</sub>Cl<sub>2</sub> were removed under reduced pressure. The concentrate was treated with EtOAc (80 mL), and was washed with Na-metabisulfite (2x35 mL), NaHCO<sub>3</sub> (2x40 mL), water (2x30 mL), and brine (2x30 mL). The organic portion was dried over Na<sub>2</sub>O<sub>4</sub>, and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 100 mg (<10 %) of 64 as a clear oil:  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.86 (t,  $J=6.3$  Hz, 3H), 1.23 (brn, 18H), 1.46 (m, 2H), 3.13-3.36 (m, 2H), 3.49-3.51 (d,  $J=5.2$  Hz, 2H), 4.03-4.23 (m, 2H), 4.59 (m, 1H), 5.01 (m, 8H), 7.26-7.34 (m, 20H);  $^{13}\text{C}$  ( $\text{CDCl}_3$ )  $\delta$  14.11, 22.68, 26.00, 29.35, 29.47, 29.54, 29.59, 29.66, 31.91, 69.01, 69.06, 69.26, 69.29, 69.34, 69.41, 71.82, 71.74, 75.52, 75.60, 77.20, 126.97, 127.82, 127.88, 127.93, 127.95, 127.99, 128.43, 128.51, 128.55, 128.60, 135.63, 135.73, 135.79, 135.83; IR (NaCl, neat) 3423, 1269, 1016, 736, cm<sup>-1</sup>; MS  $m/z$  837 ( $\text{M}+\text{H}^+$ ),  $m/z$  859 ( $\text{M}+\text{Na}^+$ ).

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**Example 7 - Synthesis of Compounds 66-68**

**Compound 66: 1,2-(3-Octadecyloxypropane)-bis(dihydrogen phosphate)**

To a solution of 63 (135 mg, 0.156 mmol) in EtOH (15 mL) was added 10 % Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 60 psi.

After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 70 mg (89 %) of 66 as a clear wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 (t,  $J=6.4$  Hz, 3H), 1.28 (s, 30H), 1.55 (m, 2H), 3.45-3.50 (m, 2H), 3.62-3.64 (m, 2H), 4.00-4.16 (m, 2H), 4.47 (m, 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.43, 19.30, 23.73, 27.20, 30.47, 30.64, 30.77, 72.80; MS  $m/z$  503 ( $\text{M}+\text{H}^+$ ); IR (NaCl Neat) 1011 cm<sup>-1</sup>.

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**Compound 65: 1,2-(3-Hexadecyloxypropane)-  
bis(dibenzylphosphate)**

To the pyridine-washed starting *d*-3-O-n-hexadecyl-1,2-propanediol (62, 500 mg, 1.57 mmol) was added 1H-tetrazole (664 mg, 9.47 mmol). To this mixture was added a 1:1 mixture of freshly distilled THF/CH<sub>2</sub>Cl<sub>2</sub> (50 mL). After 10 mins, dibenzylidisopropyl phosphororamidate (3.27 g, 9.47 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction

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To the pyridine-washed starting *d*-3-O-n-hexadecyl-1,2-propanediol (62, 500 mg, 1.57 mmol) was added 1H-tetrazole (664 mg, 9.47 mmol). To this mixture was added a 1:1 mixture of freshly distilled THF/CH<sub>2</sub>Cl<sub>2</sub> (50 mL). After 10 mins, dibenzylidisopropyl phosphororamidate (3.27 g, 9.47 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction

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To a solution of 64 (70 mg, 0.059 mmol) in EtOH (15 mL) was added 10 % Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to

afford 35 mg (94 %) of 67 as a clear wax.  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.79 ( $t$ ,  $J$ =6.7 Hz, 3H), 1.90 ( $s$ , 18H), 1.46 (m, 2H), 3.34-3.41 (m, 2H), 3.49-3.73 (m, 2H), 3.78-4.05 (m, 2H), 4.47 (m, 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.43, 23.71, 23.74, 27.20, 30.49, 30.64, 30.76, 30.81, 33.08, 66.80, 72.79; MS m/z 419 (M-H) $^+$ ; IR (NaCl Neat) 1008  $\text{cm}^{-1}$ .

**Compound 68:** 1,2-(3-Heptadecyloxypropane)-bis(dihydrogen phosphate)

To a solution of 65 (138 mg, 0.164 mmol) in EtOH (15 ml) was added 10 % Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 75 mg (96 %) of 68 as a clear wax.  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t$ ,  $J$ =6.4 Hz, 3H), 1.28 ( $s$ , 23H), 1.56 (m, 2H), 3.43-3.50 (m, 2H), 3.58-3.65 (m, 2H), 3.89-4.16 (m, 2H), 4.47 (m, 1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.74, 27.20, 30.48, 30.64, 30.80, 33.08, 72.80; MS m/z 475 (M-H) $^+$ ; IR (NaCl Neat) 1011  $\text{cm}^{-1}$ .

Example 8 - Synthesis of Intermediate Compounds 77-84

The glassware used was flame-dried and cooled to room temperature under an argon atmosphere. The starting alcohol was washed with anhydrous pyridine (3 times) and dried on high vacuum for 48 hrs. The reaction was carried out in an argon atmosphere. THF and  $\text{CH}_3\text{Cl}_2$  were freshly distilled prior to their use.

**Compound 77:** 1,2-(3-Tetradecanoxypropane)-bis(dihydroxyphosphate)

To the pyridine-washed starting monomyristine (69, 800 mg, 2.6 mmol) was added 1H-terazole (1.0 gm, 14.5 mmol). To this mixture was added freshly distilled THF (45 ml). After 10 mins, dibenzylidisisopropyl phosphoramidate (5.02 gm, 14.5 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF was removed under reduced pressure. The concentrate was treated with EtOAc (100 ml), and was washed with Na-metabisulfite (2x50 ml),  $\text{NaHCO}_3$  (2x100 ml), water (2x50 ml), and brine (2x50 ml). The organic portion was dried over  $\text{NaSO}_4$ , and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 741 mg (35 %) of 78 as a clear oil.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.87 ( $t$ ,  $J$ =6.4 Hz, 3H), 1.25 (bm, 22H), 1.53 (m, 2H), 2.17-2.32 (m, 2H), 3.95-4.24 (m, 4H), 4.61-4.70 (m, 1H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  14.09, 22.66, 24.69, 29.08, 29.23, 29.33, 29.44, 29.59, 29.62, 29.65, 31.89, 33.85, 54.23, 65.86, 69.40, 69.46, 69.48, 69.53, 69.56, 77.20, 127.84, 127.90, 127.97, 127.98, 128.03, 128.56, 128.59, 128.69, 128.71, 135.50, 135.59, 173.09; IR (NaCl, Neat) 3421, 1742, 1457, 1275, 1035, 1014, 1001  $\text{cm}^{-1}$ ; MS m/z 837 (M+H) $^+$ , m/z 859 (M+Na) $^+$ .

**Compound 78:** 1,2-(3-Pentadecanoxypropane)-bis(dihydroxyphosphate)

To the pyridine-washed starting monopentadecanoic (70, 800 mg, 2.5 mmol) was added 1H-terazole (970 mg, 13.9 mmol). To this mixture was added freshly distilled THF (45 ml). After 10 mins, dibenzylidisisopropyl phosphoramidate (4.80 gm, 13.9 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF was removed under reduced pressure. The concentrate was treated with EtOAc (100 ml), and was washed with Na-metabisulfite (2x50 ml),  $\text{NaHCO}_3$  (2x100 ml), water (2x50 ml), and brine (2x50 ml). The organic portion was dried over  $\text{NaSO}_4$ , and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

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Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 741 mg (35 %) of 78 as a clear oil.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  0.87 ( $t$ ,  $J$ =6.4 Hz, 3H), 1.25 (bm, 22H), 1.53 (m, 2H), 2.17-2.32 (m, 2H), 3.95-4.24 (m, 4H), 4.61-4.70 (m, 1H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  14.09, 22.66, 24.69, 29.08, 29.23, 29.33, 29.44, 29.59, 29.62, 29.65, 31.89, 33.85, 54.23, 65.86, 69.40, 69.46, 69.48, 69.53, 69.56, 77.20, 127.84, 127.90, 127.97, 127.98, 128.03, 128.56, 128.59, 128.69, 128.71, 135.50, 135.59, 173.09; IR (NaCl, Neat) 3422, 1742, 1457, 1274, 1035, 1001  $\text{cm}^{-1}$ ; MS m/z 8223 (M+H) $^+$ , m/z 845 (M+Na) $^+$ .

**Compound 79:** 1,2-(3-Hexadecanoxypropane)-bis(dihydroxyphosphate)

To the pyridine-washed starting monopalmitin (71, 800 mg, 2.4 mmol) was added 1H-terazole (1.00 gm, 14.2 mmol). To this mixture was added freshly distilled THF (45 ml). After 10 mins, dibenzylidisisopropyl phosphoramidate (4.90 gm,



product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF was removed under reduced pressure. The concentrate was treated with EtOAc (100 ml), and was washed with Na-metabisulfite (2×50 ml), NaHCO<sub>3</sub> (2×125 ml), water (2×75 ml), and brine (2×50 ml). The organic portion was dried over NaSO<sub>4</sub>, and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 1.47 gm (78 %) of 52 as a clear oil: <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.87 (t, J=6.3 Hz, 3H), 1.23-1.25 (bm, 30H), 1.53 (m, 2H), 2.20 (t, J=7.2 Hz, 2H), 4.02-4.24 (m, 4H), 4.66 (m, 1H), 4.99-5.03 (m, 8H), 7.29-7.36 (m, 20H); <sup>13</sup>C (CDCl<sub>3</sub>) δ 14.08, 22.65, 24.67, 29.06, 29.22, 29.32, 29.43, 29.58, 29.61, 29.66, 31.88, 33.83, 62.25, 65.84, 69.38, 69.46, 69.51, 69.54, 74.03, 74.10, 74.15, 74.22, 77.20, 127.82, 127.88, 127.96, 128.53, 128.56, 135.45, 135.55, 173.06, IR (NaCl, Neat) 3483, 1743, 1457, 1273, 1282, 1216, 1035, 1013 cm<sup>-1</sup>; MS m/z 893 (M+H)<sup>+</sup>, m/z 915 (M+Na)<sup>+</sup>.

**Compound 83: 1,2,4,3-tetraenoxypropane-  
bis(dibenzylphosphate)**

To the pyridine-washed starting Monacanthidin (75, 800 mg, 2.06 mmol) was added 1H-tetrazole (1.00 gm, 14.2 mmol). To this mixture was added freshly distilled THF (40 ml). After 10 mins, dibenzyl/diisopropyl phosphoramidate (4.92 gm, 14.2 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF was removed under reduced pressure. The concentrate was treated with EtOAc (100 ml), and was washed with Na-metabisulfite (2×50 ml), NaHCO<sub>3</sub> (2×125 ml), water (2×75 ml), and brine (2×50 ml). The organic portion was dried over NaSO<sub>4</sub>, and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 1.27 gm (71 %) of 84 as a white wax like compound: <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.87 (t, J=6.4 Hz, 3H), 1.23-1.25 (bm, 36H), 1.53 (m, 25), 2.20 (t, J=7.2 Hz, 2H), 4.02-4.24 (m, 4H), 4.66 (m, 1H), 4.99-5.03 (m, 8H), 7.29-7.36 (m, 20H); <sup>13</sup>C (CDCl<sub>3</sub>) δ 14.08, 22.65, 24.68, 29.07, 29.22, 29.32, 29.44, 29.59, 29.62, 29.66, 31.88, 33.84, 62.20, 62.26, 65.85, 69.40, 69.45, 69.48, 69.53, 69.57, 74.05, 74.16, 74.24, 77.20, 127.83, 127.88, 127.96, 127.97, 128.30, 128.52, 128.54, 128.57, 128.58, 135.46, 135.55, 173.07; MS m/z 935 (M+H)<sup>+</sup>, m/z 957 (M+Na)<sup>+</sup>.

**Example 2 - Synthesis of Compounds 85-92**

**Compound 85: 1,2,4,3-Tetradecanoxypropane-bis(dihydrogen phosphate)**

To a solution of 77 (385 mg, 0.468 mmol) in EtOH (15 ml) was added 10 %Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 50 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 210 mg (98 %) of 85 as a white wax: <sup>1</sup>H NMR (CD<sub>3</sub>OD) δ 0.89 (t, J=6.4 Hz, 3H), 1.23-1.25 (bm, 32H), 1.53 (m, 2H), 2.20 (t, J=7.2 Hz, 2H), 4.02-4.24 (m, 4H), 4.66 (m, 1H), 4.99-5.05 (m, 8H), 7.29-7.36 (m, 20H); <sup>13</sup>C (CDCl<sub>3</sub>) δ 14.09, 22.65, 24.69, 29.07, 29.23, 29.33, 29.44, 29.59, 29.63, 29.67, 31.89, 33.84, 62.21, 62.27, 65.86, 69.40, 69.45, 69.48, 69.52, 69.56, 74.05, 74.12, 74.16, 74.24, 77.20,

127.83, 127.89, 127.97, 128.53, 128.55, 128.57, 128.59, 135.47, 135.56, 173.07; IR (NaCl, Neat) 3483, 1743, 1457, 1273, 1282, 1216, 1035, 1012, 1000 cm<sup>-1</sup>; MS m/z 907 (M+H)<sup>+</sup>, m/z 929 (M+Na)<sup>+</sup>.

**Compound 84: 1,2,4,3-Docosanoxypropane-bis(dibenzylphosphate)**

To the pyridine-washed starting Monobenzen (76, 800 mg, 1.92 mmol) was added 1H-tetrazole (1.00 gm, 14.2 mmol). To this mixture was added freshly distilled THF (40 ml). After 10 mins, dibenzyl/diisopropyl phosphoramidate (5.14 gm, 14.8 mmol) was added, and the reaction was stirred under an argon atmosphere for 90 mins. The TLC of the reaction mixture showed the formation of the product. This mixture was cooled to 0 °C (ice bath), and a large excess of peracetic acid was added. The mixture was stirred for another 35 mins, followed by the addition of Na-metabisulfite to quench the excess peracetic acid. The THF was removed under reduced pressure. The concentrate was treated with EtOAc (100 ml), and was washed with Na-metabisulfite (2×50 ml), NaHCO<sub>3</sub> (2×125 ml), water (2×75 ml), and brine (2×50 ml). The organic portion was dried over NaSO<sub>4</sub>, and concentrated under reduced pressure. The residue was subjected to flash column chromatography, eluting with EtOAc/hexanes of various compositions.

Appropriate fractions were pooled, and concentrated to dryness in vacuo to afford 1.27 gm (71 %) of 84 as a white wax like compound: <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.87 (t, J=6.4 Hz, 3H), 1.23-1.25 (bm, 36H), 1.53 (m, 25), 2.20 (t, J=7.2 Hz, 2H), 4.02-4.24 (m, 4H), 4.66 (m, 1H), 4.99-5.03 (m, 8H), 7.29-7.36 (m, 20H); <sup>13</sup>C (CDCl<sub>3</sub>) δ 14.08, 22.65, 24.68, 29.07, 29.22, 29.32, 29.44, 29.59, 29.62, 29.66, 31.88, 33.84, 62.20, 62.26, 65.85, 69.40, 69.45, 69.48, 69.53, 69.57, 74.05, 74.16, 74.24, 77.20, 127.83, 127.88, 127.96, 127.97, 128.30, 128.52, 128.54, 128.57, 128.58, 135.46, 135.55, 173.07; MS m/z 935 (M+H)<sup>+</sup>, m/z 957 (M+Na)<sup>+</sup>.

**Example 2 - Synthesis of Compounds 85-92**

**Compound 85: 1,2,4,3-Tetradecanoxypropane-bis(dihydrogen phosphate)**

To a solution of 77 (385 mg, 0.468 mmol) in EtOH (15 ml) was added 10 %Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 50 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 210 mg (98 %) of 85 as a white wax: <sup>1</sup>H NMR (CD<sub>3</sub>OD) δ 0.89 (t, J=6.4 Hz, 3H), 1.23-1.25 (bm, 32H), 1.53 (m, 2H), 2.20 (t, J=7.2 Hz, 2H), 4.02-4.24 (m, 4H), 4.66 (m, 1H), 4.99-5.05 (m, 8H), 7.29-7.36 (m, 20H); <sup>13</sup>C (CDCl<sub>3</sub>) δ 14.09, 22.65, 24.69, 29.07, 29.23, 29.33, 29.44, 29.59, 29.63, 29.67, 31.89, 33.84, 62.21, 62.27, 65.86, 69.40, 69.45, 69.48, 69.52, 69.56, 74.05, 74.12, 74.16, 74.24, 77.20,

1H);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.73, 26.09, 30.71, 30.23, 30.43, 30.47, 30.61, 30.75, 33.07, 34.80, 34.94, 61.90, 61.96, 63.70, 66.24, 74.33, 77.51, 175.02; MS  $m/z$  461 (M+H); IR (NaCl Neat) 3386, 1702, 1216, 1019  $\text{cm}^{-1}$ .

5 **Compound 86: 1,2-(3-Pentadecanoyloxypropane)-bis(dihydrogen phosphate)**

To a solution of 78 (451 mg, 0.538 mmol) in EtOH (15 ml) was added 10 %Pd/C (catalytic amount). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 250 mg (97 %) of 86 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.4$  Hz, 3H), 1.28 ( $s$ , 3H), 1.28 ( $s, 2.2\text{H}$ ), 1.58 ( $m, 2\text{H}$ ), 2.24-2.38 ( $m, 2\text{H}$ ), 3.97-4.21 ( $m, 4\text{H}$ ), 4.38 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.74, 26.05, 30.16, 30.36, 30.48, 30.57, 30.76, 33.08, 35.11, 61.36, 63.70, 63.90, 66.24, 67.77, 70.22, 77.33, 77.40, 77.51, 175.63; MS  $m/z$  475 (M+H); IR (NaCl Neat) 3380, 1728, 1216, 1031  $\text{cm}^{-1}$ .

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Compound 87: 1,2-(3-Hexadecanoyloxypropane)-bis(dihydrogen phosphate)

To a solution of 79 (561 mg, 0.659 mmol) in EtOH (15 ml) was added 10 %Pd/C (610 mg). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 300 mg (92 %) of 87 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.4$  Hz, 3H), 1.28 ( $s$ , 24H), 1.56-1.63 ( $m, 2\text{H}$ ), 2.24-2.38 ( $m, 2\text{H}$ ), 3.95-4.40 ( $m, 4\text{H}$ ), 4.39 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.43, 23.73, 25.89, 26.05, 26.09, 30.15, 30.23, 30.36, 30.44, 30.47, 30.56, 30.61, 30.67, 30.75, 33.07, 34.06, 34.94, 35.11, 61.36, 64.00, 66.22, 67.74, 70.22, 77.33, 77.40, 77.51, 175.03; MS  $m/z$  489 (M+H); IR (NaCl Neat) 3357, 1729, 1216, 1029  $\text{cm}^{-1}$ .

30 Compound 88: 1,2-(3-Heptadecanoyloxypropane)-bis(dihydrogen phosphate)

To a solution of 80 (636 mg, 0.736 mmol) in EtOH (15 ml) was added 10 %Pd/C (724 mg). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 365 mg (98 %) of 88 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.4$  Hz, 3H), 1.28 ( $s$ , 26H), 1.56-1.63 ( $m, 2\text{H}$ ), 3.96-4.17 ( $m, 4\text{H}$ ), 4.22-4.42 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.74, 25.90, 26.06, 30.16, 30.24, 30.36, 30.76, 33.08, 34.81, 35.12, 63.94, 66.23, 175.03; MS  $m/z$  545 (M+H); IR (NaCl Neat) 3418, 1735, 1216, 1019  $\text{cm}^{-1}$ .

10

Compound 89: 1,2-(3-Octadecanoyloxypropane)-bis(dihydrogen phosphate)

To a solution of 81 (530 mg, 0.603 mmol) in EtOH (15 ml) was added 10 %Pd/C (617 mg). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 305 mg (97 %) of 89 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.3$  Hz, 3H), 1.28 ( $s$ , 28H), 1.56-1.61 ( $m, 2\text{H}$ ), 2.42-2.38 ( $m, 2\text{H}$ ), 3.91-4.17 ( $m, 4\text{H}$ ), 4.24-4.42 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.43, 23.74, 25.90, 26.06, 26.10, 30.16, 30.24, 30.36, 30.47, 30.57, 30.61, 30.67, 30.76, 33.08, 34.81, 34.95, 35.11, 61.37, 63.72, 66.26, 67.68, 67.75, 70.25, 77.48, 175.04; MS  $m/z$  517 (M+H); IR (NaCl Neat) 3388, 1731, 1216, 1020  $\text{cm}^{-1}$ .

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Compound 90: 1,2-(3-Nonadecanoyloxypropane)-bis(dihydrogen phosphate)

To a solution of 82 (732 mg, 1.05 mmol) in EtOH (25 ml) was added 10 %Pd/C (1.00 gm). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 555 mg (98 %) of 90 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.4$  Hz, 3H), 1.27 ( $s$ , 29H), 1.56-1.63 ( $m, 2\text{H}$ ), 2.42-2.38 ( $m, 2\text{H}$ ), 4.06-4.17 ( $m, 2\text{H}$ ), 4.22-4.42 ( $m, 2\text{H}$ ), 4.59 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.74, 25.90, 26.06, 30.16, 30.24, 30.36, 30.48, 30.57, 30.63, 30.76, 30.79, 33.08, 34.81, 35.12, 63.94, 66.23, 175.03; MS  $m/z$  531 (M+H); IR (NaCl Neat) 1735, 1216, 1012  $\text{cm}^{-1}$ .

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Compound 91: 1,2-(3-Decanoyloxypropane)-bis(dihydrogen phosphate)

To a solution of 83 (711 mg, 0.784 mmol) in EtOH (25 ml) was added 10 %Pd/C (813 mg). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 419 mg (97 %) of 91 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.4$  Hz, 3H), 1.28 ( $s$ , 32H), 1.58 ( $m, 2\text{H}$ ), 2.24-2.38 ( $m, 2\text{H}$ ), 3.95-4.42 ( $m, 4\text{H}$ ), 4.58 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.74, 25.90, 26.06, 30.16, 30.24, 30.36, 30.48, 30.57, 30.63, 30.67, 30.76, 33.08, 34.81, 35.11, 61.37, 61.98, 66.26, 67.69, 67.77, 77.42, 175.03; MS  $m/z$  545 (M+H); IR (NaCl Neat) 3418, 1735, 1216, 1019  $\text{cm}^{-1}$ .

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Compound 92: 1,2-(3-Eicosanoyloxypropane)-bis(dihydrogen phosphate)

To a solution of 84 (711 mg, 0.784 mmol) in EtOH (25 ml) was added 10 %Pd/C (813 mg). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 419 mg (97 %) of 92 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.4$  Hz, 3H), 1.28 ( $s$ , 32H), 1.58 ( $m, 2\text{H}$ ), 2.24-2.38 ( $m, 2\text{H}$ ), 3.95-4.42 ( $m, 4\text{H}$ ), 4.58 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.74, 25.90, 26.06, 30.16, 30.24, 30.36, 30.48, 30.57, 30.63, 30.67, 30.76, 33.08, 34.81, 35.11, 61.37, 61.98, 66.26, 67.69, 67.77, 77.42, 175.03; MS  $m/z$  545 (M+H); IR (NaCl Neat) 3418, 1735, 1216, 1019  $\text{cm}^{-1}$ .

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Compound 93: 1,2-(3-Hexadecanoyloxypropane)-bis(dihydrogen phosphate)

To a solution of 85 (711 mg, 0.784 mmol) in EtOH (25 ml) was added 10 %Pd/C (813 mg). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 419 mg (97 %) of 93 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.4$  Hz, 3H), 1.28 ( $s$ , 32H), 1.58 ( $m, 2\text{H}$ ), 2.24-2.38 ( $m, 2\text{H}$ ), 3.95-4.42 ( $m, 4\text{H}$ ), 4.58 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.74, 25.90, 26.06, 30.16, 30.24, 30.36, 30.48, 30.57, 30.63, 30.67, 30.76, 33.08, 34.81, 35.11, 61.37, 61.98, 66.26, 67.69, 67.77, 77.42, 175.03; MS  $m/z$  545 (M+H); IR (NaCl Neat) 3418, 1735, 1216, 1019  $\text{cm}^{-1}$ .

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Compound 94: 1,2-(3-Pentadecanoyloxypropane)-bis(dihydrogen phosphate)

To a solution of 86 (711 mg, 0.784 mmol) in EtOH (25 ml) was added 10 %Pd/C (813 mg). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours, TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 419 mg (97 %) of 94 as a white wax:  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 ( $t, J=6.4$  Hz, 3H), 1.28 ( $s$ , 32H), 1.58 ( $m, 2\text{H}$ ), 2.24-2.38 ( $m, 2\text{H}$ ), 3.95-4.42 ( $m, 4\text{H}$ ), 4.58 ( $m, 1\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ )  $\delta$  14.44, 23.74, 25.90, 26.06, 30.16, 30.24, 30.36, 30.48, 30.57, 30.63, 30.67, 30.76, 33.08, 34.81, 35.11, 61.37, 61.98, 66.26, 67.69, 67.77, 77.42, 175.03; MS  $m/z$  545 (M+H); IR (NaCl Neat) 3418, 1735, 1216, 1019  $\text{cm}^{-1}$ .

**Compound 92: 1,2-(3-Decanoyloxypropane)-bis(dihydrogen phosphate)**

To a solution of 84 (653 mg, 0.709 mmol) in EtOH (25 ml) was added 10 % Pd/C (710 mg). Hydrogenation was carried out for 4 hrs at 60 psi. After 4 hours TLC determined the completion of the reaction, the reaction mixture was filtered through celite, and the eluate was concentrated under reduced pressure to afford 400 mg (98 %) of 92 as a white wax:  $^1\text{H}$  NMR (CD<sub>3</sub>OD) δ 0.89 (t,  $J$ =6.3 Hz, 3H), 1.27 (s, 36H), 1.58 (m, 2H), 2.24-2.38 (m, 2H), 3.98-4.42 (m, 4H), 4.59 (m, 1H);  $^{13}\text{C}$  NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD) δ 13.72, 22.40, 24.71, 28.84, 28.97, 29.08, 29.18, 29.41, 31.65, 34.1660.15, 60.99, 62.42, 63.17, 65.16, 65.30, 65.98, 73.24, 173.79; MS m/z 573 (M<sup>+</sup>); IR (NaCl Neat) 3431, 1739, 1254, 1177 cm<sup>-1</sup>.

**Example 10 - *Xenopus* Oocyte Assay**

*Xenopus* oocytes which endogenously express PSP24/PLGFR were used to screen the newly designed and synthesized compounds for their LPA inhibitory activity.

Oocytes were obtained from xylazine-anesthetized adult *Xenopus laevis* frogs (Carolina Scientific, Burlington, NC) under aseptic conditions and prepared for experiment. Stage V-VI oocytes were denuded of the follicular cell layer with type A collagenase treatment (Boehringer, D) at 1.4 mg/ml in a Ca<sup>2+</sup>-free ovarian Ringers-2 solution ((OR-2) 82.5 mM NaCl, 2 mM KCl, 1 mM MgCl<sub>2</sub>, 5 mM HEPES, pH 7.5, with NaOH). Oocytes were kept in Barth's solution in an incubator between 17-20 °C and were used for 2-7 days after isolation.

Electrophysiological recordings were carried out using a standard two-electrode voltage-clamp amplifier holding the membrane potential at -60 mV (GeneClamp 500, Axon Instruments, CA). Test compounds were dissolved in MeOH, complexed with fatty acid free BSA, and diluted with frog Na<sup>+</sup>-Ringers solution (120 mM NaCl, 2 mM KCl, 1.8 mM CaCl<sub>2</sub>, 5 mM HEPES; pH 7.0), which were applied through superfusion to the oocyte at a flow rate of 5 ml/min. Membrane currents were recorded with a NIC-310 digital oscilloscope (Nicolet, Madison, WI). Applications were made at intervals of 15 mins (minimum) to allow for the appropriate washout and recovery from desensitization.

Figures 21-27 show the dose-dependent inhibition of LPA-induced chloride currents by compounds 56, 57, 66, and 92. Compound 36 was the best inhibitor among the non-phosphorylated derivatives. When compound 36 was injected intracellularly to see whether its

inhibitory effects were a result of its actions on the cell surface or whether the inhibition was a result of its actions within the cell, this intracellular application of 36 did not give any information as to its site of action. Hence, moving away from free hydroxy compounds (35-43), phosphorylated compounds (55-59) were synthesized to interact on the cell surface and to prevent the compounds from penetrating into the cell.

Compounds 56, 57, 66, and 92 were inhibitors of LPA-induced chloride current in *Xenopus* oocyte. Compounds 56, 57, 66, and 92 were able to block the actions of LPA in a dose-dependent fashion. Moreover, washing the the *Xenopus* oocyte, there was a complete recovery of the LPA response; that experiment implies that compounds 56, 57, 66, 92 were able to inhibit the LPA-induced chloride currents in a reversible fashion. Compound 66 at 5  $\mu\text{M}$  completely abolished the effect of LPA in *Xenopus* oocytes, with an IC<sub>50</sub> of about 1.2  $\mu\text{M}$  (Figures 23 and 24). Moreover, when 66 was microinjected inside the cell (arrow, Figure 23B), followed by the extracellular application of LPA (10 nM), it failed to inhibit the LPA response; that experiment suggests that the inhibitory actions of compound 66 were of an extracellular nature.

Compounds 35, and 37-43 were tested on *Xenopus* oocytes, but the results were inconclusive. Compound 55 at 1  $\mu\text{M}$  showed slight inhibition (38% against 2 nM LPA). In the SAP series, compounds 58 and 59 remain to be tested in the *Xenopus* oocyte assay. In the bisphosphate series, compound 89 inhibited the LPA-induced response (59 % against 2 nM LPA). However, compounds 67 (threshold ~ 1  $\mu\text{M}$ ), 68 (threshold ~ 10 nM), and 85 (threshold ~ 100 nM) were able to elicit a response alone; compounds 86, 87, 88, 90, and 91 have yet to be evaluated. Compound 56a was designed and synthesized to test the importance of the free amino group. When 56a was evaluated in the *Xenopus* oocyte assay, 56a enhanced the LPA response when applied in combination with LPA. Compound 56a did not elicit a response at 2  $\mu\text{M}$  (not shown), but at 10  $\mu\text{M}$ , 56a was able to elicit a response on its own (Figure 26); that experiment suggests, that a free amino group is necessary for the inhibitory activity.

**Example 11 - HEY Ovarian Cells Migrations**

It is known that two LPA receptors, EDG-2 and EDG-7, are expressed in HEY ovarian cancer cells, so compounds 56, 56a, and 66 were evaluated for their ability to inhibit LPA-induced cell motility (compound conc: 1  $\mu\text{M}$  against 0.1  $\mu\text{M}$  LPA conc.).

35 in HEY ovarian cancer cells, so compounds 56, 57, 66, and 66 were evaluated for their ability to inhibit LPA-induced cell motility (compound conc: 1  $\mu\text{M}$  against 0.1  $\mu\text{M}$  LPA conc.).

HEY ovarian cells were maintained in RPMI 1640 medium with 2 mM L-glutamine (GIBCO BRL, supplemented with 10% fetal bovine serum (FBS, Hytclone). All cells were synchronized to the G<sub>0</sub>/G<sub>1</sub> stage by growing them to confluence for 2 days. The cells were replated and harvested for experiments when cells were about 50–60% confluent on the flask. After removal of the cells from the flask, they were exposed for 5 min to 0.53 mM EDTA in PBS at 37°C. EDTA was neutralized with equal volume of RPMI 1640 plus 2 mM L-glutamine and 10% FBS. Cells were centrifuged at 800 rpm for 10 min at room temperature. Harvested cells were washed twice with RPMI 1640 with 2 mM L-glutamine medium and resuspended in the concentration of 1 x 10<sup>6</sup> cells/ml, and then rested for 1 hr at 37°C.

A modified quantitative cell migration assay (Cat. # ECM500 from Chemicon, Temecula, CA) was used to test cell motility. The Chemicon chamber membrane was coated with fibronectin-containing pores of 8 microns in diameter. A 400 µl RPMI/2 mM L-glutamine containing either no inhibitors or inhibitors (1 µM) were pipetted into the lower chamber. About 5 x 10<sup>4</sup> cells in RPMI 1640/2 mM L-glutamine were added to the top chamber. The 24-well plates with inserts were incubated for 4 hours in a 5% CO<sub>2</sub> incubator at 37°C. At the end of incubation, the chambers were removed to a fresh 24-well plate, and the cells on the inside chamber were removed by a swab several times and placed in the prepared Cell Stain Solution for 30 minutes at room temperature. At the end of incubation, Cell Stain Solution was removed from the wells. The chambers were washed 3 times with 1 mL PBS per well. After the final PBS wash, the chambers were examined to confirm proper cell morphology, and adherent cells were counted using an inverted microscope.

An effect of the newly synthesized compounds on the LPA-induced migration of HEY ovarian cancer cell is shown in Figure 27. Compound 55 inhibited the LPA-induced cell motility by about 70%; however, compound 56 (marginally) and 56a potentiated the LPA-induced cell motility.

#### Example 12 - Compound Cytotoxicity

In et al (2000) and RT-PCR data showed the presence of PLGFR's in

prostate cancer cell lines DU-145, PC-3, and LNCaP. Due to the promising inhibitory activity in *Xenopus* oocyte and the cell motility assay, the growth inhibitory effects of a number of compounds on DU-145, PC-3, and LNCaP prostate cancer cell lines were examined.

DU-145, PC-3, and LNCaP cells were propagated in 150 cm<sup>2</sup> flasks, containing RPMI-1640 or Dulbecco's modified Eagle media supplemented with 10%

fetal bovine serum (FBS). Cells were removed from stock flasks using trypsin, centrifuged, resuspended in fresh media, and plated at a density of approximately 2,000 cells/well in 96-well culture plates. Final drug concentrations ranged from 0.05 to either 10 or 50 µM. Control experiments with no drug added (negative control) and 5-fluorouracil added (positive control) were performed in parallel. Media was removed and replaced at 48 hours to minimize the effects of drug degradation during the course of the experiment. After 96 hours drug exposure, cells were fixed by the addition of cold 50% trichloroacetic acid (TCA) and incubation at 4°C for 1 hour. Fixed cells were stained with sulforhodamine B (SRB), and cell number was determined by comparison of absorbance at 540 nm, as compared to a standard curve of cell number versus absorbance. Experiments were performed in duplicate. Cell number as a percentage of control (untreated wells) was plotted versus drug concentration and the concentration that inhibited cell growth by 50% (IC<sub>50</sub>) determined by nonlinear regression (WinNonlin, Pharsight Corporation).

Cytotoxicity studies performed on prostate cancer cell lines DU-145, PC-3, and LNCaP, together with the reference compounds 5F-uracil, LPA (1:8), SFH (1:3), SPP (1:3), and N-palmitoyl L-serine phosphoric acid (1:5), are shown in Table 3 below.

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Table 3: Cytotoxicity of Synthesized Compounds on Prostate Cancer Cell Lines

Compound	DU145	PC-3	LNCaP	$IC_{50} \pm SEM$ ( $\mu M$ ) <sup>a</sup>
Fluorouracil	6.8±3.3	10.2±4.1	2.8±1.6	
LPA (1:8:1)	WA	28.5±6.3	WA	
SPP (1:3:0)	>10	WA	NA	
SPH (1:3:0)	13.9±1.1	11.7±2.3	5.7±2.1	
N-palmitoyl-L-serine (15:0)	WA	WA	WA	
27	19.7±6.0	WA	10.9±2.7	
38	38.9±8.9	?	?	
51	8.1±1.3	25.4±3.6	19.9±6.4	
55	24.9±4.1	31.6±9.0	4.9±2.6	
56	2.3±1.2	0.7±0.1	13.5±4.7	
56a	0.7±0.1	WA	30.3±7.9	
57	9.1±0.8	WA	10.7±2.1	
66	NA	NA	3.1±3.2	
67	WA	WA	25.2±12.3	
68	WA	WA	29.3±21.7	
85	NA	NA	11.6±10.3	
86	NA	NA	?	
87	NA	NA	WA	
88	NA	NA	?	
89	WA	NA	?	
90	>50	WA	WA	
91	4.2±1.9	WA	WA	
92	WA	WA	WA	

<sup>a</sup>Cell number as a percentage of control (untreated wells) was plotted versus drug concentration and the concentration that inhibited cell growth by 50% ( $IC_{50}$ ) determined by nonlinear regression (Waintonin, Pharmatight Corporation).  
WA = Weak Activity; NA = No Activity; ? = Maximum inhibition was 50%.

## Discussion of Examples 1-12

Three sets of compounds were specifically synthesized and analyzed (35-43, 55-59, 66-68, and 85-92). The first and the second sets involve the amalgamation of the endogenous inhibitors SPP and SPH with the synthetic inhibitor N-palmitoyl L-serine phosphoric acid, whereas the third series involves the bisphosphates. Compounds 56, 57, 66 and 92 were inhibitors of LPA-induced chloride currents in the *Xenopus* oocyte assay. Also, bisphosphates with shorter chain length at (*sn*-1) position were able to elicit chloride currents in *Xenopus* oocyte [67 (threshold ~ 1  $\mu M$ ), 68 (threshold ~ 10  $\mu M$ ), 69 (threshold ~ 100  $\mu M$ )].

Compound 66 was shown to inhibit the LPA-induced cell motility in HEY ovarian cancer cell lines. On evaluating the growth inhibitory effects of the above-synthesized compounds on DU-145, PC-3, and LNCaP prostate cancer cell lines, three highly potent and selective compounds (56, 56a, and 66) were discovered. The above data (Table 3) suggests that (i) compounds that contain an alcohol with no phosphate are less active (27 vs. 56), (ii) compounds with the protected phosphate moiety are less active (51 vs. 56), (iii) alkylation of the amine does not reduce activity (56a), (iv) the most potent bisphosphate has an ether linkage at the *sn*-1 position, (v) decreasing the chain length in the SAP series (55 vs. 56) decreased the potency towards DU-145 and PC-3 (however, it was more potent against LNCaP cells), (vi) on decreasing the chain length for the bisphosphate (*sn*-1 acyl) compounds, potency decreased, though selectivity towards LNCaP cell remained, and (vii) substitution at *sn*-1 position (facyl vs. acyl) did not increase the potency. The target site for these molecules is likely on the cell membrane (e.g., a membrane-spanning receptor), because the polar phosphate derivatives are unlikely to easily cross the cell membrane (although there exists the possibility that an active transport system could exist). These results suggest that differences in PLGFR's or downstream signal transduction events may play a significant role in the growth inhibitory properties of these compounds in prostate cancer cells.

## Example 13 – Preparation and Characterization of Stable Cell Lines Expressing Edg-2, Edg-4, and Edg-7

In an effort to develop selective antagonists to the Edg-2, -4, and -7 receptors, a system for screening potential compounds was first established. RHT7777 cells were chosen as a model system since they have been reported to be non-responsive to LPA in a variety of cellular assays and were found to be devoid of mRNA for any of the known Edg receptors (Fukushima et al., 1998). Stable cells lines

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Compounds 55, 56, 56a, 66, and 85 exhibited a range of growth inhibitory activities. Compound 56 was a more potent inhibitor of DU-145 and PC-3 cell growth than 5-fluorouracil. Interestingly, 56a selectively inhibited DU-145 cell growth, but was less potent against PC-3 cells; compound 55 was a more potent inhibitor of LNCaP cell growth as against DU-145 and PC-3 cells. Compound 66 selectively inhibited LNCaP cell growth, but showed no activity on PC-3 and LNCaP cells. Compound 85 was the most active among the bisphosphates (*sn*-1 acyl).

transfected with the EDG receptors, as well as control cell lines transfected with empty vector, were established in RH7777 cells.

The resulting clones were screened by monitoring intracellular  $\text{Ca}^{2+}$  transients, and by RT-PCR. This screening process led to the identification of at least three positive cell lines expressing Edg-2 and -7, while no positive cell lines expressing Edg-4 could be identified. Vector transfected cells were also found to be non-responsive to LPA. Although stable clones expressing Edg-4 were not isolated, the transient expression of Edg-4 resulted in the LPA-mediated activation of intracellular  $\text{Ca}^{2+}$  transients, demonstrating that the construct was functionally active in these cells. The stable Edg-4 cell line used in these experiments was isolated and characterized by Im et al., who kindly provided us with the same clone (Im et al., 2000).

The cell lines were further characterized in an effort to identify a suitable assay for screening potential antagonists. LPA-elicited activation of ERK 1/2 was seen in Edg-2 and transient Edg-4 expressing cells, whereas ERK 1/2 was not activated in Edg-7 expressing cells. LPA elicited  $\text{Ca}^{2+}$  transients in all stable cell lines expressing Edg-2, -4, and -7. Dose response curves revealed  $\text{EC}_{50}$  values of  $3.78 \pm 53$ ,  $9.98 \pm 67$ , and  $21.4 \pm 26 \text{ nM}$  for Edg-2, -4, -7 expressing cells, respectively (Figures 28A-C). Because the  $\text{EC}_{50}$  value determined in the stable Edg-4 clone was different from that previously reported, a dose response curve was also established for cells transiently expressing Edg-4 (Figure 28B, An et al., 1998a; An et al., 1998b), which yielded an  $\text{EC}_{50}$  value of  $8.6 \pm 39$ .

The ability of LPA to stimulate DNA synthesis in the stable cell lines was examined by measuring the incorporation of  $^3\text{H}$ -thymidine. Neither wild type, nor the vector transfected RH7777 cells showed an increase in  $^3\text{H}$ -thymidine incorporation following a 24 hr incubation with  $10 \mu\text{M}$  LPA, which is in contrast to a previous report that LPA is mitogenic in these cells. Edg-2 expressing cells showed a 1.8-fold increase in  $^3\text{H}$ -thymidine incorporation, whereas Edg-4 and -7 expressing cells did not show an increase in  $^3\text{H}$ -thymidine incorporation, as compared to control cells.

**Example 14 – Short Chain Phosphatidates Activity on Edg-2 and Edg-7 Receptors**

Since  $\text{Ca}^{2+}$  transients were elicited in all three stable cell lines expressing Edg-2, -4, and -7 (Figures 28A-C), this assay was used for screening potential antagonists. In an effort to identify selective antagonists for the LPA activated members of the Edg receptor family, Edg-2, -4, and -7, the structural features of the LPA pharmacophore were relied upon as a starting point. Short-chain (8:0)

LPA or a mixture of LPA (8:0) and LPA (18:1) were tested as inhibitors of Edg-2, -4, or -7. When the cells were challenged with the mixture of LPA 8:0 and LPA 18:1,  $\text{Ca}^{2+}$  responses were not effected in any of the three stable cell lines (see Figures 30A-C, 31A-C, and 32A-B). LPA 8:0, alone, was unable to elicit  $\text{Ca}^{2+}$  responses in any of the cells, at concentrations as high as  $10 \mu\text{M}$ .

Based on these results, applicants hypothesized that a modification of the LPA pharmacophore, which sterically restricted the mobility of the fatty acid chain, might also effect its ligand properties. For this reason, we tested compounds with a second short-chain fatty acid at the *sn*-2 position were also tested. Such short-chain phosphatidates have increased hydrophobicity over the corresponding short-chain LPA, which could exert constraints on their interaction with the ligand-binding pocket of the receptor.

Phosphatidic acid (PA) and diacylglycerol pyrophosphate (DGPP) are naturally occurring lipids which share some key chemical properties with the LPA pharmacophore, having an ionic phosphate group(s) and fatty acid chains. Neither is an agonist of the Edg receptors (see below). With this similarity in mind, short-chain DGPP were prepared and tested as an inhibitor of Edg-2, -4, or -7. Figures 29A-D show the effect of a 10-fold excess of DGPP (8:0) on the  $\text{Ca}^{2+}$  responses elicited by LPA in the stable cell lines. The  $\text{Ca}^{2+}$  responses in Edg-2 expressing cells were inhibited by approximately 50% (Figure 29A), whereas the responses in Edg-7 expressing cells were completely abolished (Figure 29C). In contrast,  $\text{Ca}^{2+}$  responses in Edg-4 expressing cells were unaffected by DGPP 8:0 (Figure 29B). Because of the discrepancy in  $\text{EC}_{50}$  values for the stable and transient expression of Edg-4 (Figure 29B), DGPP 8:0 was similarly tested on cells that were transiently transfected with Edg-4. Consistent with results from experiments in stable cells,  $\text{Ca}^{2+}$  responses were not effected by DGPP 8:0 in cells transiently expressing Edg-4 (Figure 29D). Similar observations were obtained with PA 8:0 in each of the assays described above for DGPP 8:0 (see below).

Inhibition curves were determined in cells expressing Edg-2 and -7, using increasing concentrations of DGPP 8:0, while the concentration of LPA was kept constant at the  $\text{EC}_{50}$  relative to the receptor studied.  $\text{IC}_{50}$  values of  $2.65 \pm 2.8 \text{ nM}$  for Edg-7 (Figure 30A) and  $11.0 \pm 0.68 \mu\text{M}$  for Edg-2 (Figure 31A) were determined from the curves. Using a constant amount of DGPP 8:0 near to the  $\text{IC}_{50}$  value (250 nM for Edg-7, 3  $\mu\text{M}$  for Edg-2), the dose response curves for both Edg-7 (Figure 30B) and Edg-2 (Figure 31B) were shifted to the right, indicating a competitive mechanism of inhibition.

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In order to better define the structure activity relationship for DGPP, short- (8:0) and long-chain (18:1) species of LPA, DGPP, PA, and DAG were tested on Edg-2 and -7 expressing cell lines. Figure 30C shows the effect of these lipids on the  $Ca^{2+}$  responses in Edg-7 expressing cells when exposed to a combination of LPA 18:1 and each of these lipids. For these experiments, the concentration of LPA was chosen to be near the EC<sub>50</sub>, whereas test lipids were applied at a concentration equal to the IC<sub>50</sub> of DGPP 8:0. LPA 8:0 had no effect on Edg-7, whereas both DGPP 8:0 and PA 8:0 significantly inhibited the  $Ca^{2+}$  responses by 50 and 56%, respectively. In contrast DAG 8:0 significantly increased the  $Ca^{2+}$  responses. When the chain length of DGPP and PA was increased to 18:1, these analogs were no longer inhibitors of Edg-7 (Figure 30C). DAG 18:1, likewise, did not have an inhibitory effect on Edg-7.

The same set of lipids was tested on Edg-2 expressing cells (Figure 31C). Octyl chain length analogs of DGPP, PA, and DAG, when used at 10  $\mu$ M, all decreased the responses to 50, 19, and 64% of control, respectively. When the chain length was increased to 18:1, DGPP and DAG no longer had an inhibitory effect, whereas PA 18:1 maintained a modest inhibitory effect, decreasing the  $Ca^{2+}$  response by 18%. The panel of lipids was also tested on Edg-4 expressing cells (Figures 32A-B). When these lipids were assayed in the stable cell line expressing Edg-4, none of the short- or long-chain lipids had an inhibitory effect, whereas both PA 8:0 and 18:1 significantly increased the  $Ca^{2+}$  responses, to 162 and 137% of control, respectively. To confirm the results obtained from the stable alone, the lipid panel was tested on cells transiently expressing Edg-4 (Figure 32B). Again, neither the short-, nor the long-chain species of DGPP or PA had an inhibitory effect on the  $Ca^{2+}$  response, in agreement with the results from the stable cell line. In contrast to the stable Edg-4 clone, neither PA analog enhanced the  $Ca^{2+}$  response in cells with transient expression of Edg-4. Neither species of PA when applied alone, elicited a response at concentrations up to 10  $\mu$ M, in cells stably or transiently expressing Edg-4.

The effect of DGPP 8:0 on cells that endogenously express LPA receptors was also examined. DGPP 8:0 was found to inhibit the  $Ca^{2+}$ -mediated, inward Cl<sup>-</sup> currents elicited by LPA in *Xenopus* oocytes with an IC<sub>50</sub> of 96  $\pm$  21 nM (Figure 33A). In the presence of a 200 nM concentration of DGPP 8:0, the dose response curve for LPA 18:1 was shifted to the right, indicating a competitive mechanism of action as found in Edg-2 and -7 clones (Figure 33B). To examine whether DGPP 8:0 acts through an intracellular or extracellular mechanism, DGPP 8:0 was injected intracellularly and the oocyte was exposed to LPA 18:1. Figure 32C shows that following the intracellular injection of DGPP 8:0, estimated to reach a concentration > 300 nM, the extracellular application of 5 nM LPA 18:1 elicited a

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response equal in size to that of the control. In comparison, the response normally elicited by LPA 18:1 was completely inhibited when DGPP 8:0 was applied extracellularly (Figure 33C). The inhibitory effect of DGPP 8:0 was reversible, as after a 10-min washing the response recovered to control level (Figure 33C).

To show the specificity of DGPP 8:0 for the LPA receptors expressed in the oocyte, the expression of neurotransmitter receptors was induced by the injection of polyA+ mRNA from rat brain. This resulted in the expression of the G-protein coupled receptors for serotonin and acetylcholine, which are not expressed in non-injected oocytes. These neurotransmitters activate the same inositol triphosphate- $Ca^{2+}$  signaling pathway that is activated by LPA (Tigyi et al., 1990). In these oocytes, DGPP 8:0 did not inhibit either serotonin- or carbachol-elicited responses, demonstrating the specificity of DGPP 8:0 for the LPA receptors. PA 8:0 when used at similar concentrations was also effective at inhibiting the LPA-elicited responses in the oocytes.

The effect of DGPP 8:0 on LPA-elicited responses was also examined in mammalian systems that endogenously express LPA receptors. NIH3T3 cells were screened by RT-PCR for the presence of mRNA for the Edg and PSP24 receptors. Figure 34A shows that in NIH3T3 cells mRNA transcripts for Edg-2, -5, and PSP24 were detected. To show that DGPP 8:0 was specific in inhibiting LPA-elicited but not SIP-elicited  $Ca^{2+}$  responses, NIH3T3 cells were exposed to 100 nM LPA or SIP in the presence of 10  $\mu$ M DGPP 8:0. As shown in Figure 34B, DGPP 8:0 significantly inhibited the LPA-elicited  $Ca^{2+}$  responses, whereas the SIP-elicited response was not effected.

LPA has been shown to be generated from and play a role in ovarian cancer (Xu et al., 1995a). Therefore, DGPP 8:0 was also tested on HEY ovarian cancer cells to determine if it had an effect on a therapeutically relevant target. Figure 34D shows that DGPP 8:0 inhibited the LPA-elicited  $Ca^{2+}$  response to 12% of control, whereas DGPP 18:1 had no effect. Likewise, PA 8:0 inhibited the  $Ca^{2+}$  response to 6% of control, whereas PA 18:1 had no effect. HEY express mRNA transcripts for Edg-1, -2, -5, -7 receptors (Figure 34C).

#### Example 15 - Inhibition of NIH3T3 Cell Proliferation

The hallmark effect of a growth factor is its ability to elicit cell proliferation. Since LPA has been shown to stimulate the proliferation of a variety of different cell types (Gotoz et al., 2000), the ability of DGPP 8:0 to inhibit cell proliferation was examined in NIH3T3 cells. Figure 35 shows that DGPP 8:0

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significantly inhibited the LPA-induced proliferation of NIH3T3 cells, reducing cell number to control levels, whereas it had no effect on the solvent-treated control cells. To define the structure-activity relationship for the inhibitory effect of DGPP 8:0, the short- and long-chain species of DGPP, PA, and DAG were included in the assay. As shown in Figure 35, none of the lipids included in the test panel had a significant inhibitory or stimulatory effect on the solvent-treated control cells. Only DGPP 8:0 inhibited the LPA-induced proliferation. Neither DGPP 8:1, nor long- and short-chain PA and DAG had an effect on the LPA-induced proliferation. Interestingly, PA 8:0 had no significant inhibition in this assay.

#### Discussion of Examples 13-15

RH7777 cells were used for heterologous expression of Edg-2, -4, and -7 receptors to screen potential antagonists. Based on our previous computational modeling of the Edg receptors (Parrill et al. 2000) and the available structure-activity data (Jalink et al., 1995), the above experimental results demonstrate that the short-chain phosphatidate DGPP 8:0 is a selective, competitive antagonist of Edg-7, with an IC<sub>50</sub> value of 285 ± 28 nM. The same molecule was found to be a poor inhibitor of Edg-2, with an IC<sub>50</sub> value of 11.0 ± 0.68  $\mu$ M, whereas it did not inhibit Edg-4. DGPP 8:0 inhibited the endogenous LPA response in *Xenopus* oocytes with an IC<sub>50</sub> value of 96 ± 21  $\mu$ M. PA 8:0 showed similar inhibitory properties. Therefore, these short-chain phosphatidates show a 40-100-fold selectivity for Edg-7 over Edg-2.

The above results with short-chain phosphatidates confirm those of Bandoh et al. (2000) who demonstrated that LPA, with an acyl chain-length of twelve carbons or less, does not elicit responses in insect cells expressing Edg-2, -4, or -7. As demonstrated above, LPA 8:0 was neither an agonist nor an antagonist of Edg-2, -4, or -7 in a mammalian expression system. Edg-7 has a 10-fold preference for LPA with the fatty acid chain esterified to the *sn*-2, versus the *sn*-1 position (Bandoh et al., 2000). Therefore, the distance of the hydrocarbon chain relative to the phosphate moiety, does not abolish the binding to and activation of the receptor. Edg-7 also shows a preference for long-chain, unsaturated fatty acids over their saturated counterparts. The presence of an ether linkage or vinyl-ether side chain also decreased the EC<sub>50</sub> by two orders of magnitude (Bandoh et al., 2000). Moreover, there is an optimal hydrocarbon chain-length of 18 carbons, whereas 20 carbon analogs were weaker agonists. These pharmacological properties of Edg-7 suggest that receptor activation is dependent upon the chain length, as well as the flexibility of the side chain (ester vs. ether linkage).

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Computational modeling of the Edg-1 receptor has identified three charged residues that are required for ligand binding. One of these residues, arginine 120, which is predicted to interact with the phosphate group, is conserved in all of the members of the Edg family. The second residue, arginine 292, occurs at a position where all Edg family members except Edg-8 have a nearby cationic residue. The third residue, glutamate 121, is not conserved amongst the LPA-specific Edg receptors, with a glutamine at the corresponding site in Edg-2, -4, and -7. This glutamine residue is predicted to interact with the hydroxyl moiety of LPA. Alanine replacement of this residue has led to a loss of ligand binding and activation of the receptor, suggesting that the ionic interaction between the charged moieties of the PLGF pharmacophore and these three residues is necessary for ligand binding in Edg-1 (Parrill et al., 2000). Moreover, the interaction between the receptor and the hydrocarbon chain, itself, was not sufficient for ligand binding and activation (Parrill et al., 2000). It was hypothesized, therefore, that a combination of interactions, involving both the ionic anchor and the hydrophobic tail, are required for agonist activation. In support of this hypothesis, the above results demonstrate that the short-chain LPA 8:0 was not able to activate Edg-2, -4, or -7, underlying the importance of the interaction between the hydrophobic tail and the ligand binding pocket. As a result, applicants have designated the hydrophobic tail as the "switch" region of the PLGF pharmacophore. Because of the relative tolerance of the *sn*-1 and *sn*-2 substitution of the fatty acids by these receptors, applicants focused on short-chain phosphatidates which were believed not to be able to activate the receptors due to their truncated hydrocarbon chains. The structural mobility of the acyl chains in the phosphatidates is also limited by the adjacent fatty acid moiety. Applicants also explored the effects of a pyrophosphate moiety, which does not change the negatively charged character of the anchoring region, but rather increases the charge.

This conceptual drug design was tested on clonal cell lines expressing the Edg-2, -4, and -7 receptors. The pharmacological properties of DGPP 8:0 and PA 8:0 were found to be dramatically different between the three receptors. Both molecules were effective at inhibiting Edg-7, whereas they were more than an order of magnitude less effective on Edg-2. Neither molecule was effective on Edg-4. DGPP 8:0 was also found to be a competitive inhibitor of both Edg-2 and -7, displacing the dose-response curves to the right with a subsequent increase in the EC<sub>50</sub> values for LPA on both receptors. The lack of agonist activity of the corresponding long-chain species of PA and DGPP, highlights the constraints that prevail in the binding pocket. The importance of the ionic anchor, in docking the ligand in the binding pocket, is

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supported by the lack of inhibition by DAG 8:0, although its cellular effects are likely confounded by its intracellular actions on other molecular targets, such as PKC.

Both PA and DGPP are naturally occurring phospholipids. DGPP (8:0) was discovered in 1993 as a novel lipid in plants and is a product of the phosphorylation of PA by phosphatidate kinase (Wissinger and Behrbohm, 1993; Munnik et al., 1996). DGPP has been identified in bacteria, yeast and plants, but not in mammalian cells. Recent studies have shown that DGPP activates macrophages and stimulates prostaglandin production through the activation of cytosolic phospholipase  $A_2$ , suggesting a role for DGPP in the inflammatory response (Balboa et al., 1999; Balinde et al., 2000). These authors ruled out the possibility that these effects were mediated through LPA receptors. The above results with the long-chain DGPP and PA analogs confirmed this notion, as these compounds did not possess agonist properties in the Edg receptor expressing cell lines at concentrations up to 10  $\mu$ M.

The effect of short chain phosphatides was also examined on LPA receptors expressed endogenously in three different cell types. DGPP 8:0 and PA 8:0 were found to be effective inhibitors of LPA-elicited Cl<sup>-</sup> currents in *Xenopus* oocytes. In order to determine the site of action, DGPP 8:0 was injected into oocytes followed by an extracellular application of LPA. DGPP 8:0 was only effective at inhibiting the LPA-elicited Cl<sup>-</sup> currents when applied extracellularly, demonstrating that it exerts its antagonist effect on the cell surface. The specificity of DGPP 8:0 for LPA receptors was demonstrated in oocytes and NIH3T3 cells. In these cells, DGPP 8:0 was only effective at inhibiting the LPA-elicited Ca<sup>2+</sup> responses and not the responses elicited by SIP, acetylcholine, or serotonin.

RT-PCR analysis revealed that only Edg-2, and not Edg-4, or -7 is expressed in NIH3T3 cells. In NIH3T3 cells, DGPP 8:0, at a high 100-fold excess, only inhibited the Ca<sup>2+</sup> responses by 40%. This degree of inhibition parallels that seen in the stable cell line expressing Edg-2, where it was also a weak inhibitor. When short-chain DGPP and PA were evaluated on HEY ovarian cancer cells, at a 10-fold excess over LPA, both were effective inhibitors, whereas neither long-chain molecule had any effect. RT-PCR revealed that the predominant mRNA was for Edg-7 in HEY cells, whereas only a trace of Edg-2 mRNA was detected. This degree of inhibition parallels that seen in the stable cell line expressing Edg-7, where both DGPP 8:0 and PA 8:0 were effective inhibitors.

Both short chain phosphatides were evaluated for their ability to

was effective at inhibiting the Ca<sup>2+</sup> responses, it was not effective at inhibiting cell proliferation. These results are in agreement with a previous report that PA (12:0) did not inhibit the mitogenic effect of PA 18:1 (van Corven et al., 1992). The stability of the molecules in long-term assays is a concern, since lipid phosphatases might inactivate the antagonist. The fact that both PA and DAG failed to inhibit the proliferation suggests that DGPP 8:0 is likely to be more stable for the duration of this assay. The stability of DGPP has also been demonstrated by Balboa et al. (1999), who reported that DGPP 8:0 provides an important new tool for the field in studying, not only the Edg receptors but also other PLGF receptors. The concept of an ionic anchor and hydrophobic switch of the PLGF pharmacophore derived from computational modeling of the Edg family should assist the design and synthesis of new inhibitors.

**Example 16 -**  
**Synthesis of Straight-Chain Phosphate Intermediates 101-105**

**Compound 101: Phosphoric acid dibenzyl ester butyl ester**  
74 mg (1.00 mmol) of anhydrous n-butanol and 365 mg (5.17 mmol) of 1*H*-tetrazole were dissolved in 34 mL of anhydrous methylene chloride in a 100 mL round-bottom flask. A solution of 0.895 g (2.38 mmol) of dibenzyl-*N,N*-disopropyl phosphoramidite in 5 mL of anhydrous methylene chloride was added via a syringe under an argon atmosphere with stirring. The reaction mixture was stirred at room temperature for 2 hrs. The reaction mixture was then cooled in a isopropyl alcohol/dry ice bath at  $\sim$  38 °C. 0.815 g (3.43 mmol) of 32 % peracetic acid in 28 mL of anhydrous methylene chloride were added dropwise via an addition funnel. After the addition, the temperature of the reaction mixture was raised to  $\sim$  0 °C with an ice bath. The reaction mixture was stirred in the ice bath for 1 hr. The reaction mixture was transferred to a separatory funnel and diluted with 200 mL of methylene chloride. The organic layer was washed with 10% sodium metabisulfite (2 x 40 mL), saturated sodium bicarbonate (2 x 40 mL), water (30 mL), and brine (40 mL). The organic layer was dried with anhydrous sodium sulfate, filtered, and concentrated under vacuum to dryness. The crude product was then purified by silica gel chromatography using 1:1 hexanes/ethyl acetate as the eluent to afford 101 (309 mg which contained a slight amount of impurity from excess phosphorylating reagent) as a clear oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>) 0.88 (*t*, *J* = 7.2 Hz, 3H, CH<sub>3</sub>), 1.34 (*sextet*, *J* = 7.2 Hz, 2H, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 1.59 (quintet, *J* = 6.6 Hz, 2H, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 3.99 (*dt*, *J* = 6.6 Hz, 6.6 Hz, 2H, OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 5.02 (*d*, *J* = 1.8 Hz, 2H, OCH<sub>2</sub>Ar), 5.05 (*d*, *J* = 2.1 Hz, 2H, OCH<sub>2</sub>Ar), 7.35 (br *s*, 10H, 2 x ArH); <sup>13</sup>C NMR (CDCl<sub>3</sub>) 13.55, 18.60,

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32.16 (d,  $J_{CP} = 6.8$  Hz), 67.72 (d,  $J_{CP} = 6.1$  Hz), 69.13 (d,  $J_{CP} = 5.5$  Hz), 127.90, 128.47, 128.55, 136.00 (d,  $J_{CP} = 6.8$  Hz);  $^{31}P$  NMR ( $CDCl_3$ ) 16.84; MS (positive mode):  $[M + ^{23}Na]^+$  at  $m/z$  357.3.

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**Compound 102: Phosphoric acid dibenzyl ester octyl ester**

130 mg (1.00 mmol) of anhydrous *n*-octanol were used and a procedure analogous to that for 101 was performed. The crude product was purified by silica gel chromatography using 7:3 hexanes/ethyl acetate as the eluent to afford 102 (351 mg, 90%) as a clear oil.  $^1H$  NMR ( $CDCl_3$ ) 0.88 (t,  $J = 6.9$  Hz, 3H,  $CH_3$ ), 1.24 (br s, 10H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 1.60 (quintet,  $J = 6.9$  Hz, 2H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 3.98 (dt,  $J = 6.6$  Hz, 6.9 Hz, 2H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 5.02 (d,  $J = 2.1$  Hz, 2H,  $OCH_2Ar$ ), 5.05 (d,  $J = 2.4$  Hz, 2H,  $OCH_2Ar$ ), 7.34 (br s, 10H, 2  $\times$  ArH);  $^{13}C$  NMR ( $CDCl_3$ ) 14.09, 22.62, 25.38, 29.06, 29.14, 30.17 (d,  $J_{CP} = 6.9$  Hz), 31.75, 38.05 (d,  $J_{CP} = 6.2$  Hz), 69.12 (d,  $J_{CP} = 5.5$  Hz), 127.90, 128.47, 128.56, 135.97 (d,  $J_{CP} = 6.9$  Hz);  $^{31}P$  NMR ( $CDCl_3$ ) 16.83; MS (positive mode):  $[M + ^{23}Na]^+$  at  $m/z$  413.4.

**Compound 103: Phosphoric acid dibenzyl ester dodecyl ester**

186 mg (1.00 mmol) of anhydrous *n*-butanol were employed and a procedure analogous to that for 101 was utilized. The crude product was purified by silica gel chromatography using 7:3 hexanes/ethyl acetate as the eluent to afford 103 (361 mg, 81%) as a clear oil.  $^1H$  NMR ( $CDCl_3$ ) 0.88 (t,  $J = 7.2$  Hz, 3H,  $CH_3$ ), 1.24 (br s, 18H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 1.60 (quintet,  $J = 6.9$  Hz, 2H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 3.98 (td,  $J = 6.9$  Hz, 6.6 Hz, 2H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 5.02 (d,  $J = 2.1$  Hz, 2H,  $OCH_2Ar$ ), 5.05 (d,  $J = 2.1$  Hz, 2H,  $OCH_2Ar$ ), 7.34 (br s, 10H, 2  $\times$  ArH);  $^{13}C$  NMR ( $CDCl_3$ ) 14.13, 22.70, 25.39, 29.12, 29.37, 29.50, 29.57, 29.66, 30.18 (d,  $J_{CP} = 7.0$  Hz), 31.92, 68.05 (d,  $J_{CP} = 6.1$  Hz), 69.12 (d,  $J_{CP} = 5.4$  Hz), 127.89, 128.46, 128.55, 135.97 (d,  $J_{CP} = 6.8$  Hz);  $^{31}P$  NMR ( $CDCl_3$ ) 16.84; MS (positive mode):  $[M + ^{23}Na]^+$  at  $m/z$  469.1.

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**Example 17 - Synthesis of Straight-Chain Phosphate Compounds 106-110**

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**Compound 106: Phosphoric acid monobutyl ester**

327 mg (1.00 mmol) of dodecanol were employed and an analogous procedure to that for 101 was used. The crude product was purified by silica gel chromatography using 7:3 hexanes/ethyl acetate as the eluent to afford 105 (516 mg, 88%) as a hygroscopic white solid:  $mp$  43.5-44.5 °C;  $^1H$  NMR ( $CDCl_3$ ) 0.88 (t,  $J = 6.9$  Hz, 3H,  $CH_3$ ), 1.25 (br s, 38H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 1.60 (quintet,  $J = 6.9$  Hz, 2H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 3.98 (td,  $J = 6.6$  Hz, 6.6 Hz, 2H,  $OCH_2Ar$ ), 7.35 (br s, 10H, 2  $\times$  ArH);  $^{13}C$  NMR ( $CDCl_3$ ) 14.13, 22.70, 25.39, 29.12, 29.37, 29.50, 29.57, 29.66, 29.71, 30.18 (d,  $J_{CP} = 6.9$  Hz), 31.93, 68.06 (d,  $J_{CP} = 6.0$  Hz), 69.13 (d,  $J_{CP} = 5.6$  Hz), 127.89, 128.47, 128.55, 135.98 (d,  $J_{CP} = 6.9$  Hz);  $^{31}P$  NMR ( $CDCl_3$ ) 16.83; MS (positive mode):  $[M + ^{23}Na]^+$  at  $m/z$  609.3.

**Compound 106: Phosphoric acid monobutyl ester**

200 mg (0.60 mmol) of 101 were dissolved in 30 mL of anhydrous methanol in a thick-walled pressure vessel. The vessel was purged with argon and ~200 mg of 10% Pd/C was added. The vessel was connected to a hydrogenation apparatus and a hydrogen atmosphere of ~50 psi was maintained inside the reaction vessel at room temperature for 8 hrs. The reaction mixture was then filtered by vacuum through a pad of celite which was washed with methanol. The solvent was evaporated under vacuum leaving behind 70 mg (86%) of a yellow oil 106.  $^1H$  NMR ( $CDCl_3/MeOH-d_4$ ) 0.95 (t,  $J = 7.2$  Hz, 3H,  $CH_3$ ), 1.43 (sextet,  $J = 7.5$  Hz, 2H,  $OCH_2CH_2CH_2CH_2CH_3$ ), 1.66 (quintet,  $J = 6.9$  Hz, 2H,  $OCH_2CH_2CH_2CH_2CH_3$ ),  $^{13}C$  NMR ( $CDCl_3/MeOH-d_4$ ) 13.71, 19.02, 32.72 (d,  $J_{CP} = 7.2$  Hz), 66.86 (d,  $J_{CP} = 5.5$  Hz);  $^{31}P$  NMR ( $CDCl_3/MeOH-d_4$ ) 18.84; MS (negative mode):  $[M - 1]^-$  at  $m/z$  153.0.

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**Compound 107: Phosphoric acid monoacetyl ester**

200 mg (0.51 mmol) of 102 were employed and using a procedure analogous to that for 106, 100 mg (93%) of a white/yellow tacky solid 107 was as for 101 was employed. The crude product was purified by silica gel chromatography using 7:3 hexanes/ethyl acetate as the eluent to afford 104 (474 mg, 89%) as a hygroscopic white solid:  $mp$  32-33 °C;  $^1H$  NMR ( $CDCl_3$ ) 0.88 (t,  $J = 6.9$  Hz, 3H,  $CH_3$ ), 1.25 (br s, 30H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 1.60 (quintet,  $J = 6.9$  Hz, 2H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 3.98 (td,  $J = 6.6$  Hz, 6.9 Hz, 2H,  $OCH_2CH_2(CH_2)_5CH_3$ ), 5.02 (d,  $J = 2.1$  Hz, 2H,  $OCH_2Ar$ ), 5.05 (d,  $J = 2.1$  Hz, 2H,  $OCH_2Ar$ ), 7.34 (br s, 10H, 2  $\times$  ArH);  $^{13}C$  NMR ( $CDCl_3$ ) 14.12, 22.70, 25.40, 29.13, 29.38, 29.51, 29.53, 29.65, 127.90, 128.47, 128.55, 136.00 (d,  $J_{CP} = 6.8$  Hz);  $^{31}P$  NMR ( $CDCl_3$ ) 16.83; MS (positive mode):  $[M + ^{23}Na]^+$  at  $m/z$  553.3.

isolated.  $^1\text{H}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 0.89 (t,  $J = 6.9$  Hz, 3H,  $\text{CH}_3$ ), 1.29 (br s, 5H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ ), 1.67 (quintet,  $J = 6.9$  Hz, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ ), 3.97 (dt,  $J = 6.6$  Hz, 6.6 Hz, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 14.18, 22.98, 25.89, 29.57, 29.58, 30.76 (d,  $J_{\text{CP}} = 7.3$  Hz), 32.18, 67.16 (d,  $J_{\text{CP}} = 5.2$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 20.55; MS (negative mode): [M - 1] at  $m/z$  209.1.

**Compound 108: Phosphoric acid monooctadecyl ester**

200 mg (0.45 mmol) of 103 were employed and a procedure the same as that for 106 was used to afford 112 mg (94%) of a white solid 108.  $^1\text{H}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 0.88 (t,  $J = 6.6$  Hz, 3H,  $\text{CH}_3$ ), 1.27 (br s, 18 H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ ), 1.67 (quintet,  $J = 6.6$  Hz, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ ), 3.97 (dt,  $J = 6.6$  Hz, 6.6 Hz, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 14.21, 22.98, 25.84, 29.57, 29.67, 29.89, 29.92, 29.96, 30.69 (d,  $J_{\text{CP}} = 7.4$  Hz), 32.25, 67.22 (d,  $J_{\text{CP}} = 5.7$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 21.22; MS (negative mode): [M - 1] at  $m/z$  265.0.

**Compound 109: Phosphoric acid monoocadecyl ester**

200 mg (0.38 mmol) of 104 were used and an analogous procedure to that of 106 was employed which yielded 104 mg (79%) of a white solid 109.  $^1\text{H}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 0.89 (t,  $J = 6.9$  Hz, 3H,  $\text{CH}_3$ ), 1.27 (br s, 30H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ ), 1.68 (quintet,  $J = 6.9$  Hz, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ ), 3.98 (dt,  $J = 6.6$  Hz, 6.9 Hz, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_5\text{CH}_3$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 14.26, 23.14, 26.01, 29.74, 29.84, 30.06, 30.16, 30.87 (d,  $J_{\text{CP}} = 7.2$  Hz), 32.42, 67.32 (d,  $J_{\text{CP}} = 5.8$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 21.69; MS (negative mode): [M - 1] at  $m/z$  349.1.

**Compound 110: Phosphoric acid monodecosyl ester**

200 mg (0.34 mmol) of 105 were employed and the same procedure as that for 106 was yielding 98 mg (71%) of a white solid 110.  $^1\text{H}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 0.88 (t,  $J = 6.9$  Hz, 3H), 1.26 (br s, 38H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_9\text{CH}_3$ ), 1.66 (quintet,  $J = 6.9$  Hz, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_9\text{CH}_3$ ), 3.97 (d,  $J = 6.6$  Hz, 6.6 Hz, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_9\text{CH}_3$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 14.22, 23.01, 25.87, 29.61, 29.71, 29.91, 29.97, 30.04, 30.73 (d,  $J_{\text{CP}} = 7.4$  Hz), 32.29, 67.27 (d,  $J_{\text{CP}} = 5.6$  Hz);  $^{31}\text{P}$  NMR ( $\text{CDCl}_3/\text{MeOH-}d_4$ ) 20.66; MS (negative mode): [M - 1] at  $m/z$  405.1.

**Example 18 - Straight-Chain Phosphate Compounds 106-110**

*Xenopus* oocytes which endogenously express PSP24 PLGFR were used to screen compounds 106-110 for their LPA inhibitory activity. Oocytes were obtained from xylazine-anesthetized adult *Xenopus laevis* frogs (Carolina Scientific, Burlington, NC) under aseptic conditions and prepared for experiment. Stage V-VI oocytes were denuded of the the follicular cell layer with type A collagenase treatment (Boehringer, IN) at 1.4 mg/ml in a  $\text{Ca}^{2+}$ -free ovarian Ringers-2 solution ((OR-2) 82.5 mM NaCl, 2 mM KCl, 1 mM  $\text{MgCl}_2$ , 5 mM HEPES, pH 7.5, with NaOH). Oocytes were kept in Barth's solution in an incubator between 17-20 °C and were used for 2-7 days after isolation.

Electrophysiological recordings were carried out using a standard two-electrode voltage-clamp amplifier holding the membrane potential at -60 mV (GeneClamp 500, Axon Instruments, CA). Test compounds were dissolved in MeOH, complexed with fatty acid free BSA, and diluted with frog  $\text{Na}^+$ -Ringers solution (120 nM NaCl, 2 mM KCl, 1.8 mM  $\text{CaCl}_2$ , 5 mM HEPES, pH 7.0), which were applied through superfusion to the oocyte at a flow rate of 5 ml/min. Membrane currents were recorded with a NI/Cl-310 digital oscilloscope (Nicolet, Madison, WI). Applications were made at intervals of 1.5 mins (minimum) to allow for the appropriate washout and recovery from desensitization.

Figure 36 shows the dose-dependent inhibition of LPA-induced chloride currents by compounds 106-110. Compound 108 was the best inhibitor, having an  $\text{IC}_{50}$  value of about 8.1 nM. Compounds with shorter or longer straight-chain alkyl groups showed decreasing efficacy in inhibiting LPA-induced chloride currents, although compound 107 displayed a similar efficacy with an  $\text{IC}_{50}$  value of about 10.2 nM. Figure 37 compares the  $\text{EC}_{50}$  values for positive control solution (LPA alone), 25 nM, and a solution containing LPA and 100 nM of compound 108, 343 nM. Thus, compound 108 effectively inhibits LPA signalling of PSP24 receptors in *Xenopus* oocytes.

Based on the above results, compound 108 was also examined for its effectiveness as an antagonist of Edg-2, -4, and -7 receptors in RH7777 cells which heterologously express the individual receptors.

Figure 38 shows the effect of compound 108 on the  $\text{Ca}^{2+}$  responses in Edg-2, Edg-4, and Edg-7 expressing cells when exposed to a combination of LPA 1:1 and compound 108. For these experiments, the concentration of LPA was chosen to be near the  $\text{EC}_{50}$ . Compound 108 significantly inhibited the  $\text{Ca}^{2+}$  responses to about 65% and 56% of control, respectively, in Edg-2 and Edg-7 expressing cell lines. In

contrast, compound 108 significantly increased the  $\text{Ca}^{2+}$  responses to about 148% of control in Edg-4 expressing cell lines.

Therefore, the straight-chain phosphates would be expected to selectively inhibit Edg-2 and Edg-7 activity *in vivo* and selectively enhance Edg-4 activity *in vivo*.

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Each of the references listed below is hereby incorporated by reference in its entirety into the specification of this application.

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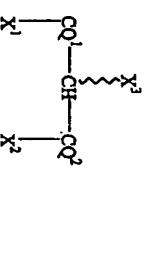
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30 Although preferred embodiments have been depicted and described in detail herein, it will be apparent to those skilled in the relevant art that various modifications, additions, substitutions, and the like can be made without departing from the spirit of the invention and these are therefore considered to be within the scope of the invention as defined in the claims which follow.

## What Is Claimed Is:

1. A compound according to formula (I)



wherein,

at least one of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  is  $(\text{HO})_2\text{PO---Z}^1$ — or

$(\text{HO})_2\text{PO---Z}^2\text{---P}(\text{OH})\text{O---Z}^1$ —,  $\text{X}^1$  and  $\text{X}^2$  are linked together as  $—\text{O---PO}(\text{OH})\text{---NH---}$ ;

at least one of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  is  $\text{R}^1\text{---Y}^1\text{---A---}$  with each

10 being the same or different when two of  $\text{X}^1$ ,  $\text{X}^2$  and  $\text{X}^3$  are  $\text{R}^1\text{---Y}^1\text{---A---}$ , or  $\text{X}^2$  and  $\text{X}^3$  are linked together as  $—\text{N}(\text{H})\text{---C}(\text{O})\text{---NR}^1$ —;

optionally, one of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  is  $\text{H}^1$ ;

$\text{A}$  is either a direct link,  $(\text{CH}_2)_k$  with  $k$  being an integer from 0

15 to 30, or  $\text{O}^1$ ;

$\text{Y}^1$  is  $—(\text{CH}_2)_l$ — with  $l$  being an integer from 1 to 30,  $—\text{O}^1$ ,

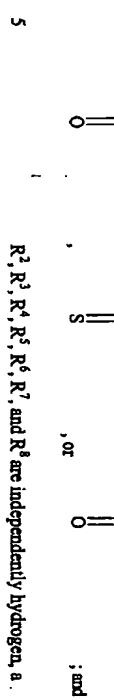
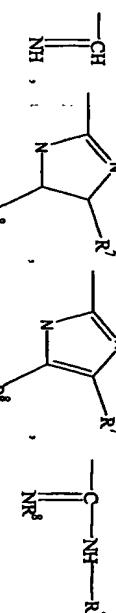
$—\text{S}^1$ ,  $\text{O}^1$  or  $\text{NR}^2$ —;

20  $\text{Z}^1$  is  $—(\text{CH}_2)_m$ — or  $—\text{O}(\text{CH}_2)_m$ — with  $m$  being an integer from 1 to 50,  $—\text{C}(\text{R}^3)\text{H}^1$ —,  $—\text{NH}^1$ —,  $—\text{O}^1$ —, or  $—\text{S}^1$ —;

$\text{Z}^2$  is  $—(\text{CH}_2)_n$ — or  $—\text{O}(\text{CH}_2)_n$ — with  $n$  being an integer from 1 to 50 or  $—\text{O}^1$ —;

25  $\text{Q}^1$  and  $\text{Q}^2$  are independently  $\text{H}_2$ ,  $=\text{NR}^4$ ,  $=\text{O}$ , a combination of  $\text{H}$  and  $—\text{NR}^5\text{R}^6$ ,

$\text{R}^1$ , for each of  $\text{X}^1$ ,  $\text{X}^2$ , or  $\text{X}^3$ , is independently hydrogen, a straight or branched-chain C1 to C30 alkyl, a straight or branched-chain C2 to C30 alkenyl, an aromatic or heteroaromatic ring with or without mono-, di-, or tri-



10 wherein the compound of formula (I) is not lysophosphatidic acid, phosphatidic acid, cyclic phosphatidic acid, alketyl glycerolphosphate, diacylglycerol pyrophosphate, or N-palmitoyl-L-serine.

15 2. The compound according to claim 1, wherein

$\text{Q}^1$  and  $\text{Q}^2$  are both  $\text{H}_2$ ;

one of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  is  $(\text{HO})_2\text{PO---Z}^1\text{---P}(\text{OH})\text{O---Z}^2$ —, with

20  $\text{Z}^1$  and  $\text{Z}^2$  being  $\text{O}$ ; and

two of  $\text{X}^1$ ,  $\text{X}^2$ , and  $\text{X}^3$  are  $\text{R}^1\text{---Y}^1\text{---A---}$ , with  $\text{A}$  being a direct link and  $\text{Y}^1$  being  $\text{O}$  for each.

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3. The compound according to claim 1, wherein  
 $Q^1$  is  $H_2$ ;  
 $Q^2$  is  $=O$ ;  
 $X^1$  is  $(HO)_2PO-Z^1-$ , with  $Z^1$  being O; and  
 $X^2$  and  $X^3$  are  $R^1-Y^1-A-$ , with A being a direct link and  $Y^1$   
5 being  $-NH-$  for each.

4. The compound according to claim 3, wherein  $X^3$  is  $-NH_2$  and  
 $X^1$  is  $-NHR^1$  with  $R^1$  being a C14 to C18 alkyl.

5. The compound according to claim 4, wherein  $R^1$  is a C14 alkyl.

6. The compound according to claim 4, wherein  $R^1$  is a C18 alkyl.

7. The compound according to claim 3, wherein  
 $X^1$  is  $-NHR^1$  with  $R^1$  being an acetyl group and  
 $X^2$  is  $-NHR^1$  with  $R^1$  being a C14 alkyl.

8. The compound according to claim 1, wherein

$Q^1$  is  $=NR^4$ ;

$Q^2$  is  $H_2$ ;

$X^1$  and  $X^2$  are linked together as  $-O-P(OH)-O-$ ; and  
 $X^3$  is  $R^1-Y^1-A-$ , with A being a direct link and  $Y^1$  being  
 $-NH-$ .

9. The compound according to claim 1, wherein  
 $Q^1$  and  $Q^2$  are both  $H_2$ ;  
two of  $X^1$ ,  $X^2$ , and  $X^3$  are  $(HO)_2PO-Z^1-$ , with  $Z^1$  being O;  
and  
30 one of  $X^1$ ,  $X^2$ , and  $X^3$  is  $R^1-Y^1-A-$ , with A being a direct  
link and  $Y^1$  being  $-O-$ .

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10. The compound according to claim 9, wherein  $R^1$  is an acyl including a C21 alkyl.

11. The compound according to claim 9, wherein  $R^1$  is a C18 alkyl.

12. A pharmaceutical composition comprising:  
a pharmaceutically-acceptable carrier and  
a compound according to claim 1.

13. A method of inhibiting LPA activity on an LPA receptor comprising:  
providing a compound according to claim 1 which has activity as an LPA receptor antagonist and  
10 contacting an LPA receptor with the compound under  
conditions effective to inhibit LPA-induced activity of the LPA receptor.

14. The method according to claim 13, wherein the LPA receptor is present on a cell and said contacting is carried out *in vitro*.

15. The method according to claim 13, wherein the LPA receptor is present on a cell and said contacting is carried out *in vivo*.

16. The method according to claim 13, wherein the LPA receptor is selected from the group consisting of EDG-2, EDG-4, EDG-7, and PAF-24.

17. A method of modulating LPA receptor activity comprising:  
providing a compound according to claim 1 which has activity as either an LPA receptor agonist or an LPA receptor antagonist and  
20 contacting an LPA receptor with the compound under  
conditions effective to modulate the activity of the LPA receptor.

18. The method according to claim 17, wherein the LPA receptor is present on a cell and said contacting is carried out *in vitro*.

19. The method according to claim 17, wherein the LPA receptor is present on a cell and said contacting is carried out *in vivo*.

5 20. The method according to claim 17, wherein the LPA receptor is selected from the group consisting of EDG-2, EDG-4, EDG-7, and PSP-24.

10 21. The method according to claim 17, wherein the compound has activity as an LPA receptor agonist and said contacting is carried out under conditions effective to induce LPA receptor activity.

15 22. The method according to claim 17, wherein the compound has activity as an LPA receptor antagonist and said contacting is carried out under conditions effective to reduce LPA receptor activity.

20 23. A method of treating cancer comprising: providing a compound according to claim 1 and administering an effective amount of the compound to a patient in a manner effective to treat cancer.

24. The method according to claim 23, wherein the cancer is prostate cancer or ovarian cancer.

25 25. The method according to claim 23, wherein the compound is an LPA receptor antagonist and said administering comprises: delivering the compound to cancer cells, where the compound binds to LPA receptors to inhibit proliferation or metastasis of the cancer cells.

26. The method according to claim 23, wherein upon delivering the compound to cancer cells, the cancer cells are destroyed.

30

5 27. A method of enhancing cell proliferation comprising: providing a compound according to claim 1 which has activity as an agonist of an LPA receptor and contacting the LPA receptor on a cell with the compound in a manner effective to enhance LPA receptor-induced proliferation of the cell.

10 28. The method according to claim 27, wherein the LPA receptor is selected from the group consisting of EDG-2, EDG-4, EDG-7, and PSP-24.

15 29. The method according to claim 27, wherein the cell is *in vitro*.

20 30. The method according to claim 27, wherein the cell is *in vivo*.

25 31. A method of treating a wound comprising: providing a compound according to claim 1 which has activity as an agonist of an LPA receptor and delivering an effective amount of the compound to a wound site, where the compound binds to LPA receptors on cells that promote healing of the wound, thereby stimulating LPA receptor agonist-induced cell proliferation to promote wound healing.

32. The method according to claim 31, wherein said delivering comprises: introducing to the wound site a composition comprising the compound and a pharmaceutically acceptable carrier.

33. The method according to claim 32, wherein the wound site is external and said introducing comprises: topically applying the composition to the wound site.

34. A method of making a compound according to claim 1 comprising:

reacting  $(Y^2O)_2PO-Z^{11}-Z^{13}$  or  $(Y^2O)_2PO-Z^{12}$  —

$P(OH)O-Z^{11}-Z^{13}$ , where

$Z^{11}$  is  $-(CH_2)_m-$  or  $-O(CH_2)_m-$  with  $m$  being an integer

from 1 to 50,  $-C(R^2)H-$ , or  $-O-$ ;

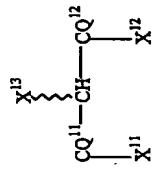
$Z^{12}$  is  $-(CH_2)_n-$  or  $-O(CH_2)_n-$  with  $n$  being an integer

from 1 to 50 or  $-O-$ ;

$Z^{13}$  is H or a first leaving group or  $-Z^{11}-Z^{13}$  together form the first leaving group; and

$Y^2$  is H or a protecting group,

with an intermediate compound according to formula (VII)



where,

at least one of  $X^{11}$ ,  $X^{12}$ , and  $X^{13}$  is  $R^{11}-Y^{11}-A-$  with each being the same or different when two of  $X^{11}$ ,  $X^{12}$ , and  $X^{13}$  are  $R^{11}-Y^{11}-A-$ , or  $X^{12}$  and  $X^{13}$  are linked together as  $-N(TD)-C(O)-N(R^{11})-$ ;

at least one of  $X^{11}$ ,  $X^{12}$ , and  $X^{13}$  is  $OH$ ,  $NH_2$ ,  $SH$ , or a second leaving group;

optionally, one of  $X^{11}$ ,  $X^{12}$ , and  $X^{13}$  is  $H$ ;

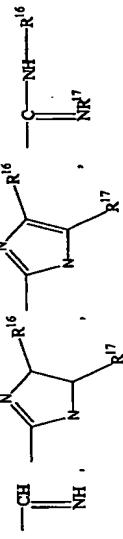
$A$  is either a direct link,  $(CH_2)_k$  with  $k$  being an integer from 0 to 30, or  $O$ ;

$Y^{11}$  is  $-(CH_2)_l-$  with  $l$  being an integer from 1 to 30,  $-O-$ ,

$\begin{array}{c} O \\ || \\ -C-S-, \text{ or } -NR^{12}- \\ | \\ Q^1 \end{array}$ , and  $Q^2$  are independently  $H$ ,  $s_3=NR^{13}$ ,  $=O$ , a combination of  $H$  and  $-NR^{14}R^{15}$ ;

30

$R^{11}$ , for each of  $X^{11}$ ,  $X^{12}$ , or  $X^{13}$ , is independently hydrogen, a straight or branched-chain C1 to C30 alkyl, a straight or branched-chain C2 to C30 alkanyl, an aromatic or heteroaromatic ring with or without mono-, di-, or tri-substitutions of the ring, an acyl including a C1 to C30 alkyl or an aromatic or heteroaromatic ring, an arylalkyl including straight or branched-chain C1 to C30 alkyl, an aryloxalkyl including straight or branched-chain C1 to C30 alkyl,



$R^{12}$ ,  $R^{13}$ ,  $R^{14}$ ,  $R^{15}$ ,  $R^{16}$ , and  $R^{17}$  are independently hydrogen, a straight or branched-chain C1 to C30 alkyl, a straight or branched-chain C2 to C30 alkanyl, an aromatic or heteroaromatic ring with or without mono-, di-, or tri-substitutions of the ring, an acyl including a C1 to C30 alkyl or an aromatic or heteroaromatic ring, an arylalkyl including straight or branched-chain C1 to C30 alkyl, or an aryloxalkyl including straight or branched-chain C1 to C30 alkyl;

including straight or branched-chain C1 to C30 alkyl;

followed by a de-protection step, if necessary, with both said reacting and the deprotection step being performed under conditions effective to afford a compound according to formula (I) where one or two of  $X^1$ ,  $X^2$ , and  $X^3$  is  $(HO)_nPO-Z^1-$  or  $(HO)_nPO-Z^2-P(OH)O-Z^1-$ .

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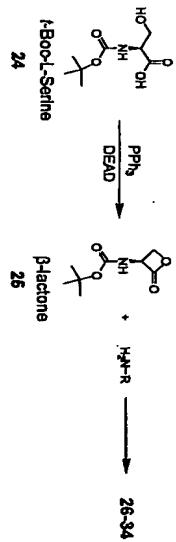
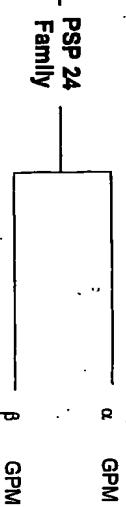
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Edg 7 GPM

Edg 2 GPM

Edg 4 GPM

Edg 6 SPM

Edg 8 SPM

Edg 3 SPM

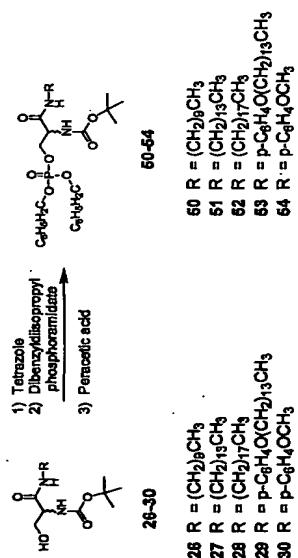
Edg 1 SPM

26 R =  $(\text{CH}_2)_9\text{CH}_3$   
 27 R =  $(\text{CH}_2)_{13}\text{CH}_3$   
 28 R =  $(\text{CH}_2)_{17}\text{CH}_3$   
 29 R =  $p\text{-C}_6\text{H}_4\text{O}(\text{CH}_2)_3\text{CH}_3$   
 30 R =  $p\text{-C}_6\text{H}_4\text{OCH}_3$   
 31 R =  $\pi\text{-C}_6\text{H}_3\text{O}(\text{CH}_2)_3\text{CH}_3$   
 32 R =  $m\text{-C}_6\text{H}_3\text{OCH}_3$   
 33 R =  $\sigma\text{-C}_6\text{H}_4\text{O}(\text{CH}_2)_3\text{CH}_3$   
 34 R =  $\sigma\text{-C}_6\text{H}_4\text{OCH}_3$

Figure 1

Figure 2

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Pd/C, H<sub>2</sub> →  
 55-59

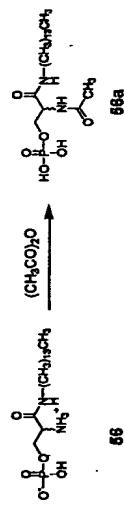


Figure 3

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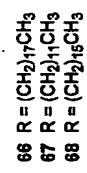
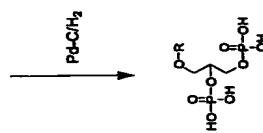
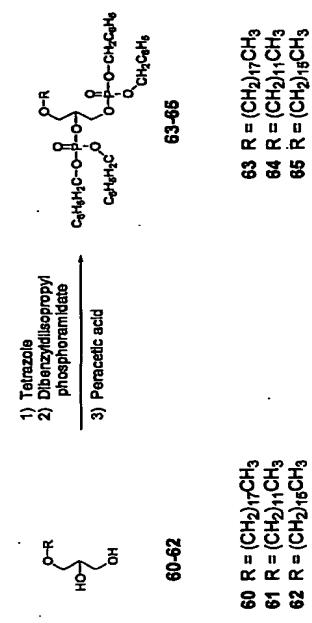


Figure 4

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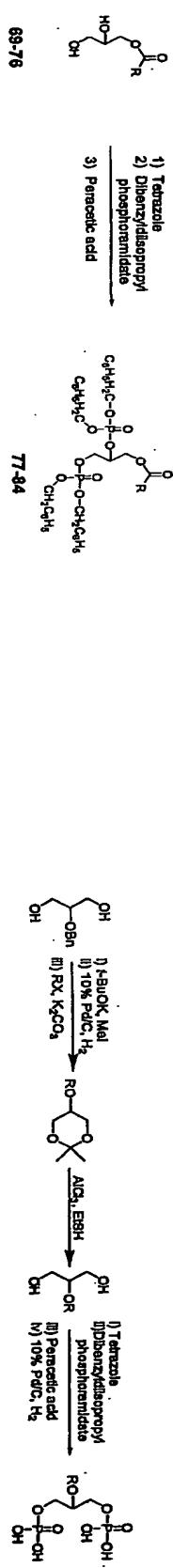


Figure 5B

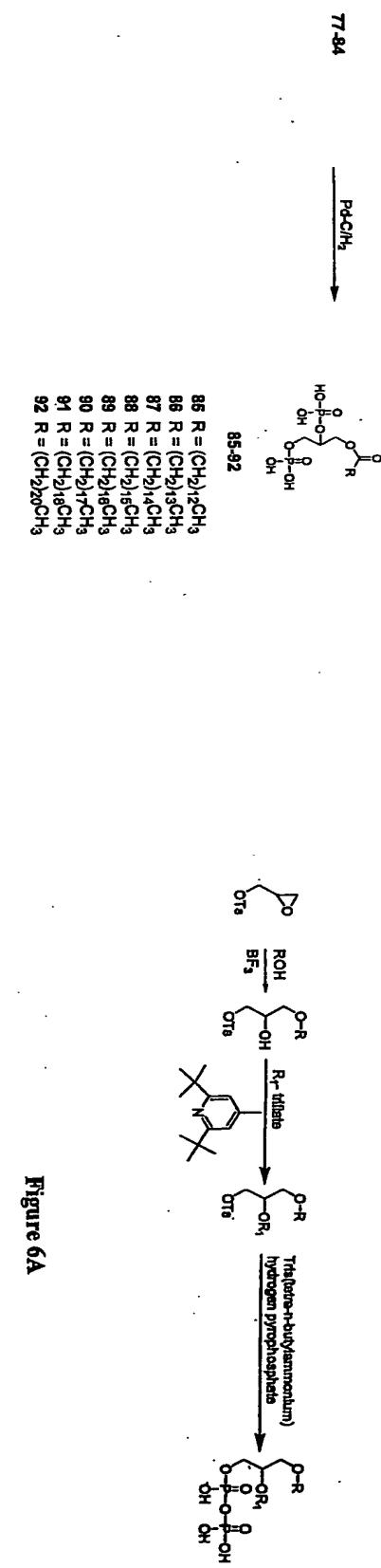


Figure 5A

Figure 6A

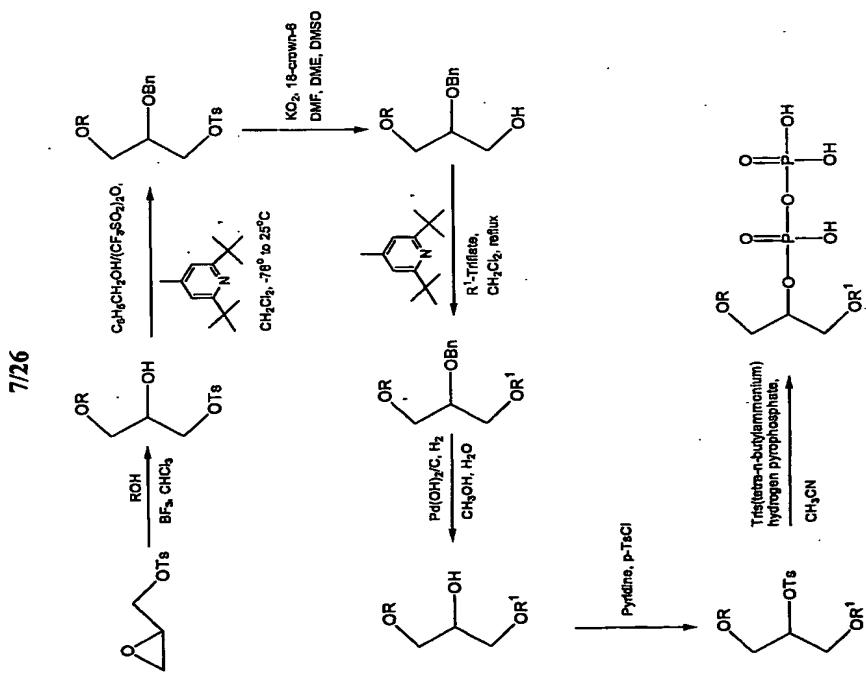


Figure 6B

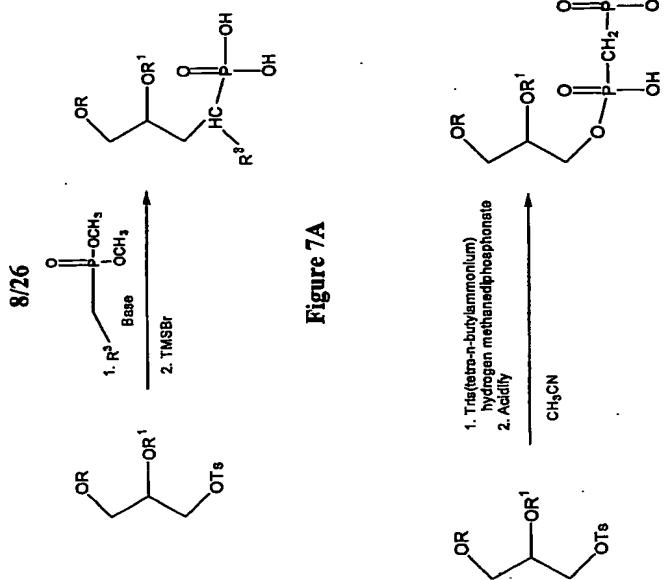


Figure 7A

Figure 7B

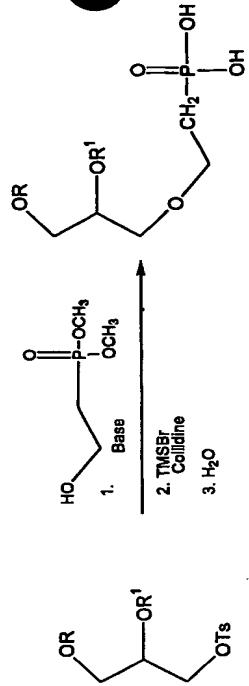


Figure 7C

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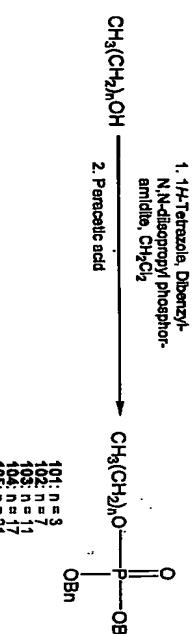


Figure 10

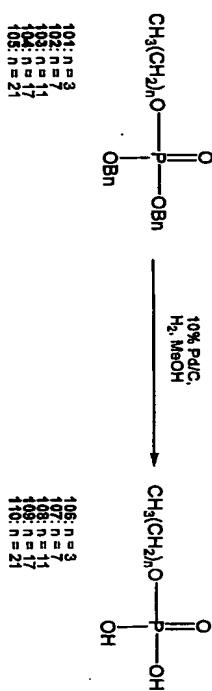


Figure 8

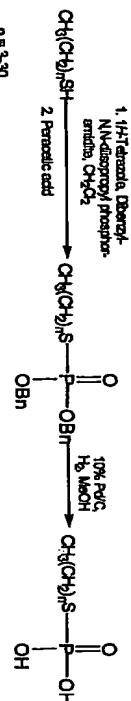


Figure 9

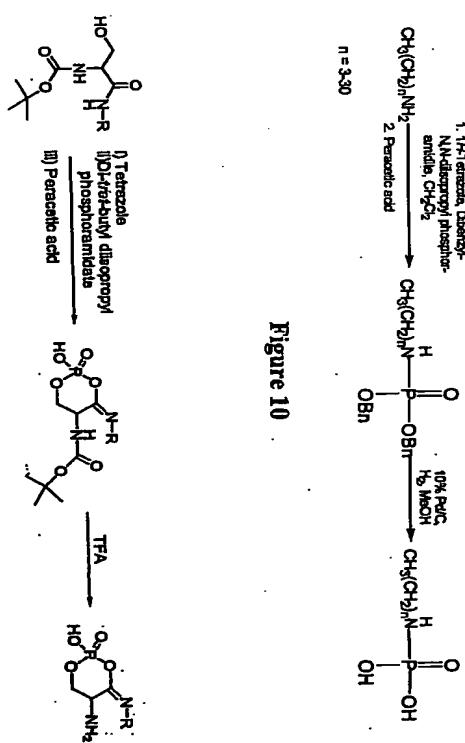


Figure 10

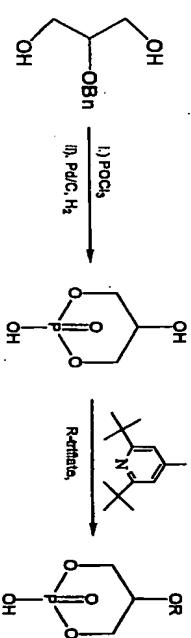


Figure 11

Figure 12

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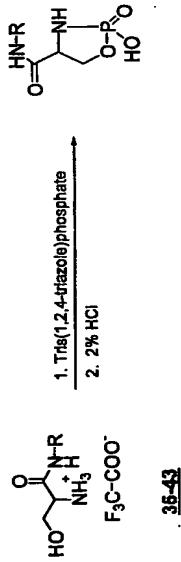


Figure 13

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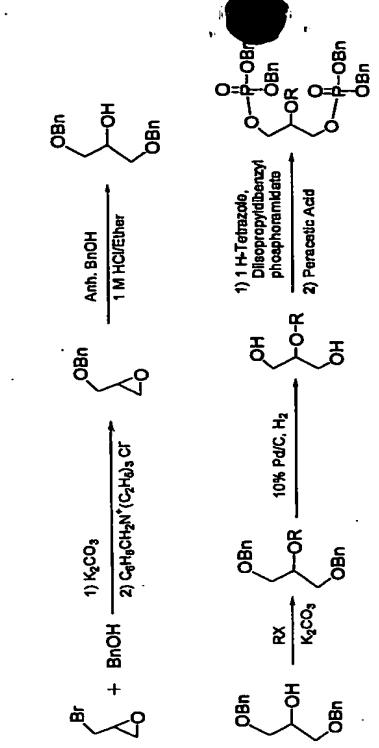


Figure 14

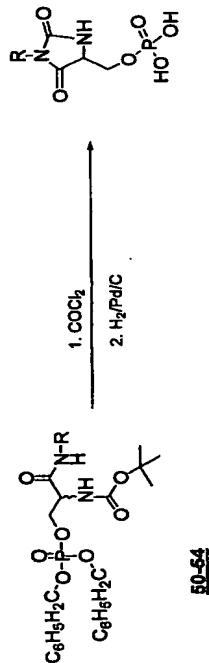


Figure 14

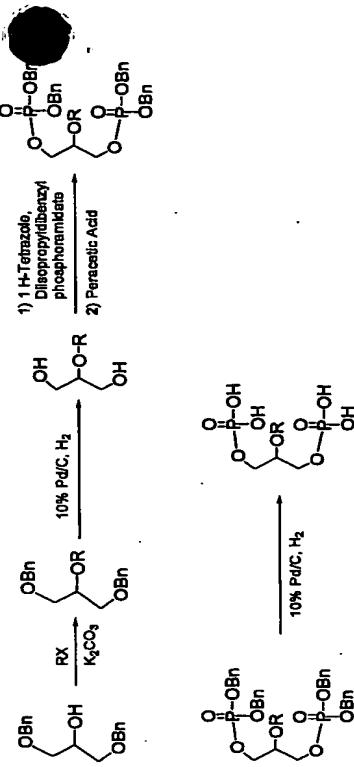


Figure 15

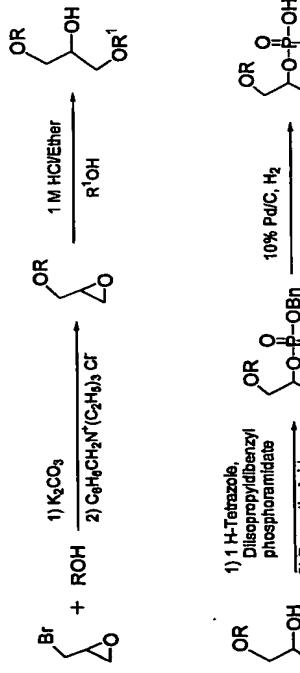


Figure 15

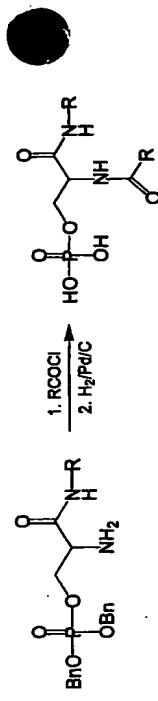


Figure 16

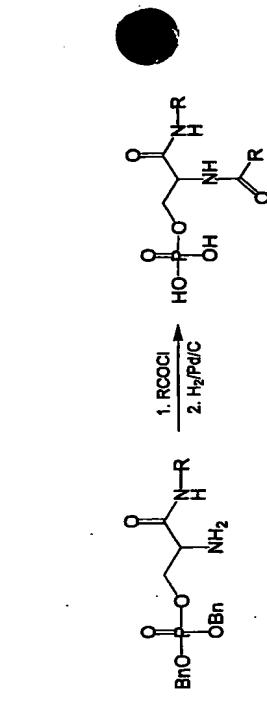


Figure 17

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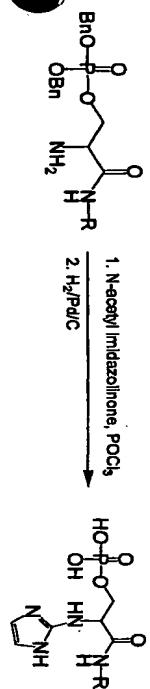


Figure 18

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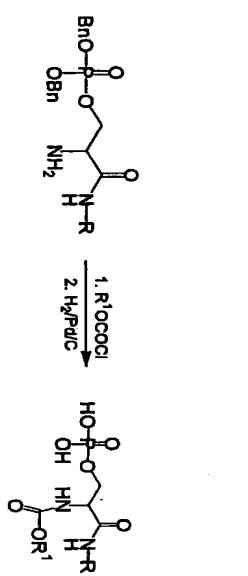


Figure 19

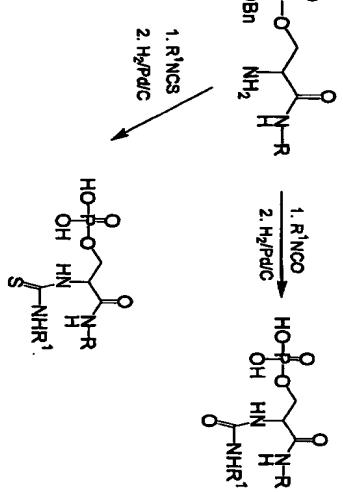


Figure 20

Figure 21

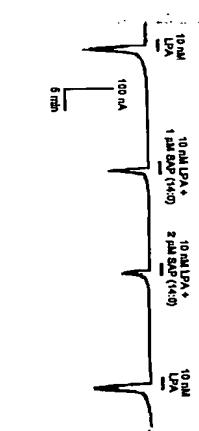
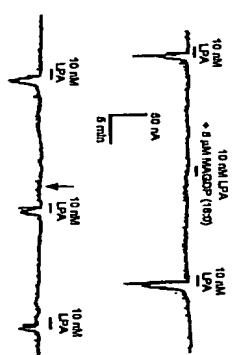


Figure 22



Figures 23A-B

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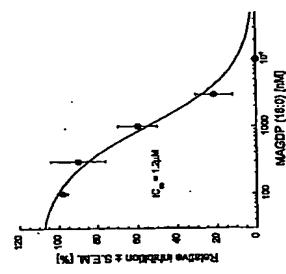


Figure 24

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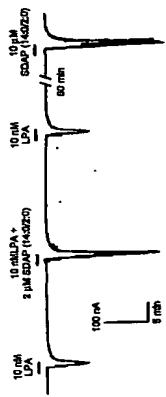


Figure 26

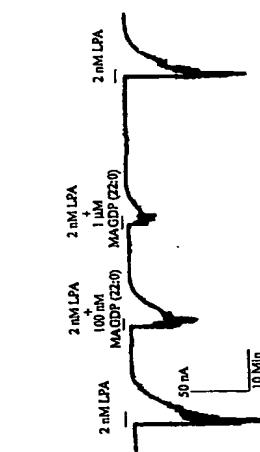


Figure 25

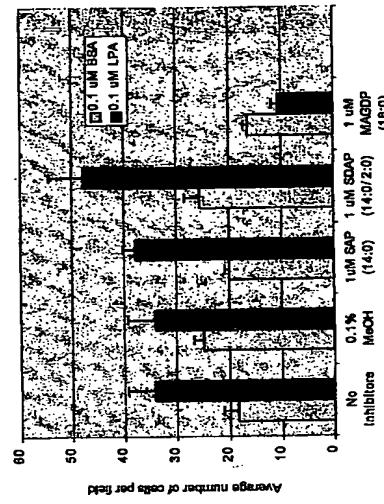
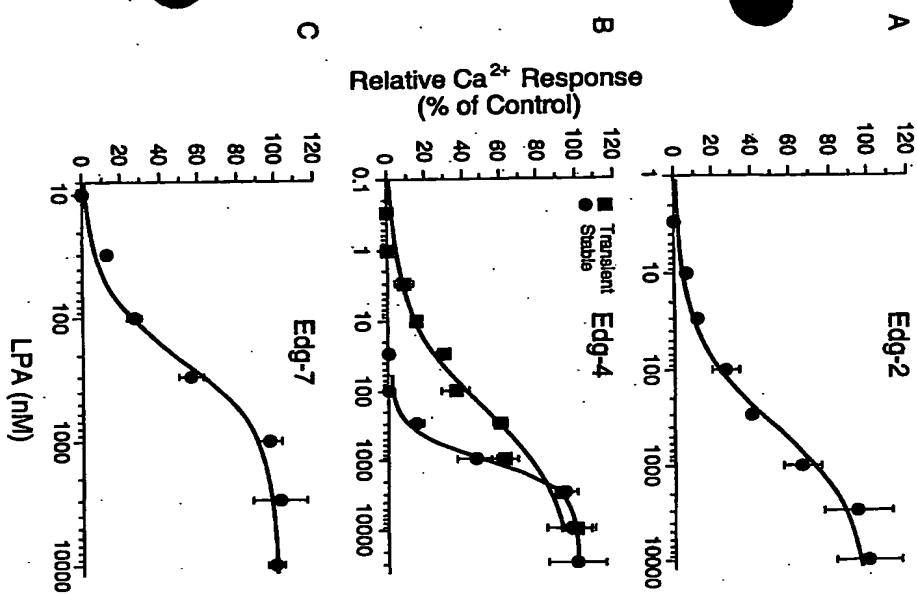


Figure 27

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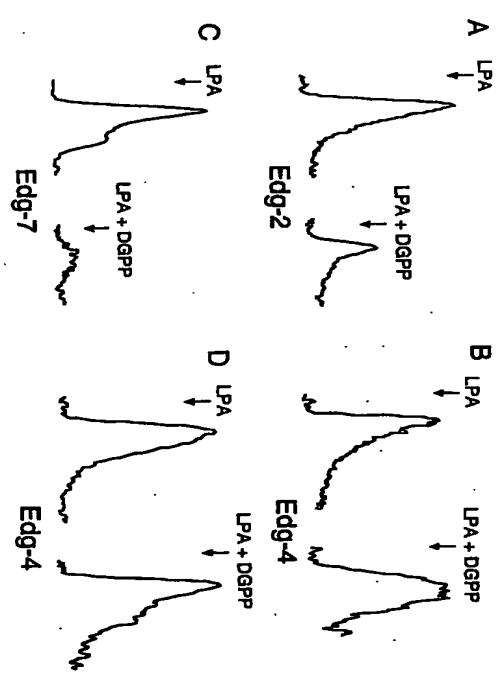
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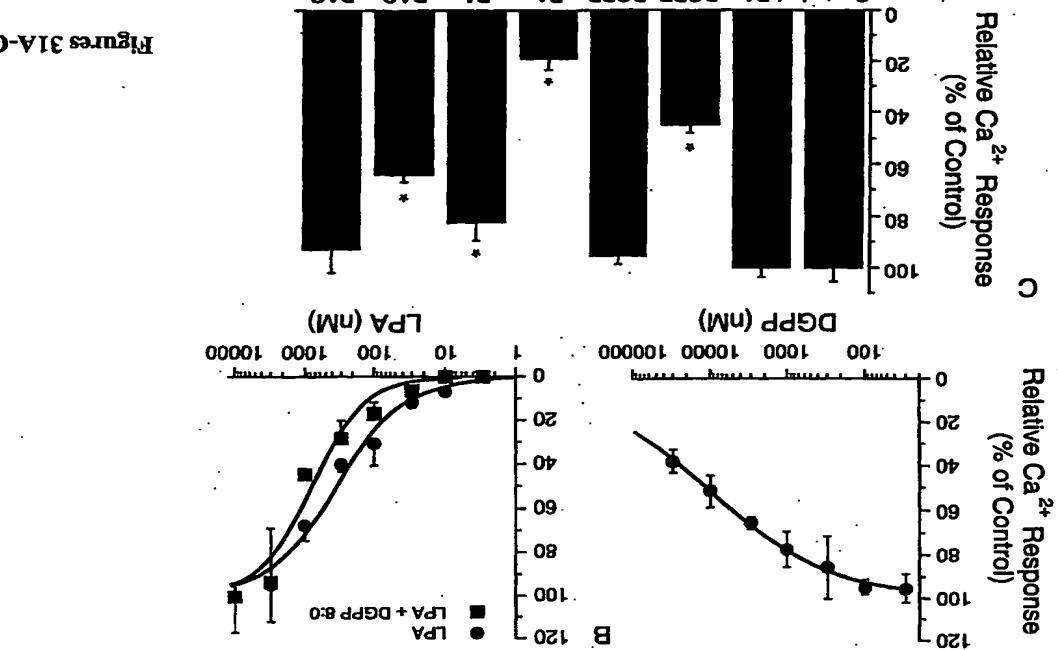
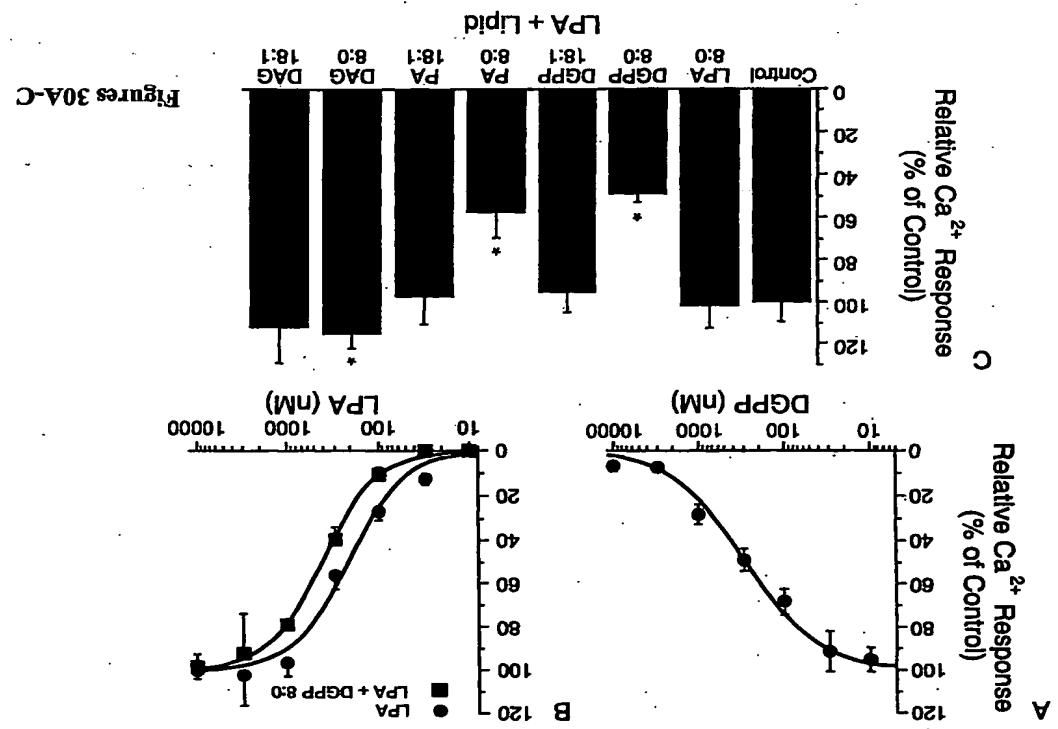
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Figures 28A-C

Figures 28A-D

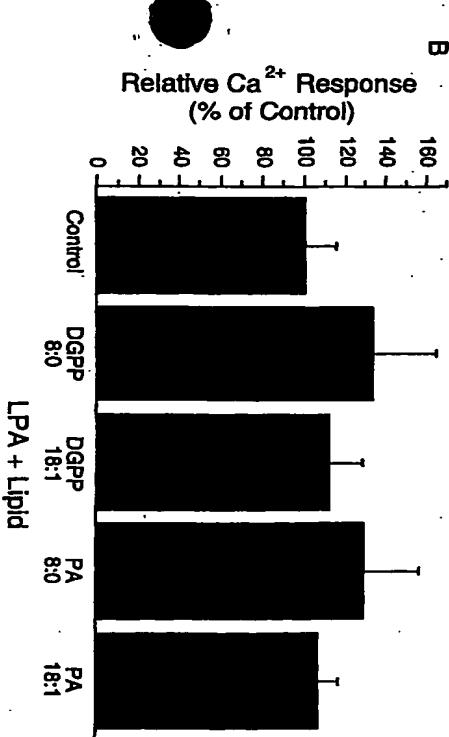
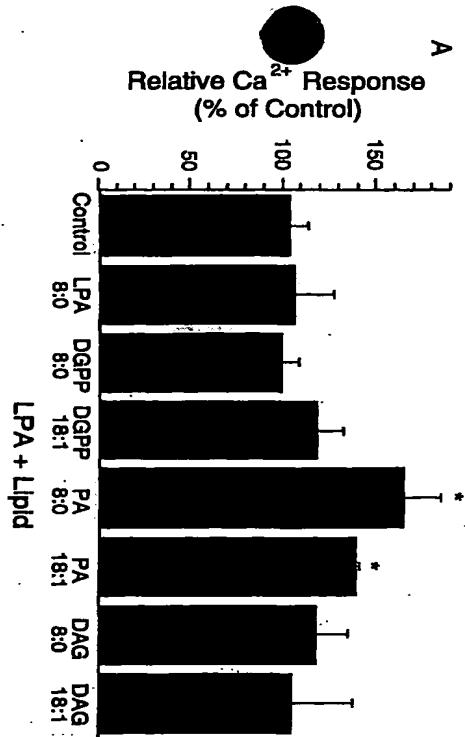




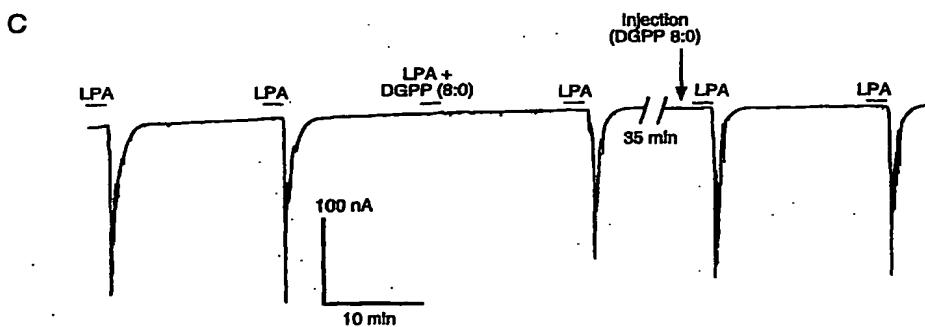
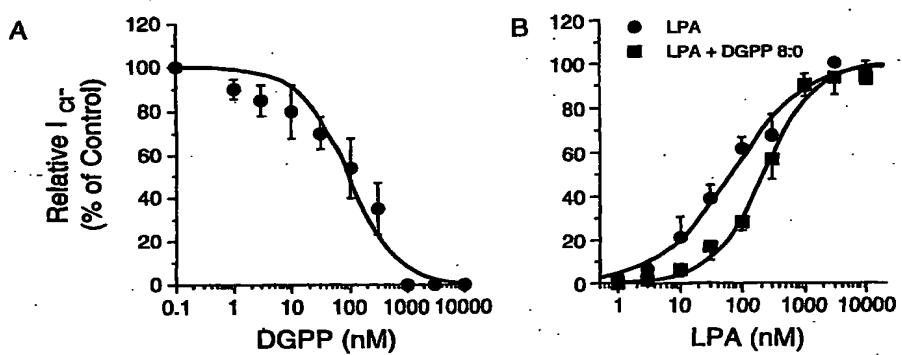
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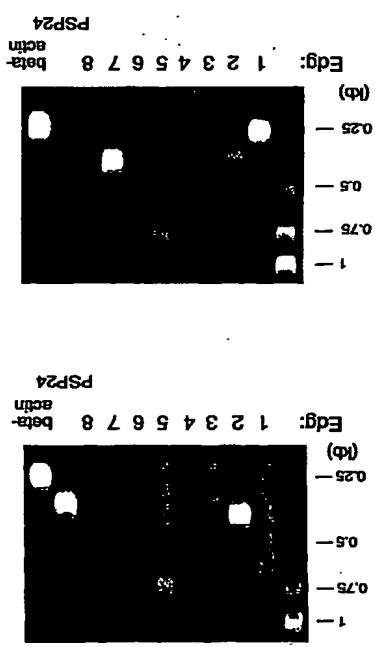
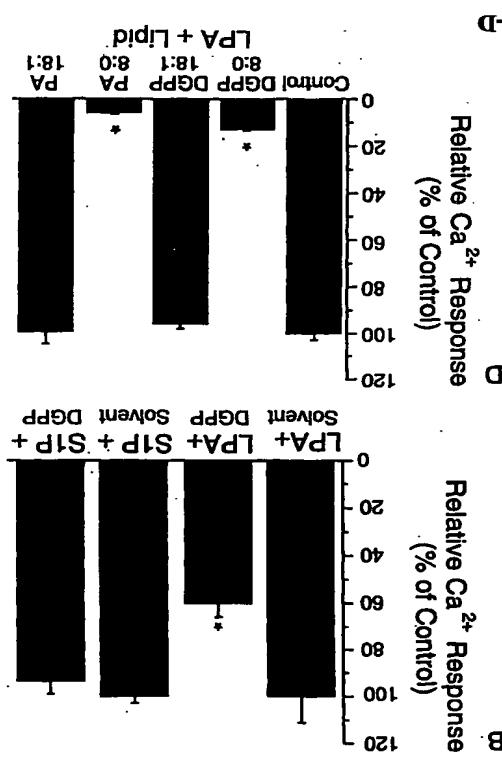
Figures 32A-C



Figures 32A-B

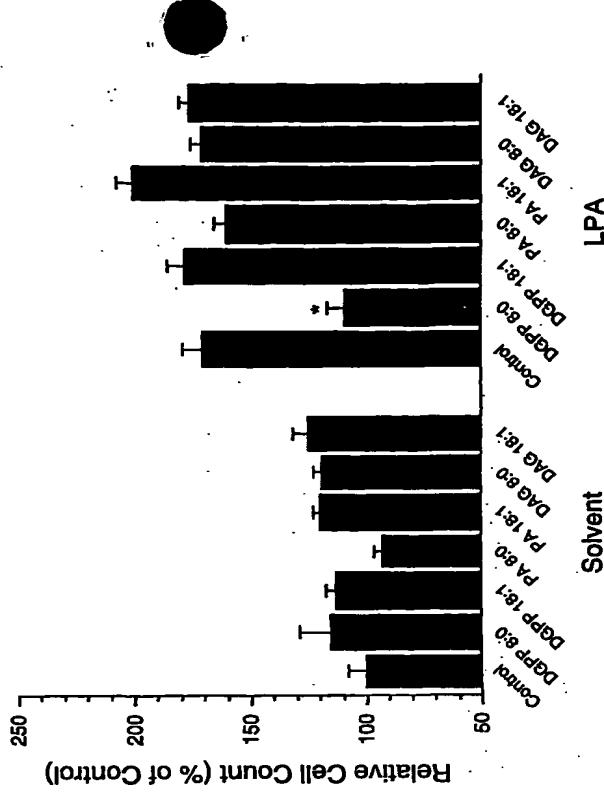


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Figure 35



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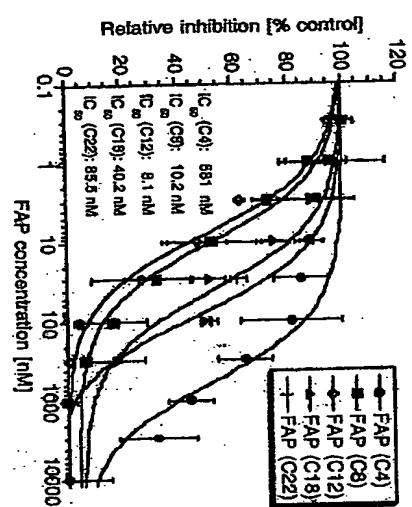


Figure 36

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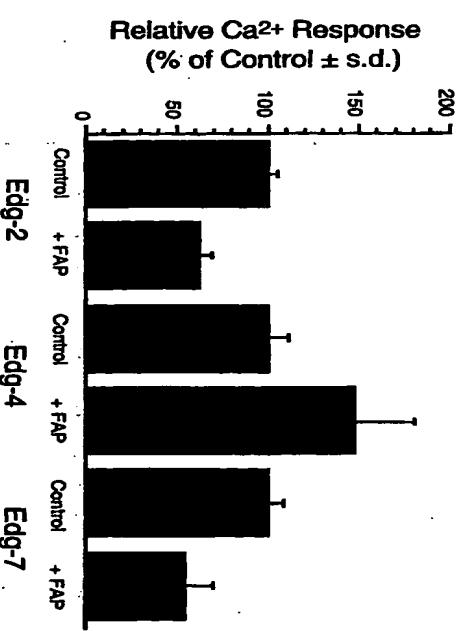


Figure 38

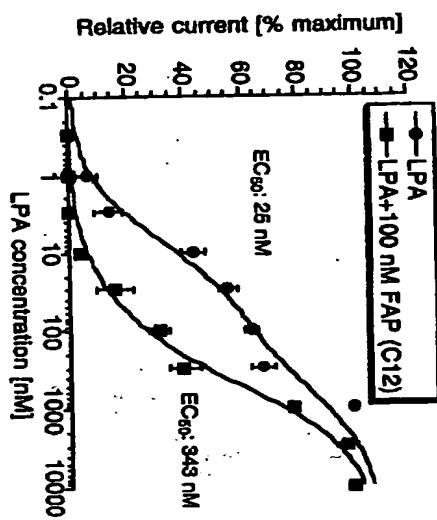


Figure 37



Arg His Ser Ser Gly Pro Arg Arg Asn Arg Asp Thr Met Met Ser Ile  
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 Ile Lys Thr Val Val Ile Val Ile Val Lys Gly Ala Phe Ile Ile Cys Trp Thr  
 260 265 270  
 Pro Gly Ile Val Val Ile Val Ile Val Lys Asp Val Cys Cys Pro Glu Cys Asp  
 275 280 285  
 Val Ile Ala Tyr Glu Lys Phe Phe Ile Ile Cys Trp Thr  
 290 295 300  
 Ala Met Asn Pro Ile Ile Tyr Ser Tyr Arg Asp Lys Glu Met Ser Ala  
 305 310 315 320  
 Thr Phe Arg Glu Ile Leu Cys Cys Gln Arg Ser Glu Asn Pro Thr Gly  
 325 330 335  
 Pro Thr Glu Ser Ser Asp Arg Ser Ala Ser Ser Leu Asn His Thr Ile  
 340 345 350  
 Leu Ala Gly Val His Ser Asn Asp His Ser Val Val  
 355 360  
  
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 <211> 1056  
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 20 25 30  
 Val Val Val Val Ala Leu Gly Ile Thr Val Ser Val Ile Val Leu Leu  
 35 40 45  
 Thr Asn Leu Leu Val Ile Ala Ile Ala Ser Asn Arg Arg Phe His  
 50 55 60  
 Gln Pro Ile Tyr Tyr Leu Leu Gly Asn Leu Asn Ala Ala Asp Leu Phe  
 65 70 75 80  
 Ala Gly Val Ala Tyr Leu Phe Ile Met Phe His Thr Gly Pro Arg Thr  
 85 90 95  
 Ala Arg Leu Ser Leu Glu Gly Trp Phe Leu Arg Glu Gly Leu Leu Asp  
 100 105 110  
  
 Thr Ser Leu Thr Ala Ser Val Ala Thr Ile Leu Ala Ile Ala Val Glu  
 115 120 125  
 Arg His Arg Ser Val Met Ala Val Gln Leu His Ser Arg Leu Pro Arg  
 130 135 140  
  
 Gly Arg Val Val Met Ile Ile Val Gly Val Trp Val Ala Ala Leu Gly  
 145 150 155 160  
 Ile Gly Ile Leu Pro Ala His Ser Trp His Cys Ile Cys Ala Leu Asp  
 165 170 175  
  
 Arg Cys Ser Arg Met Ala Pro Leu Leu Ser Arg Ser Tyr Leu Ala Val  
 180 185 190  
  
 Trp Ala Leu Ser Ser Leu Leu Val Phe Ile Leu Met Val Ala Val Tyr  
 195 200 205

Thr Arg Ile Phe Phe Tyr Val Arg Arg Arg Val Glu Arg Met Ala Glu	210	215	220	
His Val Ser Cys His Pro Arg Tyr Arg Glu Thr Thr Leu Ser Leu Val	225	230	235	240

Lys Thr Val Ile Ile Leu Gly Ala Phe Val Val Cys Trp Thr Pro  
245 240 250 255  
Gly Gln Val Val Leu Leu Asp Gly Leu Gly Cys Glu Ser Cys Asn  
260 265 270

Leu Val Asn Ala Ala Val Tyr Ser Cys Arg Asp Ala Glu Met Arg Arg  
290 295 275 280 285 300  
Val Leu Ala Val Glu Lys Tyr Phe Leu Leu Ala Glu Ala Asn Ser

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Glu Ser Val His Tyr Thr Ser Ser Ala Glu Gly Ala Ser Thr Arg  
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 tcgttcgtttt ctttttttt tttttttttt tttttttttt tttttttttt tttttttttt tttttttttt 180  
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<211> 353  
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<213> *Homo sapiens*

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Ser Asn Thr Asp Thr Val Asp Asp TPP Thr Gly Thr Lys Leu Val Ile	20	25	30	

	Val	Leu	Cys	Val	Gly	Thr	Phe	Cys	Leu	Phe	Leu	Phe	Ser	Asn
	35						40					45		
Ser	Leu	Val	Ile	Ala	Ala	Val	Ile	Lys	Asn	Arg	Lys	Phe	His	Phe
	50						55					60		

Leu Thr Ala Ser Ile Thr Asn Ile Leu Val Ile Ala Val Glu Arg His	115	120	125
Met Ser Ile Met Arg Met Arg Val His Ser Asn Leu Thr Lys Lys Arg	130	135	140
Val Thr Leu Leu Ile Leu Val Trp Ala Ile Ala Ile Phe Met Gly	145	150	155
Ala Val Pro Thr Leu Gly Trp Asn Cys Asn Ile Ser Ala Cys	165	170	175
Ser Ser Leu Ala Pro Ile Tyr Ser Arg Ser Tyr Leu Val Phe Ile Thr	180	185	190

Val Ser Asn Leu Met Ala Phe Leu Ile Met Val Val Val Tyr Leu Arg  
 195 200 205  
 Ile Tyr Val Tyr Val Lys Asp Lys Thr Asn Val Leu Ser Pro His Thr  
 210 215 220  
 Ser Gly Ser Ile Ser Arg Arg Arg Thr Pro Met Lys Leu Met Lys Thr  
 225 230 235 240  
 Val Met Thr Val Leu Gly Ala Phe Val Val Cys Tyr Thr Pro Gly Leu  
 245 250 255  
 Val Val Leu Leu Asp Gly Leu Asn Cys Arg Gly Cys Gly Val Gln  
 260 265 270  
 His Val Lys Arg Tyr Phe Leu Leu Ala Leu Leu Asn Ser Val Val  
 275 280 285  
 Asn Pro Ile Ile Tyr Ser Tyr Lys Asp Glu Asp Met Tyr Gly Thr Met  
 290 295 300  
 Lys Lys Met Ile Cys Cys Phe Ser Glu Asn Pro Glu Arg Arg Pro  
 305 310 315 320  
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 Pro Pro Phe Glu His Pro Asp Ile Ser Pro Leu Leu Arg Tyr Ser Phe  
 35 40 45  
 Glu Thr Met Ala Pro Thr Gly Ile Ser Ser Leu Thr Val Asn Ser Thr  
 50 55 60  
 Ala Val Pro Thr Thr Pro Ala Ala Phe Lys Ser Leu Asn Leu Pro Ile  
 65 70 75 80  
 Gln Ile Thr Ile Ser Ala Ile Met Ile Phe Ile Leu Phe Val Ser Phe  
 85 90 95  
 Leu Gly Asn Leu Val Val Cys Ile Met Val Tyr Gln Lys Ala Ala Met  
 100 105 110  
 Arg Ser Ala Ile Asn Ile Ile Leu Ala Ser Ile Ala Phe Ala Asp Met  
 115 120 125  
 Ile Leu Ala Val Ile Asn Met Pro Phe Ala Ile Val Thr Ile Ile Thr  
 130 135 140

Thr Arg Trp Ile Phe Gly Lys Phe Phe Cys Arg Val Ser Ala Met Phe  
 145 150 155 160 165 170 175 180 185 190 195 200 205 210 215 220  
 Phe Trp Leu Phe Val Ile Glu Gly Val Ala Ile Leu Ile Ile Ser  
 165 170 175  
 Ile Asp Arg Phe Leu Ile Ile Val Gln Arg Asp Lys Leu Asn Pro  
 180 185 190 195 200 205 210 215 220  
 Tyr Arg Ala Lys Val Ile Ala Val Ser Trp Ala Thr Ser Phe Cys  
 195 200 205  
 Val Ala Phe Pro Leu Ala Val Gly Asn Pro Asp Leu Gln Ile Pro Ser  
 210 215 220  
 Arg Ala Pro Gln Cys Val Phe Gly Tyr Thr Asn Pro Gly Tyr Gin  
 225 230 235 240  
 Ala Tyr Val Ile Leu Ile Ser Phe Phe Ile Pro Phe Leu  
 245 250 255  
 Val Ile Leu Tyr Ser Phe Met Gly Ile Leu Asn Thr Leu Arg His Asn  
 260 265 270  
 Ala Leu Arg Ile His Ser Tyr Pro Glu Gly Ile Cys Leu Ser Gln Ala  
 275 280 285  
 Ser Lys Leu Gly Leu Met Ser Leu Gln Arg Pro Phe Gln Met Ser Ile  
 290 295 300  
 Asp Met Gly Phe Phe Lys Thr Arg Ala Phe Thr Ile Leu Ile Leu Phe  
 305 310 315 320  
 Ala Val Phe Ile Val Cys Trp Ala Pro Phe Thr Thr Tyr Ser Leu Val  
 325 330 335  
 Ala Thr Phe Ser Lys His Phe Tyr Tyr Gln His Asn Phe Phe Glu Ile  
 340 345 350  
 Ser Thr Trp Leu Leu Trp Leu Cys Tyr Leu Lys Ser Ala Leu Asn Pro  
 355 360 365  
 Leu Ile Tyr Tyr Trp Arg Ile Lys Phe His Asp Ala Cys Leu Asp  
 370 375 380  
 Met Met Pro Lys Ser Phe Lys Phe Leu Pro Gln Leu Pro Gly His Thr  
 385 390 395 400

Lys Arg Arg Ile Arg Pro Ser Ala Val Tyr Val Cys Gly Glu His Arg  
 405 410 415  
 Thr Val Val

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 forward EDG-1

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<210> 11  
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22

&lt;210&gt; 13

&lt;211&gt; 22

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

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&lt;211&gt; 16

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

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<223> Description of Artificial Sequence: primer,  
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&lt;211&gt; 22

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&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 17

&lt;211&gt; 21

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

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&lt;213&gt; Artificial Sequence

&lt;210&gt;

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21

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 16

&lt;211&gt; 22

&lt;212&gt; DNA

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22

&lt;210&gt; 17

&lt;211&gt; 21

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&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 17

&lt;211&gt; 21

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 17

&lt;211&gt; 21

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 17

&lt;211&gt; 21

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 17

&lt;211&gt; 21

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 17

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&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 17

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22

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22

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&lt;211&gt; 21

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&lt;213&gt; Artificial Sequence

22

&lt;210&gt; 17

&lt;211&gt; 21

&lt;212&gt; DNA

&lt;213&gt; Artificial Sequence

11

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## INTERNATIONAL SEARCH REPORT

International Application No.

PCT/US01/00729

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/US01/00729

## C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevance to claim No.
X	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:87:53513, GIBBS, D. 'The synthesis of phosphoramidates from silylphosphites and azides', abst Tetrahedron Lett. (8), pages 679-82, 1977.	1, 9 ... 3-7
X	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:92:53762, BIJUME, A. et al 'The influence of charge on bilayer membranes. Calorimetric investigations of phosphatidic acid bilayers', abst Biophys. Acta 558(1) pages 13-21, 1979	1, 9
X	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:130:126394, WEI, S. et al 'Study on new amphoteric surfactants of phosphates I. Syntheses and properties', abst Jingxi Huagong 15(5), 1-5, 1998.	1 ... 3-7
X, P	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:134:174618, BADALASSI, F. 'A versatile periodate-coupled fluorogenic assay for hydrolytic enzymes' abst Angew. Chem. Int. Ed. 39(22) pages 4067-4070, 2000.	1, 9
X	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:59:5425, BUSHERNEV, A.S. et al 'Synthesis of rac-3-benzoyl-1-deoxyceramide-1-phosphonic acid' abst Bioorg. Khim 9 (4) pages 533-5, 1983	1
X	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:75:71703, CATES, L. 'Phosphorus-nitrogen compounds 12. Phosphamidate studies. 2. N-alkylphosphoramidic acids' abst J. Med. Chem. 14(7) pages 607-9, 1971.	1, 12
X	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:116:221452, GAS CO, M.R. et al 'Timolol in liposomes' abst Pharmazie 47 (2) pages 119-21, 1992.	1, 12
X	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:76:34548, AVAEVA, S. M. et al 'Hydrolysis of phosphoric ester serine derivatives containing free amino or carboxy groups' abst Vestn. Mosk. Univ. Khim. 12(5) pages 637-8, 1971.	1, 3
X	Database Caplus on STN, Chemical Abstracts (Columbus Ohio, USA), CA:116:236128, TONG, G. et al 'Synthesis of the simple peptide model Ac-Abu[PO3H2]-NHMe' abst Aust. J. Chem. 45(4) pages 777-84, 1992.	1, 3 ... 4-7

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages





## INTERNATIONAL SEARCH REPORT

PCT/US01/08729

INTERNATIONAL SEARCH REPORT

International application  
PCT/US01/08722

Group XXXVI, claim(s) 1 (in part), 8 (in part), 12 (in part), drawn to compounds of the formula (1) which contains a linking group of  $X_1$  and  $X_2$  into  $O-P(OH)_2-NH_2$  or  $X_2$  into  $OPO(OH)_2-NH_2$  and at least one of  $X_1$  or  $X_2$  is  $R_1-CH_2-$  where  $R_1$  is not a heterocyclic ring or an nitrogen containing group and  $Y_1$  is  $NH_2$ .  $A$  is a link or  $(CH_2)_n$ , or  $O$  and  $Z_1$  is  $NH$ ,  $Z_2$  is  $O$  or  $O(CH_2)n_1$ , and  $Q_1$  and  $Q_2$  are  $=NR^1-NR^2R^3$  or at least one  $Q_1$  or  $Q_2$  is  $=NR^1-NR^2R^3$  (i.e. nonheterocyclic amide ester  $-R(OH)_2P(OH)_2-NH_2-Q_1-Q_2$ ).

$\beta$  of  $X_1$  and  $X_2$  into  $\text{CH}_2\text{P}(\text{O})(\text{Y}_1)$  or  $\text{A}_1$  and  $\text{A}_2$  into  $\text{CH}_2\text{P}(\text{O})(\text{Y}_1\text{Y}_2)$  and  $\alpha$  of  $X_1$  is one of a heterocyclic ring or a nitrogen containing group and  $\text{Y}_1$  is  $(\text{CH}_3)_2$  or  $\text{O}$ , or carbonyl,  $\text{A}_1$  is a link or  $\text{O}$ , and  $\text{Z}_1$  is  $\text{S}$ ,  $\text{Z}_2$  is  $\text{O}$  or  $\text{OCH}_2\text{Z}_3$ .  $\text{Q}_1$  and  $\text{Q}_2$  are  $\text{H}_2$  or  $\text{O}$ , (i.e. monothiophosphate ester  $-\text{HO}(\text{O})\text{P}(\text{O})(\text{S}-\text{CQ}_1\text{---})$ ).

Group XXXI: claims 1 (in part), 12 (in part), drawn to compounds of formula (I) which contain  $X_2$  and  $X_3$  linked as  $-NH-C(=O)-N-$  and at least one of  $X_1$  or  $X_2$  or  $X_3$  is  $R_1$  or  $R_2$  where  $R_1$  is not a hydroxylic ring or a nitrogen-containing group and  $Y_1$  is  $CH_2$ ,  $U$  or  $C(O)$ .  $U$  is a link or  $C(CH_3)_2$ ,  $Z_1$  is  $O$  ( $CH_2$ ),  $Z_2$  is  $O$  or  $OCH_2$ ,  $Q_1$  and  $Q_2$  are  $=NR^1-NR^2$  or at least one  $Q_1$  or  $Q_2$  is  $=NR^1-NR^2$  (i.e., monophosphate ester- $(HO)_2PO(O-CQ_1)-$  or linked ester  $P(O-2Z-2P)$ ).

NR1— and at least one of X1 or X2 or X3 is R1—Y1—A (where R1 is not a heterocyclic ring or a 0 or nitrogen containing group) and Y1 is C(=O) or O or carbonyl. A is a link or —(CH<sub>2</sub>)<sub>n</sub>— or Z1 is NH, Z2 is O or NH<sub>2</sub> or —(CH<sub>2</sub>)<sub>n</sub>—O—NH<sub>2</sub> or at least one Q1 or Q2 is —NR<sub>1</sub>—NR<sub>2</sub>— (i.e., monophosphorus amide ester —(CH<sub>2</sub>)<sub>n</sub>—O—Q1—O—Q2—).

Group XXXIII, chlorine) (16 parts), (2) (in parts), drawn to compounds of the formula (1) which contain  $X_2$  and  $X_3$  linked  $\text{--NH}_2\text{CO}(\text{O})\text{NR}_1$ , and at least one of  $X_1$  or  $X_2$  or  $X_3$  is  $\text{R}_1\text{--Y}_1\text{--Y}_2$ , wherein  $\text{R}_1$  is not a heterocyclic ring or a hydrogen containing group and  $\text{Y}_1$  is  $(\text{CH}_2)_n$ , or O, or carbonyl,  $\text{Y}_2$  is a link of  $(\text{CH}_2)_m$ , or O and  $\text{Z}_1$  is S,  $\text{Z}_2$  is O or

Group XXXIV, claim 1 (in part), 11 (in part), drawn to compounds of the formula (1) which contain X and X3 linked as  $\text{NH-C(=O)-NR}_1$  and at least one of X1 or X2 or X3 is  $\text{R}_1-\text{Y}_1-\text{A}$ , wherein  $\text{R}_1$  is not a heterocyclic ring or is a nitrogen containing group and  $\text{Y}_1$  is  $(\text{CH}_2)_n$  or O, or carbonyl. A is a link  $(\text{CH}_2)_m$ , O and Z1 is  $(\text{CH}_2)_n$  or  $\text{C(=O)NR}_2$ , or  $\text{C(=O)OCH}_2\text{R}_1$ , Q1 and Q2 are  $=\text{NR}_1-\text{NR}_2$  or at least one Q1 or Q2 is  $=\text{NR}_1-\text{NR}_2$ , (i.e. piperazine and an ester  $-\text{R}(\text{O})\text{CH}_2\text{C(=O)Q}_1\text{Q}_2-$ ).

Group XXXV, classes (in part), (17 in part), shown to be compounds of formula (1) which contain X and Y3 linked as NH-C(=O)-NH<sub>2</sub>, and at least one of X<sub>1</sub> or X<sub>2</sub> is a J3-Y1-A wherein Y1 is an aromatic ring or on a nitrogen containing and group Y1 is R<sub>2</sub>, or A is CR<sub>2</sub>Y1-A wherein Y1 is an aromatic ring or on a nitrogen containing and group Y1 is CR<sub>2</sub>Y1-A, or R<sub>2</sub> and R<sub>1</sub> is an (hetero)aromatic ring or O, Z2 is O or O(=CH<sub>2</sub>), Q1 and Q2 are =NR<sup>1</sup>-NR<sup>2</sup>R<sup>3</sup> or at least one Q1 or Q2 is =NR<sup>1</sup>-NR<sup>2</sup>R<sup>3</sup>.

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Group XXXVII, claim 1 (iia part), 12 (ii part), drawn to components of the formula (1) which contains X and X3 linked as  $-\text{NR}_1\text{C}(\text{O})-\text{NR}_1$  and X3 and at least one of X1 or X2 or X3 is  $\text{R}_1-\text{Y}_1-\text{A}$ , wherein R1 is not a heterocyclic ring or a nitrogen containing group and Y1 is NR2. A is a link or  $(\text{CH}_2)_2$ , or O and Z1 is NH, Z2 is O or  $\text{OCH}_2\text{H}_2$ , Q1 and Q2 are  $-\text{NR}_1\text{R}$  or at least one Q1 or Q2 is  $=\text{NR}_1-\text{NR}_1\text{R}$ , (i.e. monodrophosphine amido ester,  $(\text{RO}_2\text{P}(\text{O})\text{NR}_1\text{CQ}_1$ ).

Group XXXIX-LXVII are the same as the above 1-XXXIX except that R1 is a heterocyclic or a nitrogen donor. These will be the same claims (in part) as those above.

Group LXVII claim 1 (in part) and 2 (in part) any compounds not found in Groups I-LXVII.

Group LXXX, claims 22-26, drawn to a method of treating cancer.

Group LXXXI, claims 31-33, drawn to a method of treating a wound.

Under PCT Rule 13.1, they fail the same or corresponding special technical feature for the following reason:  
 Group I has a series of various structures which only contains the common core -CH<sub>2</sub>. The -CH<sub>2</sub>- does not constitute a large structural common core since there are hundreds of thousands of groups which contains -CH<sub>2</sub>. It is also noted that the -CH<sub>2</sub>- is not a small structural portion which is applicant's contribution to the art. The claimed composition, acidic, amides, diesters, heterocyclic phosphorus and iron phosphorus containing groups which only have a CH<sub>2</sub> as the common core. Therefore there is no special technical feature using all the various compounds.

Also there is no special technical feature among the compounds being claimed and the method of preparing the compounds and the various methods of using the compounds since there are known compounds being claimed such as those compounds found in CA.16:23068, CA.117:16626, CA.83:11702, CA.117:39162 or 115:20251. These compounds have their own various utilities and their various methods of being prepared. Therefore no utility of invention since there is no special technical feature which units the groups by providing a compound, a method especially designed to prepare the compound and a method of using the compound that is specially linked to a special technical feature.